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(54) Pharmaceutically active amides

(57) Compounds of general formula I

$$R_{3} = \begin{bmatrix} R_{4} \\ \vdots \\ R_{5} \end{bmatrix}$$

$$R_{1}$$

$$R_{2}$$

(wherein, in outline, R1 and R2 represent alkyl or cycloalkyl groups or together with the nitrogen atom to which they are attached, represent a cyclic imino group, R<sub>3</sub> represents a hydrogen or a halogen atom, an optionally substituted hydroxy, mercapto, amino, carboxy or aminocarbonyl group, or a nitro, alkanoyl, aminosulfonyl, alkyl, trifluoromethyl or cyano group, R4 represents a hydrogen atom or an alkyl group, R5 represents a hydrogen or a halogen atom or an alkyl group, A represents a bond or an optionally substituted methylene, ethylene, cycloalkylidene or vinylidene group, B represents a methylene or ethylene group optionally substituted by an alkyl group and W represents a hydrogen or a halogen atom, a cyano, alkanoyl or nitro group, an optionally substituted amino or aminocarbonyl group, a carboxy group, or an ester thereof, a formyl group or an acetal thereof or an optionally substituted alkyl or alkenyl group); and salts thereof formed with acids and bases. Processes for the preparation of the new compounds as well as pharmaceutical compositions containing them are also objects of this invention.

The new compounds show valuable pharmacutical properties, especially effects on intermediary metabolism and a blood-sugar lowering activity.

## **SPECIFICATION**

## Chemical compounds

5 This invention relates to new carboxylic acid amides, to processes for their preparation and to pharmaceutical compositions containing them, and also to their use in the treatment of disorders of intermediary metabolism.

According to one feature of the present invention there are provided compounds of general

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$$R_{4}$$

$$R_{5}$$

$$R_{1}$$

$$R_{2}$$

$$R_{5}$$

$$R_{1}$$

$$R_{2}$$

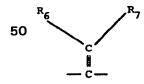
20 20 [wherein R<sub>1</sub> and R<sub>2</sub>, which may be the same or different, each represents an alkyl group containing 1 to 6 carbon atoms or a cycloalkyl group containing 5 to 7 carbon atoms, or R1 and R<sub>2</sub> together with the nitrogen atom to which they are attached represent an unbranched alkyleneimino group containing 3 to 6 carbon atoms optionally substituted by 1 or 2 alkyl

25 groups, each containing 1 to 3 carbon atoms, or by a hydroxy group and in which a methylene group may optionally be replaced by a carbonyl group, by an oxygen or sulfur atom or by an imino group (which may optionally be substituted by an alkyl group containing 1 to 3 carbon atom, an aralkyl group containing 7 to 10 carbon atoms or by a phenyl or halophenyl group) or an ethylene group may optionally be replaced by an O-phenylene group; and unbranched 30 alkenyleneimino group containing 4 to 6 carbon atoms; a saturated or partly unsaturated

azabicycloalkyl group containing 6 to 10 carbon atoms; an aza-1,4-dioxaspiro-alkyl group containing 6 to 8 carbon atoms; or a heptamethyleneimino, octamethyleneimino, nonamethyleneimino or decamethyleneimino group; R<sub>3</sub> represents a hydrogen or halogen atom, a trifluoromethyl, alkyl, hydroxy, alkoxy, alkanoyloxy, mercapto, alkylmercapto, nitro, amino, cyano, alkanoyl, 35 carboxy, alkoxycarbonyl, aminocarbonyl, alkylaminocarbonyl, dialkylaminocarbonyl, aminosulfonyl, alkylamino, dialkylamino, alkanoylamino, alkoxycarbonylamino or alkylsulfonylamino group

(wherein each alkyl part in the above mentioned groups may contain from 1 to 3 carbon atoms), an aralkoxy group containing 7 to 10 carbon atoms or an arylcarbonylamino group; R4 represents a hydrogen atom or an alkyl group containing 1 to 3 carbon atoms; R₅ represents a 40 hydrogen atom, a halogen atom or an alkyl group containing 1 to 3 carbon atoms; A represents 40 a bond, a methylene or ethylene group optionally substituted by an alkyl group containing 1 to 5 carbon atoms, a methylene or ethylene group substituted by two alkyl groups each containing 1 to 3 carbon atoms, a methylene group substituted by a cycloalkyl group containing 3 to 7

carbon atoms or by a hydroxyalkyl, alkoxyalkyl, cyano, carboxyl, alkoxycarbonyl, aminocarbonyl, 45 alkylaminocarbonyl, dialkylaminocarbonyl, aryl or aralkyl group, wherein each of the alkyl parts may contain from 1 to 3 carbon atoms, a cycloalkylidene group containing 3 to 7 carbon atoms or a vinylidene group of formula



55 wherein R<sub>6</sub> and R<sub>7</sub>, which may be the same or different, each represents a hydrogen atom or an 55 alkyl group containing 1 to 3 carbon atoms or one of the radicals R<sub>8</sub> and R<sub>7</sub> represents a cycloalkyl group containing 3 to 7 carbon atoms or an aryl or aralkyl group and the other is as defined above or R<sub>8</sub> and R<sub>7</sub> together with the carbon atom to which they are attached, represent a cycloalkylidene radical containing 5 to 7 carbon atoms; B represents a methylene or ethylene

60 group optionally substituted by an alkyl group containing 1 to 3 carbon atoms and W represents 60 a hydrogen or halogen atom, a nitro group, an amino group (optionally substituted by an alkanoyl group containing 1 to 3 carbon atoms) an alkyl group containing 1 to 3 carbon atoms (optionally substituted by a hydroxy or carboxy group or by one or two alkoxycarbonyl groups containing 2 to 4 carbon atoms each), an alkenyl group containing 2 to 5 carbon atoms

65 substituted by a carboxy or alkoxycarbonyl group containing 2 to 4 carbon atoms, an alkanoyl

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group containing 1 to 3 carbon atoms, a dialkoxymethyl or trialkoxymethyl group containing 1 to 3 carbon atoms in each alkyl part, an alkylenedioxymethyl group containing 2 or 3 carbon atoms in the alkylene part, a 1,3-oxazoline-2-yl or cyano group, an aminocarbonyl group (optionally substituted by one or two alkyl groups containing 1 to 4 carbon atoms in each alkyl part), an unbranched alkyleneiminocarbonyl group containing 5 to 8 carbon atoms a morpholinocarbonyl group, a (dialkyldioxolane-yl)-alkoxycarbonyl group containing 7 to 10 carbon atoms or a carboxy group or esterfied carboxy group wherein if the said ester group consists of an alkyl group containing 1 to 6 carbon atoms this may be substituted, in any but the α-position, by a hydroxy, alkoxy, amino, alkylamino, dialkylamino, 1,3-dimethylxanthine-7-yl, alkanoyloxy, aroy-10 loxy, aralkanoyloxy or pyridine-carbonyloxy group or by two hydroxy groups—except in the case of any methyl or methylene group in the above cases, which can only be substituted by one hydroxy group or by a group of formula

20 wherein A, B, R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub> and R<sub>5</sub> are as hereinbefore defined whereby each alkyl part of the 20 above alkyl ester substituents may contain from 1 to 3 carbon atoms), and salts thereof.

The new compounds possess interesting pharmacological properties, especially in general an effect an intermediary metabolism and in particular a blood-sugar lowering activity.

For pharmaceutical use, the salts referred to above will of course be physiologically compatible salts formed with acids or bases, but other salts may find use in the preparation of the compounds of formula I and their physiologically compatible salts. The term "salts formed with acids or bases" includes salts formed with inorganic or organic acids or bases.

The invention extends to all possible isomers, including optional isomers, of compounds of formula I. R<sub>1</sub> and R<sub>2</sub> together with the nitrogen atom may represent for example, dimethylam-30 ino, diethylamino, dipropylamino, dibutylamino diisobutylamino, dipentylamino, dihexylamino, N-methyl-N-ethylamino, N-methyl-N-propylamino, N-isopropylamino, N-isopropylamino, N-isopropylamino, N-methyl-N-butylamino, N-ethyl-N-butylamino, N-ethyl-N-cyclopentylamino, N-propyl-N-butylamino, N-methyl-N-cyclopentylamino, N

5 N-cyclohexylamino, N-isobutyl-N-cyclohexylamino, pyrrolidino, piperidino, hexamethyleneimino, heptamethyleneimino, octamethylenimino, nonamethyleneimino, decamethyleneimino, dimethylazetidino, methyl-pyrrolidino, dimethyl-pyrrolidino, ethyl-piperidino, dimethyl-piperidino, ethyl-piperidino, propyl-piperidino, methyl-piperidino, propyl-piperidino, methyl-propyl-piperidino, isopropyl-piperidino, cis-3,5-dimethyl-piperidino, trans-3,5-dimethyl-piperidino,

40 piperidino, morpholino, thiomorpholino, piperazino, N-methyl-piperazino, N-ethyl-piperazino, N-propyl-piperazino, N-isopropyl-piperazino, N-benzylpiperazino, N-(2-phenyl)-piperazino, N-(3-phenylpropyl)-piperazino, N-phenyl-piperazino, N-fluorophenylpiperazino, N-chlorophenyl-piperazino, N-bromophenyl-piperazino, hydroxy-pyrrolidino, hydroxy-piperidino, hydroxy-hexamethyleneimino, pyrrolidone-1-yl, piperidone-1-yl, hexahydroazepinone-1-yl, tetrahydro-isoquinoline-2-

45 yl, octahydro-isoquinoline-2-yl, decahydro-isoquinoline-2-yl, dihydro-isoindole-2-yl, hexahydro-isoindole-2-yl, octahydro-isoindole-2-yl, tetrahydro-3-benzazepine-3-yl, decahydro-3-benzazepine-3-yl, 3-aza-bicyco[3.2.0]heptane-3-yl, 3-aza-bicyclo[3.2.1]octane-3-yl, 3-aza-bicyclo[3.3.2]nonane-3-yl, 1,4-dioxa-7-aza-spiro[4,4]nonane-7-yl, 1,4-dioxa-7-azaspiro[4,5]decane-7-yl, 1,4-dioxa-8-aza-spiro[4,5]decane-8-yl, 1,4-dioxa-8-aza-spiro[4,6]undecane-8-yl, pyrrolino or tetrahy-

R<sub>3</sub> may represent, for example, a hydrogen, fluorine, chlorine, bromine or iodine atom, or a methyl, ethyl, propyl, isopropyl, hydroxy, methoxy, ethoxy, propoxy, isopropoxy, acetoxy, propionyloxy, mercapto, methylmercapto, ethylmercapto, propylmercapto, isopropylmercapto, trifluoromethyl, nitro, cyano, formyl, acetyl, propionyl, aminosulfonyl, amino, methylamino, ethylamino, propylamino, dimethylamino, diethylamino, dipropylamino, diisopropylamino, N-methyl-N-ethyl-amino, N-methyl-N-isopropylamino, N-ethyl-N-propylamino, formylamino, acetylamino, propionylamino, methylsulfonylamino, ethylsulfonylamino, propylsulfonylamino, isopropylsulfonylamino, carboxy, methoxycarbonyl, ethoxycarbonyl, propoxycarbonyl, isopropoxycarbonyl, methoxycarbonylamino, ethoxycarbonylamino, propoxycarbonylamino, isopropoxycarbonylamino, benzyloxy, 1-phenylethoxy, 2-phenyl-ethoxy, 3-phenyl-propoxy, amino-

O lamino, benzoylamino, benzyloxy, 1-phenylethoxy, 2-phenyl-ethoxy, 3-phenyl-propoxy, amino-carbonyl, methylaminocarbonyl, ethylaminocarbonyl, isopropylaminocarbonyl, dimethylaminocarbonyl, diethylaminocarbonyl, dipropylaminocarbonyl, methyl-ethylaminocarbonyl, or methyl-propylaminocarbonyl group;

R<sub>4</sub> may represent a hydrogen atom, or a methyl, ethyl, propyl or an isopropyl group; R<sub>5</sub> may represent a hydrogen, fluorine, chlorine, bromine or an iodine atom, or a methyl,

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ethyl, propyl or an isopropyl group;

A may represent, for example, a single bond, or a methylene, ethylidene, ethyl-methylene propyl-methylene, isopropyl-methylene, butyl-methylene, pentyl-methylene, dimethyl-methylene, diethyl-methylene, dipropyl-methylene, methyl-ethylmethylene, methyl-propyl-methylene, ethyl-5 propyl-methylene, ethyl-isopropyl-methylene, ethylene, methylene, ethyl-ethylene, propyl-5 ethylene, dimethylethylene, cyclopropyl-methylene, cyclobutyl-methylene, cyclopentyl-methylene, cyclohexyl-methylene, cycloheptyl-methylene, cyclopropylidene, cyclobutylidene, cyclopentylidene, cyclohexylidene, cycloheptylidene, carboxymethylene, methoxycarbonyl-methylene, ethoxycarbonyl-methylene, propoxycarbonyl-methylene, hydroxymethyl-methylene, 1-hydroxye-10 thyl-methylene, 2-hydroxyethyl-methylene, 1-hydroxypropyl-methylene, 3-hydroxypropyl-methy-10 lene, methoxymethylmethylene, ethoxymethyl-methylene, propoxymethyl-methylene, 1-methoxyethyl-methylene, 2-methoxyethyl-methylene, 2-ethoxyethyl-methylene, cyano-methylene, aminocarbonylmethylene, methylaminocarbonyl-methylene, dimethylaminocarbonyl-methylene, ethylaminocarbonyl-methylene, diethylaminocarbonyl-methylene, propylaminocarbonyl-methylene, 15 phenyl-methylene, benzyl-methylene, 1-phenylethyl-methylene, 2-phenylethyl-methylene, 3-phe-15 nylpropyl-methylene, 2-phenylpropyl-methylene, vinylidene, methyl-vinylidene, dimethyl-vinylidene, ethyl-vinylidene, diethyl-vinylidene, propyl-vinylidene, dipropyl-vinylidene, ethyl-methylvinylidene, ethyl-propyl-vinylidene, methylpropyl-vinylidene, cyclopentyl-vinylidene, cyclohexylvinylidene, phenyl-vinylidene, benzyl-vinylidene, 2-phenethyl-vinylidene, cyclopropylidene-me-20 thylene, cyclopentylidene-methylene, cyclohexylidene-methylene or cycloheptylidene-methylene 20 group; B may represent, for example, a methylene, ethylene, ethylidene, propyl-methylene or isopropyl-methylene group; and W may represent, for example, a hydrogen, chlorine, bromine or iodine atom, or a methyl, ethyl, propyl, isopropyl, hydroxymethyl, 1-hydroxyethyl, 2-25 hydroxyethyl, 1-hydroxypropyl, 3-hydroxypropyl, carboxymethyl, 1-carboxyethyl, 2-carboxyethyl, 3-carboxy-propyl, methoxycarbonyl-methyl, ethoxycarbonyl-methyl, propoxycarbonyl-methyl, 2methoxycarbonyl-ethyl, 2-ethoxycarbonyl-ethyl, 3-ethoxycarbonylpropyl, bis-(methoxycarbonyl)methyl, bis-(ethoxycarbonyl)-methyl, 2,2-bis-(ethoxycarbonyl)-ethyl, carboxyl-vinyl, carboxy-propenyl, carboxy-pentenyl, methoxycarbonyl-vinyl, ethoxycarbonyl-vinyl, propoxycarbonyl-vinyl, for-30 myl, acetyl, propionyl, dimethoxymethyl, diethoxy-methyl, dipropoxy-methyl, trimethoxymethyl, 30 triethoxy-methyl, 1,2-ethylenedioxy-methyl, 1,3-propylenedioxy-methyl, cyano, nitro, amino, formylamino, acetamino, propionylamino, 1,3-oxazoline-2-yl, aminocarbonyl, methylaminocarbonyl, ethylaminocarbonyl, propylaminocarbonyl, isopropylaminocarbonyl, butylaminocarbonyl, dimethylaminocarbonyl, diethylaminocarbonyl, dipropylaminocarbonyl, dibutylaminocarbonyl, pyr-35 rolidinocarbonyl, piperidinocarbonyl, hexamethyleneininocarbonyl, heptamethyleneiminocarbo-35 nyl, morpholinocarbonyl, carboxy, methoxycarbonyl, ethoxycarbonyl, propoxycarbonyl, isopropoxycarbonyl, butoxycarbonyl, tert.butoxycarbonyl, pentoxycarbonyl, hexoxycarbonyl, heptoxycarbonyl, octoxycarbonyl, allyloxycarbonyl, butenyloxycarbonyl, benzyloxycarbonyl, 1-phenylethoxycarbonyl, 2-phenylethoxycarbonyl, 3-phenylpropoxycarbonyl, 2-hydroxyethoxycarbonyl, 2-40 hydroxypropoxycarbonyl, 3-hydroxypropoxycarbonyl, 2-methoxyethoxycarbonyl, 2-ethoxyethoxycarbonyl, (2,2-dimethyl-dioxolane-4-yl)-methoxycarbonyl, 2-(2,2-dimethyl-dioxolane-4-yl)-ethoxycarbonyl, (2,2-diethyl-dioxolane-4-yl)-methoxy-carbonyl, 2-(2,2-diethyl-dioxolan-4-yl)-ethoxycarbonyl, 3-(2,2-dimethyl-dioxolane-4-yl)-propoxycarbonyl, 2-aminoethoxycarbonyl, 2-dimethylaminoethoxycarbonyl, 2-diethylamino-ethoxycarbonyl, 2-(1,3-dimethyl-xanthine-7-yl)-ethoxycarbonyl, 45 2-acetoxy-ethoxycarbonyl, 2-benzyloxy-ethoxycarbonyl, 2-phenylacetoxyethoxycarbonyl, 2-pyridinecarbonyloxy-ethoxycarbonyl, 2,3-dihydroxy-propoxycarbonyl, 3,4-dihydroxy-butoxycarbonyl, 2-[4-[(1,(2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoyloxy]ethoxycarbonyl or 3-[4-[(1-(2-piperidino-pheyl)-ethyl)-aminocarbonylmethyl]-benzoyloxy]propoxycarbonyl group. Preferred compounds of the above general formula I are, however, those wherein R, and R, 50 together with nitrogen atom to which they are attached represent a dialkylamino or N-alkyl-50 cyclohexylamino group, wherein each alkyl part may contain from 1 to 4 carbon atoms, an unbranched alkyleneimino group containing 3 to 6 carbon atoms (optionally substituted by one or two methyl groups), a hydroxypiperidino, piperidone-1-yl, tetrahydro-pyridino, morpholino, thiomorpholino, N-methylpiperazino, N-benzylpiperazino, N-chlorophenyl-piperazino, heptame-55 thyleneimino or octamethyleneimino group, a saturated or partly unsaturated azabicycloalkyl 55 group containing 7 to 9 carbon atoms, an unbranched alkyleneimino group containing 4 to 6 carbon atoms wherein one ethylene group is replaced by a o-phenylene group, or a 1,4-dioxaaza-spiro-alkyl group containing 7 or 8 carbon atoms; R<sub>3</sub> represents a hydrogen, fluorine, chlorine, bromine or iodine atom, or a methyl, trifluorome-60 thyl, hydroxy, methoxy, benzyloxy, acetoxy, mercapto, methylmercapto, nitro, amino dimethy-60

R<sub>4</sub> represents a hydrogen atom or a methyl group; R<sub>5</sub> represents a hydrogen atom, a chlorine atom or a methyl group;

A represents a bond, or a methylene group (optionally substituted by an alkyl group

lamino, acetylamino, methylsulfonylamino, benzoylamino, ethoxy-carbonylamino, cyano, carboxy, methoxycarbonyl, ethoxycarbonyl, aminocarbonyl, acetyl or aminosulfonyl group;

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containing 1 to 3 carbon atoms, or by a phenyl, cyclohexyl, carboxy, methoxycarbonyl or a hydroxymethyl group), a dimethyl-methylene, cyclopropylidene or ethylene group or a vinylidene group of formula

wherein  $R_8$  and  $R_7$ , which may be the same or different, each represents a hydrogen atom or a methyl group or  $R_6$  and

R<sub>7</sub> together with the carbon atom to which they are attached represent a cycloalkylidene radical containing 1 to 3 carbon atoms:

B represents a methylene, ethylidene or ethylene group; and
W represents a hydrogen atom, or a methyl, ethyl, hydroxymethyl, cyano or carboxyvinylene
group, an alkyl group containing 1 to 3 carbon atoms substituted by a carboxy group or by one
or two alkoxycarbonyl groups containing 2 to 4 carbon atoms each, a carbonyl group

(substituted by a hydrogen atom, a methyl, ethyl, hydroxy alkoxy, (2,2-dimethyl-dioxolane-4-yl)20 methoxy, benzyloxy, pyridyl-methyoxy, amino, alkylamino, dialkylamino, piperidino or morpholino group), whereby any alkyl part in the aforementioned groups may contain from 1 to 3
carbon atoms, or a group of formula

$$\begin{array}{c|c}
O \\
25 & || \\
-C - O - (CH_2)_0 - R_8,
\end{array}$$
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wherein n is 2, 3, or 4; and

R<sub>8</sub> represents a hydroxy, methoxy, ethoxy, acetoxy, benzoyloxy, pyridinecarbonyloxy group, a dialkylamino group containing 1 to 3 carbon atoms in each alkyl part, 1 1,3-dimethylxanthiene-30 7-yl group of a group of formula

40 wherein A, B, R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub> and R<sub>5</sub> are as hereinbefore defined; and especially those compounds of general formula I wherein the radical

is in the 2-position and the radical W is in the 4'-position. Especially preferred are compounds 50 of general formula la

$$\begin{array}{c|c}
R_{3} & \stackrel{\text{H}}{\longrightarrow} R_{1} \\
R_{2} & & \\
\end{array}$$
(Ia)

wherein R<sub>1</sub> and R<sub>2</sub> together with the nitrogen atom to which they are attached, represent a 60 dimethylamino, pyrrolidino, methylpyrrolidino, piperidino, methylpiperidino, dimethylpiperidino, 60 tetrahydro-pyridino, 2-octahydroisoindolo or hexamethyleneimino group; R<sub>3</sub> represents a hydrogen, fluorine or a chlorine atom or a methyl group;

A represents a methylene group (optionally substituted by a cyclohexyl, phenyl, methoxycarbonyl, ethoxycarbonyl or an alkyl group containing 1 to 3 carbon atoms), or a dimethylmethy-65 lene group or a vinylidene group of formula

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wherein  $R_{\rm g}$  and  $R_{\rm g}$  each represents a hydrogen atom or together with the carbon atom to which they are attached represent a cyclohexylidene group; and W represents a methyl, hydroxymethyl 10 or a carboxymethyl group, or a carbonyl group (substituted by a hydrogen atom, a methyl, hydroxy, methoxy, ethoxy, propoxy, isopropoxy, 2-hydroxyethoxy, 2-methoxyethoxy, (2,2dimethyl-dioxolane-4-yl)-methoxy, or a 2-diethylaminoethoxy group). The compounds of formula I may, for example, be prepared by the following processes, which

processes constitute further features of the present invention:

(a) Acylation of an amine of general formula II

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25 wherein A, R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub> and R<sub>4</sub> are as hereinbefore defined, (or if A represents one of the above 25 mentioned vinylidene groups one of its tautomers, or its lithium or magnesium halide complex) with a carboxylic acid of general formula III

.(III)

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35 wherein R<sub>6</sub> and B are as hereinbefore defined and W' represents W as hereinbefore defined or represents a carboxyl group protected by a protective radical, or with reactive derivatives thereof optionally prepared in the reaction mixture.

Suitable reactive derivatives of a compound of general formula III includes, for example, ester (such as the methyl, ethyl or benzyl ester), thioesters (such as the methylthio or ethylthioester),

40 halides (such as the acid chloride), anhydrides or imidazolides thereof. The reaction is conveniently carried out in a solvent, such as for example methylene chloride, chloroform, carbon tetrachloride, ether, tetrahydrofuran, dioxan, benzene, toluene, acetonitrile or dimethylformamide, optionally in the presence of an acid-activating or a dehydrating agent, (e.g. in the presence of ethyl chloroformate, thionyl chloride, phosphorous trichloride, phosphorus pentox-

45 ide, N,N'-dicyclohexylcarbodiimide, N,N-dicyclohexylcarbodiimide/N-hydroxy-succinimide, N,N'carbonyldiimidazole, N,N'-thionyldiimidazole, or triphenyl phosphine/carbon tetrachloride), or of an agent activating the amino group (e.g. phosphorous chloride) and optionally in the presence of an inorganic base such as, for example, sodium carbonate or a tertiary organic base such as triethyl-amine or pyridine, which simultaneously may serve as a solvent, at temperatures

50 between - 25 and 250°C, preferably, however, at temperatures between - 10°C and the boiling temperature of the used solvent. The reaction may also be carried out without a solvent. Furthermore, the water which is formed during the reaction may be removed by azeotropic distillation (e.g. by heating with toluene in a water separator funnel) or by addition of a drying agent such as magnesium sulfate or a molecular sieve.

If necessary, the subsequent removal of a protective radical is preferably carried out hydrolytically, conveniently in the presence of either an acid (such as, for example, hydrochloric, sulfuric, phosphoric or trichloroacetic acid) or a base such as sodium hydroxide or potassium hydroxide in a solvent such as for example water, methanol, ethanol, ethanol/water, water/isopropanol or water/dioxan at temperature between - 10 and 120°C, e.g. at temperatures 60 between room temperature and the boiling temperature of the reaction mixture. A tert.butyl radical used as protective radical may also be removed thermolytically (optionally in an inert

solvent such as methylene choride, chloroform, benzene, toluene, tetrahydrofuran or dioxan) and preferably in the presence of a catalytic amount of an acid such as, for example, ptoluenesulfonic, sulfuric, phosphoric or polyphosphoric acid.

Furthermore, a benzyl radical used as protective radical may also be removed hydrogenolyti-

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cally (in the presence of a hydrogenation catalyst such as palladium/charcoal) in a solvent such as, for example, methanol, ethanol/water, glacial acetic acid, ethyl acetate, dioxan or dimethyl formamide.

(b) For the preparation of compounds of general formula I, wherein W represents a carboxy group:

Cleavage of a compound of general formula IV

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$$R_3$$
 $R_4$ 
 $R_5$ 
 $R_1$ 
 $R_5$ 
 $R_1$ 
 $R_2$ 
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wherein R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>5</sub>, A and B are defined as mentioned before and D represents a group which may be converted into a carboxy group by means of hydrolysis, thermolysis or hydrogenolysis.

Suitable hydrolysable groups include, for example, carboxy derivatives (such as unsubstituted or substituted amides, esters, thioesters, orthoesters, iminoethers, amidines or anhydrides), a nitrile group, a malonic ester-(1)-yl group, a tetrazolyl group or an optionally substituted 1,3oxazole-2-yl or 1,3-oxazoline-2-yl group.

Suitable thermolytically cleavable groups include, for example, esters with tertiary alcohols,

25 e.g. the tert.butyl ester.

Suitable hydrogenolytically cleavable groups include, for example, aralkyl groups, e.g. the

benzyl group.

The hydrolysis is conveniently carried out either in the presence of an acid (such as for example, hydrochloric, sulfuric, phosphoric or trichloroacetic acid) or a base (such as sodium 30 hydroxide or potassium hydroxide) in a solvent such as, for example, water/methanol, ethanol, water/ethanol, water/isopropanol or water/dioxan at temperatures between - 10 and 120°C, e.g. at temperatures between room temperature and the boiling temperature of the reaction mixture.

Thus if, for example, D in a compound of general formula IV represents a nitrile or 35 aminocarbonyl group, these groups may be converted into a carboxy group with a nitrite, e.g. sodium nitrite, in the presence of an acid (such as sulfuric acid), whereby conveniently this acid is simultaneously used as a solvent, at temperatures between 0 and 50°C; if for example, D represents a tert.butyloxycarbonyl group, the tert.butyl group may be split off thermolytically (optionally in an inert solvent such as methylene chloride, chloroform, benzene, toluene,

40 tetrahydrofuran or dioxan) and preferably in the presence of a catalytic amount of an acid such as p-toluenesulfonic, sulfuric, phosphoric or polyphosphoric acid preferably at the boiling temperature of the used solvent, e.g. at temperatures between 40 and 100°C; or if for example D represents a benzyloxycarbonyl group, the benzyl group may be split off hydrogenolytically in the presence of a hydrogenation catalyst such as palladium/charcoal in a solvent such as for

45 example, methanol, ethanol/water, glacial acetic acid, ethyl acetate, dioxan or dimethyl 45 formamide preferably at temperatures between 0 and 50°C, e.g. at room temperature, and at a hydrogen pressure of 1 to 5 bar. During the hydrogenolysis other groups may optionally simultaneously be reduced, e.g. a halogen compound may be dehalogenated, a nitro group may be converted into the corresponding amino group, or a vinylidene group into the corresponding 50 alkylidene group.

Reaction of a compound, optionally formed in the reaction mixture, of general formula V

wherein R<sub>3</sub>, R<sub>4</sub>, R<sub>5</sub>, A, B, and W are as hereinbefore defined and, R<sub>2</sub>' represents a hydrogen atom or has the meanings mentioned before for R2, with a compound of general formula VI

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## $R_1'-E$ (VI)

[wherein R<sub>1</sub>' has the meanings mentioned before for R<sub>1</sub> or together with the radical R<sub>2</sub>' of formula V represents a straight-chained alkylene group containing 4 to 6 carbon atoms

5 (optionally substituted by one or two alkyl groups containing 1 to 3 carbon atoms) or an nephylene group in which the third methylene group is replaced by an oxygen or sulfur atom, and E represents a nucleophilically exchangeable group such as a halogen atom or a sulfonyloxy group (e.g. a chlorine, bromine or an iodine atom or a methanesulfonyloxy or p-toluenesulfonyloxy group), or also a hydrogen atom if in R<sub>1</sub>' one methylene group is replaced by an aldehyde or ketone carbonyl group], if necessary in the presence of a reducing agent, and optional subsequent hydrolysis.

Suitable alkylating agents of formula VI include, for example, the corresponding halides or sulfates such as methyl iodide, ethyl iodide, propyl bromide, dimethyl sulfate or diethyl sulfate.

The reaction is conveniently carried out in a solvent such as, for example, acetone, tetrahydrofuran, dimethyl formamide, dimethylsulfoxide or hexamethyl phosphoric acid triamide, optionally in the presence of an inorganic base (such as sodium carbonate, potassium carbonate or potassium tert.butylate) or tertiary organic base (such as pyridine) at temperatures between 0 and 150°C; preferably, however, at temperatures between 20 and 75°C. If a compound of general formula V is used wherein W represents a carboxyl group, this carboxyl group may simultaneously be converted into the corresponding ester depending on the reaction conditions, e.g. at temperatures above room temperature and in the presence of a base, for example sodium

carbonate.

The methylation may optionally also be carried out so that a compound of general formula V is reacted with formalin in the presence of a reducing agent, e.g. formic acid or hydrogen in the presence of a hydrogenation catalyst (e.g. palladium or platinum), optionally in a solvent such as 25 formic acid or glacial acetic acid at temperatures up to the boiling temperatures of the reaction mixture.

Moreover, the alkylation may optionally also be carried out with a corresponding carbonyl compound in the presence of a hydride such as sodium cyanoborohydride in a solvent such as for example acetonitrile/glacial acetic acid or dimethyl formamide/acetic acid preferably at pH 7 and at temperatures between 0 and 50°C.

The subsequent hydrolysis is preferably carried out in an aqueous solvent such as water/-methanol, water/ethanol or water/dioxan in the presence of an acid (such as hydrochloric or sulfuric acid) or a base (such as sodium or potassium hydroxide) at temperatures between 50 and 100°C.

(d) For the preparation of compounds of general formula I wherein W represents a carboxy group, an alkanoyl group containing 1 to 3 carbon atoms or an alkyl group containing 1 to 3 carbon atoms:

Reaction of a compound of general formula VII

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$$R_{3} = \begin{bmatrix} R_{4} & R_{4} & R_{5} \\ R_{1} & R_{2} \end{bmatrix}$$
, (VII)

50 wherein R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub>, R<sub>5</sub>, A and B are as hereinbefore defined, with phosgene, an oxalyl halide, an alky or alkanoyl halide containing 1 to 3 carbon atoms in the alkyl part or with hydrogen cyanide and a hydrogen halide (preferably hydrogen chloride), in the presence of a Lewis acid.

Suitable halides include chlorides and bromides, and the Lewis acid is preferably aluminium

The reaction is preferably carried out in a solvent such as methylene chloride, nitrobenzene, chlorobenzene, dichlorobenzene, tetrachloroethane or carbon disulfide or in polyphosphoric acid at temperatures between 0 and 120°C, preferably, however at temperatures between 20 and 80°C. If in a compound of, general formula VII, R<sub>3</sub> represents a hydrogen atom, this may simultaneously be replaced by a corresponding alkyl or acyl radical.

(e) For the preparation of compounds of general formula I wherein w represents a carboxy

Reaction of a compound of general formula VIII

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wherein R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub>, R<sub>5</sub>, A and B are as hereinbefore defined, with a hypohalide optionally prepared in the reaction mixture. The reaction is conveniently carried out in a solvent (such as for example water/tetrahydrofuran or water/dioxan) and in the presence of a base (such as sodium hydroxide or potassium hydroxide) at temperatures between 0 and 80°C; preferably, 15 however, at temperatures between 25 and 50°C.

(f) For the preparation of compounds of general formula I wherein W represents a carboxy group:

Oxidation of compound of general formula IX

20
$$R_{4}$$

$$R_{7}$$

$$R_{1}$$

$$R_{8}$$

$$R_{1}$$

$$R_{2}$$

$$R_{3}$$

$$R_{1}$$

$$R_{2}$$

$$R_{3}$$

$$R_{4}$$

$$R_{5}$$

$$R_{5}$$

$$R_{5}$$

$$R_{5}$$

wherein R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub>, R<sub>5</sub>, A and B are as hereinbefore defined and G represents a group which 30 may be converted by means of oxidation into a carboxy group.

Such an oxidizable group includes for example a formyl group or one of its acetals, a hydroxymethyl group or one of its ethers, or an unsubstituted or substituted acyl group (such as an acetyl, chloroacetyl, propionyl, malonic acid-(1)-yl group or a malonic ester-(1)-yl group).

The reaction is carried out by means of an oxidizing agent in a solvent (such as for example water, glacial acetic acid, pyridine or carbon tetrachloride) at temperature between 0 and 100°C, conveniently, however, at temperatures between 20 and 50°C. The reaction is preferably carried out with silver oxide/sodium hydroxide solution, manganese dioxide/acetone or methylene chloride, hydrogen peroxide/sodium hydroxide solution, bromine or chlorine/sodium or potassium hydroxide solution or chromium trioxide/pyridine.

O (g) For the preparation of compounds of general formula I, wherein R<sub>3</sub> represents a nitro group:

Reaction of a compound of general formula X

45
$$R_{3} \longrightarrow X$$

$$A - N - CO - B$$

$$R_{5} \longrightarrow X$$

(wherein R<sub>4</sub>, R<sub>5</sub>, A, B and W are as hereinbefore defined, R<sub>3</sub> represents a nitro group and Y represents a nucleophilically exchangeable radical such as a halogen atom) with an amine of general formula XI

(wherein R<sub>1</sub> and R<sub>2</sub> are defined as mentioned before), and optional subsequent hydrolysis.

The term "a halogen atom" used in the definition of the exchangeable radical Y particularly represents a fluorine, chlorine or a bromine atom, and preferably in the o- or p-position relative to the nitro group.

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The reaction is conveniently carried out in a solvent such as for example, water, water/methanol, water/ethanol, water/isopropanol, water/dioxan, methanol, ethanol, dimethyl formamide, or in an excess of the amine of general formula XI and/or the N-formyl derivate thereof (optionally in the presence of an inorganic or tertiary organic base), optionally in the presence of 5 a reaction accelerator such as copper or a copper salt and optionally in a closed vessel at temperatures between 20 and 150°C; preferably, however at the boiling temperature of the reaction mixture (e.g. at 100°C). The reaction may, however, be carried out without a solvent.

The optional subsequent hydrolysis is conveniently carried out in an aqueous solvent such as for example methanol/water, ethanol/water or dioxan/water in the presence of an acid (such as 10 hydrochloric or sulfuric acid) or a base such as sodium or potassium hydroxide at temperatures between 50 and 100°C.

(h) For the preparation of compounds of general formula I, wherein A represents a group of formula

wherein R<sub>6</sub> and R<sub>7</sub> are as hereinbefore defined: Reduction of an enamide of general formula XII

25
$$R_{6}$$
 $R_{7}$ 
 $R_{4}$ 
 $R_{5}$ 
 $R_{1}$ 
 $R_{1}$ 
 $R_{2}$ 
 $R_{3}$ 
 $R_{4}$ 
 $R_{5}$ 
 $R_{5}$ 
 $R_{5}$ 
 $R_{6}$ 
 $R_{1}$ 
 $R_{2}$ 
 $R_{3}$ 

wherein R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>5</sub>, R<sub>6</sub>, R<sub>7</sub>, B and W are as hereinbefore defined.

The reduction is preferably carried out with hydrogen in the presence of a hydrogenation catalyst such as palladium/charcoal or platinum in a solvent such as for example methanol, ethanol, isopropanol, ethanol/water glacial acetic acid, ethyl acetate, dioxan, tetrahydrofuran, dimethyl formamide, benzene, or benzene/ethanol at temperatures between 0 and 100°C, preferably, however at temperatures between 20 and 50°C, and a hydrogen pressure of 1 to 5 40 bar. When using a chiral hydrogenation catalyst such as a transition metal  $\pi$ -complex, e.g. a complex made from rhodium chloride and (+) or (-) 0,0-isopropylidene-2,3-dihydroxy-1,4-bis-(diphenylphosphino)-butane ( = DIOP), the hydrogenation is effected enantioselectively. Moreover, other reduceable groups may be reduced during the catalytic hydrogenationm e.g. a nitro group to an amino group or a chlorine or a bromine atom to a hydrogen atom.

(i) For the preparation of compounds of general formula I, wherein R4 represents a hydrogen atom and A represents a methylene or ethylene group (optionally substituted by an alkyl group containing 1 to 5 carbon atoms), a methylene or ethylene group substituted by two alkyl groups containing 1 to 3 carbon atoms each, a methyl group (substituted by a cycloalkyl group containing 3 to 7 carbon atoms, by an alkoxyalkyl, carboxyl, alkoxycarbonyl, aryl or an aralkyl 50 group, whereby each of the aforementioned alkyl parts may contain from 1 to 3 carbon atoms), 50 or a cycloalkylidene group 4 to 7 carbon atoms:

Reaction of a compound of general formula XIII

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[wherein R<sub>1</sub>, R<sub>2</sub> and R<sub>3</sub> are as hereinbefore defined and A' represents a methylene or ethylene group (optionally substituted by an alkyl group containing 1 to 5 carbon atoms), a methylene or ethylene group substituted by two alkyl groups containing 1 to 3 carbon atoms each, a 65 methylene group (substituted by a cycloalkyl group containing 3 to 7 carbon atoms, an

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alkoxyalkyl, carboxyl, alkoxycarbonyl, aryl, or an aralkyl group, whereby each of the above mentioned alkyl parts may contain from 1 to 3 carbon atoms), or a cycloalkylidene group containing 4 to 7 carbon atoms], with a compound of general formula XIV

wherein  $R_s$ , B and W are as hereinbefore defined. The reaction is carried out in the presence of a strong acid, which simultaneously may serve as solvent, preferably in concentrated sulfuric acid, at temperatures between 20 and 150°C,

preferably at temperatures between 80 and 100°C.

	preferably at temperatures between 80 and 100 C.	
15	According to a further feature of the present invention, a compound of general formula I thus obtained wherein W represents the carboxy group, may if desired, subsequently be converted	15
	into a corresponding compound of general formula I wherein W represents an ester or amide	
	group by esterification or amidation and/or a compound of general formula I wherein R <sub>3</sub> and/or	
	W represent(s) a nitro group, may subsequently be converted by reduction into a corresponding	
20	compound of general formula I wherein R <sub>2</sub> and/or W represent(s) an amino group; and/or a	20
	compound of general formula I wherein R <sub>2</sub> and/or W represent(s) amino group, may subse-	
	quently be converted via a corresponding diazonium salt into a corresponding compound of	
	general formula I wherein R, represents a hydrogen or a halogen atom, a hydroxy, alkoxy,	
	mercanto, alkylmercanto, chlorosulfonyl, or cyano group and/or W represents a hydrogen or a	
25	halogen atom or a cyano group. Optionally a compound of general formula I thus obtained,	25
	wherein R, represents a hydroxy group, may subsequently be converted by alkylation into a	
	corresponding compound of general formula I wherein R <sub>3</sub> represents an alkoxy group, or a	
•	compound of formula I thus obtained, wherein R <sub>s</sub> represents a chlorosultonyl group, may	
	subsequently be converted by ammonia into a corresponding compound of general formula I	
30	wherein R <sub>3</sub> represents an aminosulfonyl group; and/or a compound of general formula I	30
	wherein R <sub>3</sub> represents an amino group may subsequently be converted by means of acylation	
	into a corresponding compound of general formula I wherein R <sub>3</sub> represents an alkanoylamino,	
	aroylamino, alkoxycarbonylamino or an alkylsulfonylamino group; and/or a compound of	
	general formula I wherein R <sub>3</sub> represents an amino may subsequently be converted by means of	25
35	alkylation into a corresponding compound of general formula I wherein R <sub>3</sub> represents an	35
	alkylamino or a dialkylamino group; and/or a compound of general formula I wherein R <sub>3</sub>	
	represents a chlorine or a bromine atom may subsequently converted by means of dehalogena-	
	tion into a corresponding compound of general formula I wherein R <sub>3</sub> represents a hydrogen	
	etom; and/or a compound of general formula I wherein R <sub>3</sub> represents a nitrile group may	40
40	subsequently be converted by means of hydrolysis or alcoholysis into a corresponding	40
	compound of general formula I, wherein R <sub>3</sub> represents an aminocarbonyl, carboxy or an alkoxycarbonyl group; and/or a compound of general formula I wherein R <sub>3</sub> represents a carboxy	
	or alkoxycarbonyl group and/or W represents an (optionally esterified) carboxy group may	
	subsequently be converted by means of reduction into a corresponding compound of general	
AE	formula I wherein R <sub>3</sub> and/or W represents a formyl or hydroxymethyl group; and/or a	45
49	compound of general formula I wherein W represents an alkoxycarbonyl group (wherein the	. •
	alkoxy group may contain from 2 to 6 carbon atoms) substituted in any but the $\alpha$ -position by a	
	hydroxy group may be converted into a compound of general formula I wherein the said	
	hydroxy group is replaced by an acyloxy group, by acylation; and/or a compound of general	
50	formula I, wherein W represents a hydroxymethyl group may subsequently be converted (via a	50
•	corresponding halomethyl compound) by reaction with a malonic acid diester, into a correspond-	
	ing compound of general formula I wherein W represents an ethyl group substituted by two	
	alkoxycarbonyl groups: and/or a compound of general formula I wherein W represents a formyl	
	group may subsequently be converted by condensation and optional subsequent hydrolysis	
55	and/or decarboxylation into a corresponding compound of general formula I wherein W	55
	represents a vinyl group substituted by a hydroxycarbonyl or alkoxycarbonyl group; and/or a	
	compound of general formula I wherein W represents an ethyl group substituted by two	
	alkoxycarbonyl groups may subsequently be converted by hydrolysis and decarboxylation into a	
	corresponding compound of general formula I wherein W represents an ethyl group substituted	
60	by a carboxy group; and/or a compound of general formula I wherein W represents a carboxy	60
	group may subsequently be converted via a sulfonic acid hydrazide and subsequent dispropor-	
	tionation into a corresponding compound of general formula I wherein W represent a formyl	
	group; and/or a compound of general formula I wherein R <sub>1</sub> and R <sub>2</sub> together with the nitrogen	
^-	atom to which they are attached represent an aza-1,4-dioxa-spiro-alkyl group containing 6 to 8	Q E

65 carbon atoms, may subsequently be converted by means of hydrolysis in the presence of an acid 65

5	into a corresponding compound of general formula I wherein R <sub>1</sub> and R <sub>2</sub> together with the nitrogen atom to which they are attached represent an unbranched alkyleneimino group containing 4 to 6 carbon atoms wherein a methylene group is replaced by a carbonyl group; and/or a compound of general formula I wherein R <sub>1</sub> and R <sub>2</sub> together with the nitrogen atom to which they are attached represent an unbranched alkyleneimino group containing 4 to 6 carbon atoms, wherein a methylene group is replaced by a carbonyl group, may subsequently be converted by means of reduction into a corresponding hydroxy-alkyleneimino compound of general formula I; and/or a compound of general formula I wherein W represents an	5
10	aminocarbonyl group may subsequently be converted by means of denydration into a corresponding compound of general formula I wherein W represent a cyano group.  The dehydratation is preferably carried out with a dehydrating agent such as for example phosphorus pentoxide, sulfuric acid or p-toluene sulfonic acid chloride optionally in a solvent such as methylene chloride or pyridine at temperatures between 0 and 100°C, preferably, at	10
15	temperatures between 20 and 80°C.  The esterification is conveniently carried out in a solvent, such as, for example, the corresponding alcohol, pyridine, toluene, methylene chloride, tetrahydrofuran or dioxan, in the presence of an acid-activating and/or dehydrating agent such as thionyl chloride, ethyl chloroformate, carbonyl diimidazole, N,N'-dicyclohexylcarbodiimide or the isourea ether thereof, optionally in the presence of a reaction accelerator such as copper chloride or by transesterifica-	15
20	tion, e.g. with a corresponding carbonic acid diester, at temperatures between 0 and 100 C, preferably, however, at temperature between 20°C and the boiling temperature of the	20
25	The amidation is conveniently carried out in a solvent such as methylene chloride, chloroform, carbon tetrachloride, ether, tetrahydrofuran, dioxan, benzene, toluene, acetonitrilie or dimethyl formamide, optionally in the presence of an acid activating agent or a dehydrating agent, e.g. in the presence of ethyl chloroformate, thionyl chloride, phosphorus trichloride, phosphorus pentoxide, N,N'-dicyclohexyl carbodiimide, N,N'-dicyclohexyl carbodiimide/N-hydroxy succinim-	25
30	ide, N,N'-carbonyldiimidazole, N,N'-thionyliddmidazole, or triphenyl phosphine/carbon tetrachlo- ride, or of an agent activating the amino group, e.g. phosphorus trichloride, and optionally in the presence of an inorganic base such as sodium carbonate or a tertiary organic base such as triethylamine or pyridine, which simultaneously may serve as solvent, at temperatures between  — 25 and 250°C, preferably, however, at temperatures between — 10°C and the boiling	30
.35	temperature of the used solvent. The reaction may also be carried out without a solvent.  Moreover the water, which is formed during the reaction, may be removed by means of azeotropic distillation, e.g. by heating with toluene in a water separator funnel, or by addition of a during egent such as magnesium sulfate or a molecular sieve.	35
40	The reduction of the nitro compound is preferably carried out in a solvent such as water, water/ethanol methanol, glacial acetic acid, ethyl acetate or dimethyl formamide appropriately with hydrogen in the presence of a hydrogenation catalyst such as Raney-nickel, platinum or palladium/charcoal, with metals such as iron, tin or zinc in the presence of an acid, with metal salts such as iron(II)sulfate, tin(II)chloride or sodium dithionite, or with hydrazine in the presence of Raney-nickel at temperatures between 0 and 50°C, preferably, however, at room tempera-	40
45	The reaction of the diazonium salt, (e.g. the fluoroborate, the hydrosulfate in sulfuric acid, the hydrochloride or the hydroiodide) is carried out, if necessary, in the presence of copper or a corresponding copper (I) salt such as copper (I) chloride/hydrochloric acid, copper (I) bromide/hydrobromic acid, trisodium copper(1)tetracyanide at pH 7, or an alkali metal xantogenate, or copper (II) chloride/sulfur dioxide in glacial acetic acid optionally with the addition of	45
50	magnesium chloride, at slightly elevated temperatures, e.g. at temperatures between 15 and 100°C. The subsequent reaction with hypophosphorous acid is preferably carried out at -5 to 0°C. The diazonium salt is conveniently prepared in a solvent such as, for example water/hydrochloric acid, at the said methanol /hydrochloric acid, ethanol /hydrochloric acid, acid, methanol /hydrochloric acid,	50
55	tetrahydrofuran or in an excess of the used acylating agent e.g. formic acid, acetic acid or	55
60	inorganic or a tertiary organic base, which simultaneously may serve as solvent, and optionally in the presence of an acid-activating agent or of a dehydrating agent at temperatures between — 25 and 150°C, preferably, however, at temperatures between — 10°C and the boiling	60
65	The N-alkylation is conveniently carried out with a corresponding halide or sulfonic acid ester, (e.g. methyl iodide, dimethyl sulfate, ethyl bromide or p-toluenesulfonic acid ethyl ester), optionally in the presence of a base such as sodium hydride, potassium hydroxide or potassium tert.butylate and preferably in a solvent such as for example, diethyl ether, tetradhydrofuran,	65

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5	dioxan, ethanol, pyridine or dimethyl formamide, at temperatures between 0 and 75°C; preferably, however, at room temperature. The methylation may, also be carried out with formaldehyde/formic acid (appropriately at the boiling temperature of the reaction mixture) and the alkylation may be carried out with a corresponding carbonyl compound in the presence of a hydride such as sodium cyanoborohydride in a solvent such as acetonitrile acetic acid or dimethyl formamide/acetic acid preferably at pH 7 and at temperatures between 0 and 50°C. The dehalogenation is conveniently carried out in a solvent such as methanol, ethanol, ethyl acetate, glacial acetic acid or dimethyl formamide by means of catalytically activated hydrogen,	5
10	e.g. with hydrogen in the presence of platinum or palladium/charcoal, at temperatures between 0 and 75°C, preferably, however, at room temperature, and at a hydrogen pressure of 1–5 bar. The hydrolysis is conveniently carried out either in the presence of an acid such as hydrochloric sulfuric, phosphoric, polyphosphoric or trifluoroacetic acid or in the presence of a	10
15	base such as sodium hydroxide or potassium hydroxide in a solvent such as for example, water, ethanol, water/ethanol, water/isopropanol or water/dioxan at elevated temperatures, e.g. at the boiling temperature of the reaction mixture. The hydrolysis can however, be also carried out with a nitrite, e.g. sodium nitrite, in the presence of an acid such as sulfuric acid, whereby this may conveniently serve simultaneously as solvent, at temperatures between 0 and 50°C. The subsequent alcoholysis is conveniently carried out in the presence of a hydrogen halide, e.g. hydrogen chloride, at tmeperatures between 20°C and the boiling temperature of the used	15
20	alcohol.  The reduction is preferably carried out with a metal hydride, e.g. with a complex metal hydride such as lithium aluminium hydride, in a solvent such as, for example, diethyl ether, tetrahydrofuran or dioxan at temperatures between 0 and 100°C, preferably however, at temperature between 20 and 60°C.	20
25	The 0-alkylation is conveniently carried out with a corresponding halide, sulfonic acid ester or diazoalkane, e.g. with methyl iodide, dimethyl sulfate, ethyl bromide, p-toluene sulfonic acid ethyl ester, methanesulfonic acid isopropyl ester or diazomethane optionally in the presence of a base such as sodium hydride, potassium hydroxide or potassium-tert. butylate and preferably in	25
30	a solvent such as diethyl ether, tetrahydrofuran, dioxan, methanol, ethanol, pyridine or dimethyl formamide at temperatures between 0 and 75°C, preferably, however, at room temperature.  The conversion of a hydroxymethyl group into a halomethyl group is carried out with a halogenating agent such as for example, thionyl chloride, phosphorus tribromide or phosphorus pentachloride in a solvent such as methylene chloride, carbon	30
35	tetrachloride, benzene or nitrobenzene and subsequent reaction with a malonic acid ester, e.g. with an alkali salt of the malonic acid diethyl ester, at temperatures between 0 and 100°C, preferably, however, at temperatures between 20 and 50°C.	35
40	The condensation of a formyl compound is conveniently carried out in a solvent such as pyridine or tertahydrofuran with malonic acid, with a malonic acid ester, with a dialkylphosphonoacetic acid ester or an alkoxycarbonylmethylene-triphenyl-phosphone, optionally in the presence of a base as a condensation agent, e.g. in the presence of piperidine, potassiumtert.butylate or sodium hydride, at temperatures between 0 and 100°C. By subsequent acidification, (e.g. with hydrochloric or sulfuric acid) or by subsequent alkaline hydrolysis, the desired acid is obtained.	40
45	The hydrolysis is decarboxylation is conveniently carried out in the presence of an acid such as hydrochloric, sulfuric, phosphoric, polyphosphoric or trifluoroacetic acid in a solvent such as for example, water, ethanol, water/ethanol, water/isopropanol or water/dioxan at elevated temperatures, e.g. at the boiling temperature of the reaction mixture.	45
50	The disproportonation of a sulfonic acid hydrazide, which is obtained by reacting the corresponding hydrazine with the corresponding reactive carboxylic acid derivative, is carried out in the presence of a base such as sodium carbonate in a solvent such as ethylene glycol at temperatures between 100 and 200°C, preferably, however, at 160–170°C.  The compounds of general formula I obtained by the above processes may if desired be	50
55	converted into their addition salts, especially into their physiologically compatible salts with inorganic or organic acids or bases by conventional methods such as by reacting the compounds as bases with a solution of the corresponding acids in a suitable solvent, or by reacting the compounds as acids with a solution of the corresponding bases in a suitable solvent. Suitable acids include, for example, hydrochloric acid, hydrochloric acid, hydrobromic acid sulfuric acid, phosphoric acid, lactic acid, citric acid, tartaric acid, succinic acid, maleic acid and fumaric acid.	55
60	Suitable bases include, for example, sodium or potassium hydroxide and cyclohexylamine.	60
65	obtained by reduction of the corresponding nitro compound, for example by means of catalytically activated or nascent hydrogen or by means of sodium dithionite or by reaction of the corresponding compound by a Hofmann, Curtius, Lossen, or Schmidt reaction.	65

	For example a compound of general formula II, wherein, A represents a vinylidene group or the tautomeric ketimine can be obtained by reaction of the corresponding nitrile with the corresponding Grignard or lithium compound and subsequent hydrolysis or by reaction of the corresponding ketone with the corresponding amine in the presence of titanium tetrachloride.	E
5	For further reaction with a compound of general formula III of its reactive derivations, especially acid chlorides, an organometallic complex can be used.  For example a compound of general formula II, wherein A does not represent a bond or a for example a compound by reduction of the corresponding nitrile with lithium	5
10	aluminium hydride, by reaction of the corresponding filline with the corresponding filline wi	10
15	hydrogen, with a complex metal hydrode of with hydrogen, by reaction of the corresponding hydrazinolysis of the corresponding phthalimido compound, by reaction of the corresponding ketone with ammonium formate and subsequent hydrolysis or with a ammonium salt in the presence of sodium cyanoborohydride, by reduction of the corresponding oxime with lithium aluminium hydride, with catalytically activated or nascent hydrogen, by reduction of the corresponding N-benzyl or N-1-phenylethyl Schiff's base, e.g. with a complex metal hydride in 78° and the holling temperature of the	15
20	ether or tetrahydrofuran at temperatures between — 78 and the bolding temperature of a solvent and subsequent cleavage of the benzyl or 1-phenylethyl group by means of catalytic hydrogenation by Ritter reaction of a corresponding alcohol and potassium cyanide in sulfuric acid, or by a Hofmann, Curtius, Lossen or Schmidt reaction. An amine of general sulfuric acid, or by a Hofmann, Curtius, Lossen by resolved, e.g., by fractional crystallization of	20
25	the diastereoisomeric salts using optionally active acids and subsequent decomposition in the formation of diastereoisomeric compounds, their separation and subsequent resolution into enantiomers. Furthermore, an optionally active amine of general formula II can also be prepared by enantioselective reduction of the corresponding ketimine by means of the program of the bydride hydrogen atoms are	25
30	replaced by optically active alcoholate radicals, or by means of hydrogen in the property of suitable chiral hydrogenation catalyst, or in an analogous manner starting from an N-benzyl or suitable chiral hydrogenation catalyst, or in an analogous manner starting from an N-benzyl or optionally active N-1-phenethyl Schiff's base and optionally subsequent cleavage of the benzyl or 1-phenethyl radical.	30
35	by reduction of the corresponding N-acyl compound, e.g. by means of intrical statements hydride.  The compounds of general formulae IV, V, and VII to X used as starting materials may each be obtained by reaction of an amine with a carboxylic acid or one of its reactive derivatives and be obtained by reaction of an amine with a carboxylic acid or one of its reactive derivatives and be obtained by Friedel-	35
40	Crafts acetylation of the corresponding acetyl-unsubstituted compound.  A compound of general formula XII used as a starting material can be obtained preferably by acrylation of the corresponding ketimine or tautomeric forms with the corresponding carboxylic acid or one of its reactive derivatives.	40
45	A compound of general formula XIII used as a starting matched formula that used as a starting matched formula formula in the corresponding Grignard or lithium reagent. The compounds of general formula I posses valuable pharmacological properties, and in general show beneficial effects on intermediary metabolism, and especially, however, a blood-sugar lowering activity.  For example the following compounds have been tested with regard to their biological	45
50	properties:  A = 4-[2-Pyrrolidino-benzyl)-aminocarbonylmethyl]benzoic acid,  B = 4-[(1-(2-Pyrrolidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid,  C = 4-[(1-(5-Chloro-2-pyrrolidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid,	50
5!	E = 4-[(1-(2-Piperidino-phenyl)-ethenyl)-aminocarbonylmethyl]benzoic acid, F = 4-[(1-(6-Chloro-2-piperidino-phenyl)-ethenyl)-aminocarbonylmethyl]benzoic acid,  G = 4-[(1-(6-Methyl-2-piperidino-phenyl)-ethenyl)-aminocarbonylmethyl]benzoic acid, H = 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid,	55
6	K = ( + )Ethyl 4-[(1-(1-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid-(2,2-dimethyl-dioxolane- L = 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]toluene, M = 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]toluene,	60
6	N = 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzaldehyde, O = 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]phenyl acetic acid, P = 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid, 5 Q = 4-[(1-(4-Chloro-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid,	65

	R = 4-[(1-(5-Chloro-2-piperidino-phenyl)-ethyl)-aminocarbonyl-methyl]benzoic acid,	
	S = Ethyl 4-[(1-(6-Chloro-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoate,	
	T = 4-[(1-(5-Fluoro-2-piperidino-phenyl)-ethyl)-aminocarbonyl-methyl]benzoic acid,	
	U = 4-[(1-(4-Methyl-2-piperidino-phenyl)-ethyl)-aminocarbonyl-methyl]benzoic acid,	_
5	V = 4-[(1-(5-Methyl-2-piperidino-phenyl)-ethyl)-aminocarbonyl-methyl]benzoic acid,	5
	W = 4-[(2-(2-Piperidino-phenyl)-2-propyl)-aminocarbonylmethyl]benzoic acid,	
	X = 4-[(1-(2-Piperidino-phenyl)-2-methyl-propyl)-aminocarbonyl-methyl]benzoic acid,	
	Y = 4 - [(2 - Piperidino - benzhydryl) - aminocarbonylmethyl]benzoic acid,	
	7 = 4 - (1.42 - (1.2.3.6 - Tetrahydro-pyridino) - phenyl) - ethyl) - aminocarbonylmethyl] benzoic acid,	
10	$\Delta \Delta = 4 - (1 - (2 - (3 - Methyl-piperidino) - phenyl) - ethyl) - aminocarbonylmethyl] benzoic acid,$	10
	AR = 4-f(1-12-Hexahydroazepino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid,	
	AC = 4-[(1-(2-Octahydroisoindolo-phenyl)-ethyl)-aminocarbonyl-methyl]benzoic acid,	
	AD = Ethyl 4-[(α-Methoxycarbonyl-2-piperidino-benzyl)-aminocarbonylmethyl]benzoate and	
	AE = (+) 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid.	
15	Ar = ( ) 4 ff ( fr. ) bottom broady sayly	15
13	1. Blood-sugar lowering activity:	
	The blood-sugar lowering activity of the test compounds was determined in home-bred female	
	rats with a weight of 180-220 g. 24 hours before starting the test the animals were starved.	
	Before the test the compounds were suspended in 1.5% methyl cellulose and administered to	
20	the animals by means of an oesophageal tube.	20
20	Blood was taken before administering the test compounds as well as at 1, 2, 3 and 4 hours	20
	after administration from the retroorbital plexus vein. 50 $\mu$ g of each sample were deproteinized	
	with 0.5 ml of 0.33 N perchloric acid and centrifuged. The glucose content in the supernatant	
	was determined according to the Hexokinase method by means of an analysis photometer. The	
	was determined according to the nexokmase method by means of an analysis photometer. The statistical evaluation was performed with the t-test according to Student with $p = 0.05$ .	25
25	statistical evaluation was performed with the t-test according to student with p = 0.05.	23

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The following table contains the obtained values in percent compared with the controls:

Table 1:

	com	25 mg	/kg			10 mg	1/kg 			5 mg/	′kg 			-
	pound	1 hours	2	3	4	1 hours	2	3	4	1 hours	2	3	4	, -
						- 36	<b>– 23</b>	- 14	n.s.	- 22	n.s.	- 10	n.s.	
	A				•				<b>– 13</b>			n.s.	n.s.	
	B C	- 40	- 30	<b>- 26</b>	- 22	- 26	<b>- 17</b>	n.s.	n.s.					
	D	- 40	J <b>U</b> .			- 38	- 36	- 25	- 14	- 27	<b>-</b> 16	- 11		
:	E					<b>- 42</b>	<b>- 39</b>	- 34	- 32	<b>– 45</b>	<b>-41</b>		<b>– 21</b>	
•	F	<u>-</u> 45	- 42	- 38	- 32	- 44	<b>- 39</b>	- 32	- 24		- 33	<b>– 26</b>		
	G									- 31	<b>- 15</b>	n.s.	n.s.	
	H	- 40	- 43	<b>- 45</b>	<del>-</del> 38		- 38			<b>- 45</b>	<b>- 45</b>	- 36		
	ï					- 24	<b>– 27</b>	<del>-</del> 17	- 13	- 22	- 22	n.s. - 31	n.s. - 22	
0	ĸ								. 04	<b>-47</b>	<b>- 42</b>			
_	L					<b>– 39</b>	<b>– 37</b>	- 32	- 24	<b>- 43</b>	- 34	- 29	- 13	
	M	- 45	- 44	- 38	<b>- 32</b>				21	25	- 29	n.s.	n.s.	
	N					<b>- 40</b>		- 30					- 23	
	0	- 46	47	<b>– 37</b>	- 36	<b>- 46</b>	- 41			- 43 - 27			n.s.	
5	P					- 41	<b>- 26</b>		n.s. - 30	- 21	- 10	11.3.	11.3.	
	Q					- 35 - 36	- 38	- 33 - 36	<b>–</b> 30	_ 17	- 18	- 11	n.s.	
	R	<b>– 36</b>	- 36	- 34	- 28	- 30	- 34	- 20	- 20	.,		• •		•
	S	- 44	<b>- 46</b>	- 39	<b>– 37</b>	40	_ 47	_ 46	_ 46	_ 43	- 36	- 29	- 29	
_	T						- 47 - 18				<b>– 15</b>	n.s.	n.s.	
0	U			25	20	-3/	- 10	11.0.	11.5.	-7-				
	V.	<b>– 28</b>	<del>-</del> 23	<b>– 25</b>	<b>– 20</b>	_ 32	_ 34	- 27	- 20	- 19	- 24	<b>–</b> 16	n.s.	
	W	40	- 45	- 43	- 36	_ 43	- 41	<b>– 36</b>	<b>– 28</b>	- 36	- 40	- 32	<b>- 32</b>	
	X	<b>- 46</b>	- 45 - 44°			- 40	-71					<b>-41</b>		
_	Y	<b>- 44</b>	- 44	-41	- 72					<b>- 45</b>		<b>–</b> 35		
Ð	Z	- 46	- 38	<b>– 44</b>	<b>– 46</b>	- 42	- 32	- 26	<b>– 35</b>	- 48	- 36	- 33	<b>– 20</b>	
	AA	- 46 - 45	- 36 - 46	- 39	<b>– 34</b>	- 41	<b>– 35</b>	<b>– 24</b>	<b>— 17</b>	- 29	<b>–</b> 18	n.s	. n.s.	
	AB	- 45 - 41	<b>– 40</b> <b>– 44</b>	<b>– 33</b>	<b>– 26</b>									
	AC AD	- 41				- 40	- 32	- 31	<b>– 17</b>					
^	AE**									<b>- 41</b>	<b>-</b> 34	- 20	n.s.	

<sup>=</sup> dose: 20 mg/kg

Acute toxicity:

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The acute toxicity was determined in home-bred female and male mice with a body weight of 20-26 g after oral administration (suspension in 1% methyl cellulose) of a single dose.

Observation time: 14 days

The following table contains the values obtained:

	Test compound	orientating toxicity
55	H R	>2 000 mg/kg p.o. (1 out of 10 animals died) >2 000 mg/kg p.o. (0 out of 10 animals died) >2 000 mg/kg p.o. (0 out of 6 animals died)

The compounds of general formula I are suitable for the treatment of diabetes mellitus due to their benefical effects on intermediary metabolism and their blood-sugar lowering activity. According to a yet further feature of the present invention there are provided pharmaceutical

compositions comprising as active ingredient at least one compound of general formula I as 65 hereinbefore defined or a physiologically compatible salt thereof, in association with one or more 65

<sup>&</sup>quot;= dose: 1 mg/kg

n.s. = statistically not significant

5	pharmaceutical carriers or excipients.  For pharmaceutical administration, the compounds of general formula I or their physiologically compatible salts may be incorporated into conventional preparations in either solid or liquid form, optionally in combination with other active ingredients. The compositions may, for example, be presented in a form suitable for oral or parenteral administration. Preferred forms include, for example, tablets, coated tablets, capsules, powders or suspensions.  The active ingredient may be incorporated in excipients customarily employed in pharmaceutical compositions such as for example, corn starch, lactose, magnesium stearate, aqueous or non-aqueous vehicles, fatty substances of animal or vegetable origin, paraffin derivatives,								
10	D polyvinyl pyrrolidone, potato starch, various wetting, dispersing or emulsifying agents and/or preservatives.  Advantageously the compositions may be formulated as dosage units, each dosage unit being adapted to supply a fixed dose of active ingredient. Suitable single dosage units for adults contain from 1 to 50 mg, preferably 2.5 to 20 mg of active ingredient according to the								
15	invention. Such dosage units may, for example, be administered 1 or 2 times daily. The total daily dosage may, however, be varied according to the compound used, the subject treated and the complaint concerned.  According to a yet further feature of the present invention there is provided a method of	15							
20	treating a patient suffering from, or susceptible to disorders of intermediary metabolism and/or blood sugar which comprises administering to the said patient an effective amount of a compound of formula I, as hereinbefore defined, or a physiologicaly compatible salt thereof.  The following non-limiting examples serve to illustrate the present invention:	20							
25	Example 1 4-[(1-(5-Chloro-2-dimethylamino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid methyl ester 1.67 g (0.0103 mol) of carbonyl diimidazole were added with stirring at 20°C to a solution of 2.00 g (0.0103 mol) of 4-methoxycarbonyl-phenyl acetic acid in 13.5 ml of absolute tetrahydrofuran. Subsequently the mixture was heated to reflux temperature for 45 minutes excluding moisture. After cooling to room temperature 2.05 g (0.0103 mol) of 1-(5-chloro-2-	25							
30	dimethylamino-phenyl)-ethylamine in 7 ml of absolute tetrahydrofuran were added and the reaction mixture was stirred over night at 20°C. After evaporating in vacuo the evaporation residue was purified by column chromatography on silica gel (toluene/acetone = 10/1).	30							
35	Yield: 2.6 g (66.7% of theory),         M.p.: 153-155°C (from ether).         Calc.:       C 64.08       H 6.18       Cl 9.46       N 7.47         Found:       64.30       6.04       9.70       7.39	35							
40	Analogously to Example 1 the following compounds were prepared: 4-[(1-(5-Chloro-2-dipropylamino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid methyl ester	40							
	Yield: 42% of theory, M.p: 135- 137°C (from ether/petroleum ether)								
45	Calcd.: C 66.83 H 7.25 Cl 8.23 N 6.50 Found: 66.95 7.35 8.35 6.05	45							
50	4-[(1-(5-Chloro-2-dibutylamino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid methyl ester Yield: 64.8% of theory, M.p.: 110–112°C.	50							
	Calc.: 68.03 H 7.69 Cl 7.72 N 6.10 Found: 67.86 7.61 7.73 6.17								
55	4-[(1-(5-Chloro-2-N-cyclohexyl-N-methylamino-phenyl)ethyl)-aminocarbonylmethyl]benzoic acid methyl ester Yield: 63.9% of theory, M.p.: 152-153°C (ether).	55							
60	Calc.: C 67.78 H 7.05 Cl 8.00 N 6.32 Found: 67.70 6.92 8.24 6.46	60							
65	4-[(5-Chloro-2-pyrrolidino-benzyl)-aminocarbonylmethyl]benzoic acid methyl ester Yield: 68.1% of theory, M.p.: 139-141°C (methanol)	65							

17							
	Calc.:	C 65.19	H 5.99	CI 9.17	: N 7.24		
	Found:	65.46	5.91	9.26	7.41		
5	4-[(1-(5-Chloro Yield: 58.3% o M.p.: 133-13	of theory,		)-aminocarbor	nylmethyl]benzoic	acid methyl ester	5
10	Calc.: Found:	C 65.91 66.24	H 6.29 6.19	CI 8.84 8.75	N 6.99 7.13		10
10	4-[(5-Chloro-2- Yield: 75.1% ( M.p.: 123-12	of theory,	nzyl)-aminocal	rbonylmethyl]	benzoic acid meth	yl ester	
15	Calc.: Found:	C 65.91 66.05	H 6.29 6.13	CI 8.84 8.86	N 6.99 7.21		15
20	4-[(1-(5-Chlord Yield: 70.4% ( M.p.: 142-14	of theory,	benzyl)-amino	carbonyl)-eth	/l]benzoic acid me	thyl ester	20
	Calc.: Found:	C 66.57 66.50	H 6.56 6.49	CI 8.55 8.44	N 6.75 6.86		
25	4-[(1-(5-Chlord Yield: 69.5% ( M.p.: 147-14	of theory,	phenyl)-ethyl)	-aminocarbon	ylmethyl]benzoic a	acid methyl ester	25
30	Calc.: Found:	C 66.57 66.33	H 6.56 6.54	CI 8.55 8.67	N 6.75 6.85		30
35	4-[(1-(5-Chlord ester Yield: 54.3% M.p.: 160-16	of theory,		enyl)ethyl)-am	inocarbonylmethy	/]benzoic acid methyl	35
	Calc.: Found:	C 67.20 67.27	N 6.81 6.81	CI 8.27 8.13	N 6.53 6.45		
40	4-[(1-(5-Chlord methyl ester Yield: 44% of M.p.: 190-19	theory,		lino)-phenyl)-e	thyl)-aminocarbor	nylmethyl]benzoic acid	40
45	Calc.: Found:	C 67.78 67.50	H 7.05 7.05	CI 8.00 8.25	N 6.32 6.48		45
50	4-[(1-(5-Chlore Yield: 65.9% M.p.: 142-14	of theory,	phenyl)-propy	/I)-aminocarbo	onylmethyl]benzoid	c acid methyl ester	50
	Calc.: Found:	C 67.20 67.45	H 6.81 6.63	CI 8.26 8.38	N 6.53 6.63		
55	4-[(1-(5-Chlore ester Yield: 61.4% M.p.: 156-15	of theory,	-phenyl)-2-me	thyl-propyl)-a	minocarbonylmeth	yl]benzoic acid methyl	55
60	Calc.: Found:	C 67.78 67.80	H 7.05 7.17	CI 8.00 7.89	N 6.32 6.28		60
65	4-[(1-(5-Chlore Yield: 69.8% M.p.: 156-18	of theory,	o-phenyl)-eth	yl)-aminocarbo	onylmethyl]benzoi	c acid methyl ester	65

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				·			
	Calc.: Found:	C 63.38 63.24	H 6.04 6.12	CI 8.50 8.70	N 6.72 6.85		
5	Yield: 68.2	nloro-2-thiomorph 2% of theory, -169°C (ether).	olino-phenyl)-	ethyl)-aminoc	arbonylmethy	l/benzoic acid methyl ester	5
_	Calc.: Found:	C 61.03 60.83	H 5.82 5.77	CI 8.19 8.33	N 6.47 6.49	S 7.41 7.39	10
	thyl ester Yield: 41.7	<i>lloro-2-(hexahydr</i> '% of theory, –147°C (methyle	•			nylmethyl]benzoic acid me-	
	Calc.: Found:	C 67.19 66.90	H 6.81 6.66	CI 8.27 8.30	N 6.53 6.39		15
		of theory,	azocino-pheny	/l)-ethyl)-amind	ocarbonylmet	hyl]benzoic acid methyl este	r 20
	Calc.: Found:	mol peak m	/e = 442/44 /e = 442/44	4 (1 chlorine) 4 (1 chlorine)			2
	thyl ester Yield: 38%	·		-phenyl)-ethyl)	-aminocarbon	ylmethyl]benzoic acid me-	
	Calc.: Found:	C 68.32 68.10	H 7.28 7.30	N 6.13 6.28			30
		1% of theory,	-phenyl)-2-pro	ppyl)-aminocar	bonylmethyl]	benzoic acid methyl ester	3!
)	Calc.: Found:			0 (1 chlorine) 0 (1 chlorine)			-40
	Yield: 68.3	<i>tro-2-piperidino-p</i> 3% of theory, –180°C (toluene	*	aminocarbony	lmethyl]benze	pic acid methyl ester	7
	Calc.: Found:	C 64.93 65.05	H 6.40 6.43	N 9.88 9.87			4!
)	<i>4-[(1-(2-Pi</i> Yield: 59.1 M.p.: 145	p <i>eridino-phenyl)-</i>  % of theory,  147°C	ethyl)-aminoce	arbonylmethyl	]benzoic acid	methyl ester	·5(
	Calc.: Found:	C 72.61 72.35	H 7.42 7.39	N 7.36 7.40	•		
	Yield: 32.9	y <i>l-2-piperidino-b</i> 3% of theory, –126°C (petrole			]benzoic acid	methyl ester	5!
<b>)</b>	Calc.: Found:		/e = 380 /e = 380				60
	Yield: 62.4	<i>phenacetyl)-N-[1</i> I% of theory, –167°C (ether)	-(2-piperidino-	phenyl)-ethyl]	amin <del>a</del>		

19		_				
	Calc.: Found:	C 68.64 68.73	H 6.86 6.88	N 11.44 11.63		
5	N-(4-Acetyl-pho Yield: 32.4% o M.p.: 162–16	of theory,	-(2-piperidine	o-phenyl)-ethyl	]amine	5
10	Calc.: Found:	C 75.79 75.51	H 7.74 7.86	N 7.69 7.38		10
	N-(4-Acetyl-pho Yield: 50.3% o M.p.: 162-16	of theory,	-(5-chloro-2-	piperidino)phe	nyl)-ethyl]amine	
15	Calc.: Found:	C 69.24 66.88	H 6.82 6.63	N 7.02 6.70		15
20	2-[(1-(2-Piperio Yield: 82% of M.p.: 107-10	theory,	thyl)-aminoc	arbonylmethyľ	benzoic acid methyl ester	20
	Calc.: Found:	C 72.60 72.79	H 7.42 7.38	N 7.36 7.53		
25	3-[1-(2-Piperid Yield: 47% of M.p.: 155°C	<i>ino-phenyĺ)-et</i> theory,	hyl)-aminoca	rbonylmethyl]	benzoic acid ethyl ester	25
30	Calc.: Found:	C 73.07 73.30	H 7.67 7.58	N 7.10 7.17		30
35	3-Chloro-4-[(1- Yield: 63% of M.p.: 123-12	theory,	phenyl)-ethy	l)-aminocarbor	nylmethyl]benzoic acid ethyl ester	35
33	Calc.: Found:	C 67.20 67.28	H 6.81 6.84	CI 8.27 8.36	N 6.53 6.50	
40	4-[(1-(2-(1,2,3 ethyl ester Yield: 43% of M.p.; 142-14	theory,	-isoquinoline	-2-yl)-phenyl}-	ethyl)-aminocarbonylmethyl]benzoic acid	40
45	Calc.: Found:	C 75.99 75.64	H 6.83 6.75	N 6.33 6.35		45
50	4-[(1-(2-Piperio Yield: 59% of M.p.: 136-13	theory,	thyl)-aminoc		]toluene	50
	Calc.: Found:	C 78.53 78.58	H 8.39 8.16	N 8.33 8.26		
55	4-[(5-Chloro-2- Yield: 40.3% ( M.p.: 156-15	of theory,		ylmethyl]benz	oic acid methyl ester	55
60	Calc.: Found:	C 65.19 65.20	H 5.99 6.15	CI 9.16 9.40		60
	4-[2-(2-Piperio Yield: 26.9% M.p.: 71-73°	of theory,		]benzoic acid-	metnyl ester	

20					GB 2 U9U 8 3 4 A	20		
	Calc.:	C 72.10	H 7.15	N 7.65				
	Found:	72.00	7.09	7.94				
5	ester Yield: 63.4%		-pyridino)-phe	enyl)-ethyl)-aı	nino-carbonylmethyl]benzoic acid ethyl	5		
	т.ро	-						
0	Calc.: Found:	C 73.44 73.38	Н 7.19 7.13	N 7.14 7.13		10		
	4-[(2-(5-Chlo Yield: 68% o M.p.: 95-97	of theory,	-phenyl) <del>:</del> ethyl)	)-aminocarbo	nylmethyl]benzoic acid ethyl ester			
5		0.07.00		01.0.07	N C 52	15		
	Calc.: Found:	C 67.20 67.75	H 6.81 6.76	CI 8.27 8.22	N 6.53 6.24			
0	Yield: 47.39		-phenyl)-ethyl)	)-aminocarbo	nylmethyl]benzoic acid ethyl ester	20		
	Calc.:	C 69.88	н 7.99	N 6.79				
	Found:	70.10	7.10	6.87		25		
25	Yield: 56.59	4-[(1-(5-Nitro-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester Yield: 56.5% of theory, M.p.: 144–147°C (ethanol)						
30	Calc.:	C 65.59	Н 6.65	N 9.56		30		
	Found:	65.78	6.56	9.73				
35	4-[(2-(1-(2-P Yield: 90% ( M.p.: 129-	of theory,	d)-ethyl)-amind	ocarbonyl)-et	hyl]benzoic acid methyl ester	35		
	Calc.: Found:	C 73.06 72.61	H 7.67 7.77	N 7.10 7.52				
10	Yield: 44.49	<i>ty-1-(2-piperidir</i> 6 of theory, 135°C (petrolet			bonylmethyl]benzoic acid ethyl ester	40		
45	Calc.: Found:	C 70.22 70.02	H 7.37 7.25	N 6.82 6.77	m/e = 410 m/e = 410	48		
	Yield: 64.29 M.p.: 150-		no-phenyl)-eth	yl)-aminocari	bonylmethyl]benzoic acid ethyl ester	F		
50	Calc.: Found:	C 70.22 70.37	Н 7.37 7.17	N 6.82 6.81	m/e = 410 m/e = 410	50		
					·			
55	Yield: 59%				onylmethyl]benzoic acid ethyl ester	58		
•-	Calc.: Found:	C 68.47 68.57	H 6.90 6.64	N 6.39 6.46	m/e = 438 m/e = 438	e,		
60	4-[(1-(5-Chloester Yield: 71.39 M.p.: <20	% of theory,	-piperidino)-pl	henyl)-ethyl)-	aminocarbonylmethyl]benzoic acid ethyl	60		

```
m/e = 442/444 (1 chlorine)
   Calc.:
               m/e = 442/444 (1 chlorine)
   Found:
    4-[(1-(2-Hexahydroazepino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester
                                                                                                      5
   Yield: 68% of theory,
   M.p.: 145-148°C (toluene)
                                  H 7.90
                                               N 6.86
                   C 73.50
   Calc.:
                                                  6.89
                      73.35
                                    8.04
   Found:
                                                                                                     10
10
    4-[(1-(2-[1,4-Dioxa-8-azaspiro[4,5]decyl-(8)]phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid
    ethyl ester
    Yield: 64.3% of theory,
    M.p.: 143-145°C (petroleum ether/acetone)
                                                                                                     15
15
                   C 69.01
                                  H 7.13
                                               N 6.19
    Calc.:
                      69.30
                                    7.38
                                                  6.21
    Found:
    4-[(1-(2-(2-Methyl-pyrrolidino)-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester
                                                                                                     20
20 Yield: 72% of theory,
    M.p.: 94-97°C
                                  H 7.66
                    C 73.07
                                                N 7.10
    Calc.:
                                                  7.11
                      72.25
                                    7.67
    Found:
                                                                                                     25
    4-[(1-(3-Methyl-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester
    Yield: 39,5% of theory), m.p. 178-179°C
    Calc.: m/e = 408
    Found: m/e = 408
                                                                                                     30
30
    4-[(1-(3-Chloro-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester
    Yield: 52,6% of theory,
    Calc.: m/e = 428/430 (1 chlorine)
    Found: m/e = 428/430 (1 chlorine)
                                                                                                     35
35
    Example 2
    (+) 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester
      231.4 mg (1.43 m mol) of carbonyl diimidazole were added to a solution of 290.9 mg (1.40
    m mol) of 4-ethoxycarbonylphenyl acetic acid in 6 ml of tetrahydrofuran. Subsequently the
                                                                                                     40
40 mixture was heated to reflux temperature for 1.5 hours excluding moisture. After cooling to
    room temperature 0.385 ml ( = 2.78 m mol) of triethylamine (dried over potassium hydroxide)
    and 360 mg (1.30 m mol) of (+) 1-(2-piperidino-phenyl)-ethylamine dihydrochloride [m.p.
    242°C (decomp.); [\alpha]_{D}^{20} = +14.8^{\circ} (c = 1; methanol)] together with 2 ml of tetrahydrofuran
    were added and the mixture was stirred for 4 hours at 50°C in an oil bath. After evaporating in
                                                                                                      45
45 vacuo the evaporation residue was distributed between chloroform and water. The chloroform
    extract was dried over sodium sulfate, filtered through a G3-glas frit and evaporated in vacuo to
    dryness. The obtained residue was purified by column chromatography on silica gel (chloro-
    form/methanol = 6:1).
                                                                                                      50
50 Yield: 229 mg (44.7% of theory),
    M.p.: 89-90°C (ether)
    [\alpha]_{0}^{20} = 8.2^{\circ}C (c = 1; methanol)
                                                           m/e = 394
                    C 73.07
                                  H 7.66
                                                N 7.10
    Calc.:
                                                                                                      55
                                                           m/e = 394
                      73.20
                                     7.68
                                                  7.14
55 Found:
      Analogously to Example 2 was prepared:
    ( - ) 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester
60 from (-) 1-(2-piperidino-phenyl)-ethylamino dihydrochloride [m.p.: 239-242°C (decomp.);
                                                                                                      60
    [\alpha]_{D}^{20} - 19.6° (c = 1; methanol)].
    Yield: 41.1% of theory,
    M.p.: 77-79°C (ether/cyclohexane)
    [\alpha]^{20} = -6.2^{\circ} (c = 1; methanol)
```

	Calc.: Found:	C 73.07 72.67	H 7.66 7.75	N 7.10 6.82	m/e = 394 m/e = 394	
5	2.3 ml (0.02 1-(4-chloro-2-pi	23 mol) of car iperidinophen 028 mol) of to	bon tetrachlo yl)-ethylamine iphenyl phos	ride were add e, 4.8 g (0.02 phine and 3.2	ylmethyl]benzoic acid ethy ed to a solution of 5.5 g ( 23 mol) of 4-ethoxycarbon 2 ml (0.023 mol) of triethy urs at room temperature.	0.023 mol) of ylphenyl acetic ylamine in 50
10	evaporating in	vacuo the eva The combined ated in vacuo	poration residual portain organic extra and the evar	due was distri icts, which wo ooration resid	buted between 100 ml of ere dried over sodium sulfa ue was purified by column	water and 10 ate. were
15	Yield: 6.1 g (6: M.p.: 126-12		,			15
•	Calc.: Found:	C 76.20 67.43	H 6.81 6.97	CI 8.27 8.16	N 6.53 6.40	20
20	Analogously	to Example 3	the following	compounds	were prepared:	20
25	4-[(1-(4-Methy) Yield: 48.2% c M.p.: 120-12	of theory,	phenyl)-ethyl	)-aminocarbor	nylmethyl]benzoic acid eth	yl ester 25
	Calc.: Found:	C 73.50 73.61	H 7.89 7.95	N 6.86 6.73		
30	4-[(1-(2-(4-Met Yield: 55.8% o M.p.: 125-12	of theory,	)-phenyl)-ethy	/l)-aminocarbo	onylmethyl]benzoic acid et	hyl ester 30
35	Calc.: Found:	C 73.50 73.30	H 7.90 7.99	N 6.86 7.20		35
40	4-[(1-(2-Piperio Yield: 71% of M.p.: 147-14	theory,	thyl)-aminoca	rbonylmethylj	benzoic acid ethyl ester	40
40	Calc.: Found:	C 73.06 73.54	H 7.67 8.04	N 7.10 6.95		40
45	4-[1-(2-Piperid Prepared from Yield: 27% of M.p.: 186-18	m 1-(2-piperio theory,	<i>hyl)-aminocar</i> lino-phenyl)-e	<i>bonylmethyl]</i> thylamine and	ohenyl acetic acid d p-phenylene diacetic acid	i. 45
50	Calc.: Found:	C 72.60 72.75	H 7.42 7.65	N 7.36 7.11		50
56	4-[(2-Piperiding Yield: 87.4% o M.p.: 160–16	of theory,	aminocarbon	ylmethyl]-ben	zoic acid ethyl ester	55
55	Calc.: Found:	C 76.29 76.44	H 7.06 7.08	N 6.14 6.17		33
60	4-[(5-Chloro-2- Yield: 78% of M.p.: 202-20	theory,	nzhydryl)-amii	nocarbonylme	thyl]benzoic acid ethyl est	er 60
	Calc.: Found:	C 70.93 70.85	H 6.36 6.40	CI 7.22 7.11	N 5.71 5.45	

23						
	<i>4-[(1-(4-Piperio</i> Yield: 39% of M.p.: 118–12	theory,	ethyl)-amino	carbonylmet	hyl]benzoic acid ethyl ester	·
.5	Calc.: Found:	C 73.07 73.20	Н 7.67 7.78	N 7.10 7.11		5
10	<i>4-[(1-(2-(4-Me</i> : Yield: 53% of M.p.: 130–13	theory,	no)-phenyl)-e	thyl)-aminoc	arbonylmethyl]benzoic acid ethyl este	er . 10
	Calc.: Found:	C 70.38 70.41	H 7.63 7.53	N 10.26 10.13		
15	<i>4-[(1-(2-(4-Ber</i> Yield: 75% of M.p.: 135–13	theory	o)-phenyl)-e	thyl)-aminoca	arbonylmethyl]benzoic acid ethyl este	or 15
20	Calc.: Found:	C 74.20 74.45	H 7.26 7.34	N 8.66 8.54		20
	4-[(1-(2-(4-p-c	hlorophenyl- <sub>l</sub>	piperazino)-p	henyl)-ethyl)	-aminocarbonylmethyl]benzoic acid e	othyl
25	ester Yield: 48.5% M.p.: 178–18	of theory,				25
	Calc.: Found:	C 68.83 68.71	H 6.37 6.22	N 8.30 8.41	CI 7.01 6.82	
30	4-[(α-Cyclohex Yield: 75% of M.p.: 135°C	y <i>l-2-piperidii</i> theory,	no-benzyi)-ai	minocarbony	Imethyl]benzoic acid ethyl ester	30
35	Calc.: Found:	C 75.29 75.11	H 8.28 8,13	N 6.06 5,99		35
	N-(4-Chloro-pl Yield: 79% of M.p.: 150-1	theory,	[1-(2-piperio	lino-phenyl)-	ethyl]amine	40
40	Calc.: Found:	C 70.67 70.94	H 7.06 7.84	CI 9.93 10.09	N 7.85 7.90	
45	4-[(2-Pyrrolidi Yield: 57% of M.p.: 163-10	theory,	yl)-aminocar	bonylmethyi	benzoic acid ethyl ester  -	45
50	Calc.: Found:	C 75.99 75.45	H 6.83 6.52	N 6.33 6.10		50
50	4-[(2-Hexame Yield: 68% of M.p.: 151-1	f theory,	-benzhydryl	)-aminocarbo	nylmethyl]benzoic acid ethyl ester	
55	Calc.: Found:	C 76.56 76.43	H 7.28 7.19	N 5.95 6.01		55
60	11.2 g (0.0 phosphine, 2 tetrachloride	0539 mol) of 2.6 ml (0.16 were success	4-ethoxyca 2 mol) of tr ively added	rbonyl-pheny iethylamine i with stirring	nethyl]-benzoic acid ethyl ester vlacetic acid, 17 g (0.0647 mol) of to and 5.2 ml (0.0539 mol) of carbon to a solution of 10.9 g (0.0539 mol) ine in 100 ml of acetonitrile. The sol	l) of lution,
65					20 hours at 20°C. The resultant preche filtrate was evaporated in vacuo.	

		due was puri	fied by colum	nn chromatography on silica gel (toluene/ace-	
	tone = $10:1$ ).			•	
5	M.p.: 112-115	6°C (ether)			5
	Calc.: Found:	C 73.44 73.28	H 7.19 7.32	N 7.14 6.96	
10	Analogously t	to Example 4	the following	g compounds were prepared:	10
15	Found: 73.28 7.32 6.96  Analogously to Example 4 the following compounds were prepared:  4.((-C/cyclohaxy/idene-2-piperidino-benzyl)-aminocarbonylmethyl]benzoic acid ethyl ester Yield: 24% of theory, M.p.: 131-133°C  Calc.: C 75.62 H 7.88 N 6.08 Found: 75.59 7.47 6.01  4.((1-(2-Piperidino-phenyl)-propenyl)-aminocarbonylmethyl]benzoic acid ethyl ester Yield: 85.0% of theory (E- and Z-isomeric mixture) M.p.: of the polar isomer: 82-84°C  Calc.: C 73.85 H 7.44 N 6.89 Found: 73.73 7.57 7.01  Example 5 4.((1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester A solution of 60.6 g (0.267 mol) of 4-ethoxycarbonyl-phenacetyl chloride in 120 ml of methylene chloride was dropped with slight ice cooling to a stirred solution of 49.6 g (0.243 mol) of 1-(2-piperidino-phenyl-ethylamine [b.p. 0.6: 100-107°C: mp. of the dihydrochloride: 234-237°C (decomp.)] and 37.3 ml (0.267 mol) of triethylamine in 245 ml of methylene chloride phase were stracted successively twice with water, once with 35 10% aqueous ammonia, twice with water, once with 100 ml of 3% hydrochloric acid and twice with water. The methylene chloride phases were stracted successively twice with water, once with 100 ml of 3% hydrochloric acid and twice with water. The methylene chloride phases was dried over sodium sulfate and evaporated in vecuo. The evaporation residue was crystallized from ether. Yield: 88.8 g (92.7% of theory), M.p.: 148-150°C  4.((-6.Methyl-2-piperidino-benzyl)-aminocarbonylmethyl]benzoic acid ethyl ester Yield: 22.5% of theory, M.p.: 173.49 7.60 N 7.10  Found: 73.48 7.62 7.15  4.((-1.(-6.Methyl-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester Yield: 25.5% of theory, M.p.: 131-132°C (ethanol)	15			
20	Yield: 65,0% o	f theory (E- a	nd Z-isomeric	nocarbonylmethyl]benzoic acid ethyl ester c mixture)	20
25					25
30	Example 5 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester A solution of 60.6 g (0.267 mol) of 4-ethoxycarbonyl-phenacetyl chloride in 120 ml of methylene chloride was dropped with slight ice cooling to a stirred solution of 49.6 g (0.243 mol) of 1-(2-piperidinophenyl)-ethylamine [b.p. 0.6: 100-107°C; m.p. of the dihydrochloride: 234-237°C (decomp.)] and 37.3 ml (0.267 mol) of triethylamine in 245 ml of methylene chloride at an internal temperature of 20-30°C. After stirring for 2 hours at room temperature, the resultant precipitate was filtered off, washed once with methylene chloride, and the combined methylene chloride phases were extracted successively twice with water, once with 10% aqueous ammonia, twice with water, once with 100 ml of 3% hydrochloric acid and twice with water. The methylene chloride phase was dried over sodium sulfate and evaporated in vacuo. The evaporation residue was crystallized from ether.				
40	Analogously '	to Example 5			40
45	Yield: 22.5% o M.p.: 116.5–1 Calc.:	of theory, 17°C (ethand C 73.07	ol/petroleum H 7.66	ether) Ň 7.10	<b>45</b> .
50	Yield: 20.2% o	f theory,		l)-aminocarbonylmethyl]benzoic acid ethyl ester	50
==	Found:				55
ອອ	4-[(1-(5-Metho: Yield: 35.8% c	of theory,		nyl)-aminocarbonylmethyl]benzoic acid ethyl ester	
60			H 7.60 7.59		60

25							
•	<i>4-[(1-(2-Piperid</i> Yield: 65.2% o M.p.: <20°C	lino-phenyl)-et of theory,	hyl)-N-methyla	amino-carbon	ylmethyl]benzo	nic acid ethyl ester	
_	Calc.: Found:	C 73.50 72.99	H 7.90 7.60	N 6.86 6.87			5
	<i>A_[(1_(2-Decah)</i>	vdro-isoquino	line-2-yl)-phen	yl)-ethyl)-amiı	nocarbonylmet	hyl]benzoic acid ethyl ester	
	Yield: 44% of M.p.: 159°C	theory,					10
	Calc.: Found:	C 74.96 75.09	Н 8.08 8.01	N 6.24 6.01			
15	4-[(1-(2-(1,2,3 zoic acid ethyl Yield: 35% of M.p.: 115-11	ester theory,	ctahydro-isoqu	inoline-2-yl)-p	ohenyi)-ethyi)-a	minocarbonylmethyl]ben-	15
20	Calc.: Found:	C 75.30 75.18	н 7.67 7.37	N 6.27 5.89			20
25	4-[(1-(2-Octah Yield: 36% of M.p.: 141°C	ydro-isoindole theory,	-2-yi)-phenyi)-	ethyl)-aminoc	arbonylmethyl <u></u>	]benzoic acid ethyl ester	25
	Calc.: Found:	C 74.62 74.70	H 7.88 7.97	N 6.44 6.42			
30	4-[(1-(3-Piperi Yield: 24% of M.p.: 164°C	idino-phenyl)-( theory,	ethyl)-aminoca	rbonylmethyl <u></u>	]benzoic acid e	thyl ester	30
35	Calc.: Found:	C 73.07 72.80	H 7.66 7.48	N 7.10 7.13			35
	4-[(1-(6-Chlor Yield: 17% of M.p.: <20°C	f theory,	-phenyl)-ethyl	)-aminocarboi	nylmethyl]benz	oic acid ethyl ester	40
40	Calc.: Found:	C 67.20 67.96	H 6.81 6.56	CI 8.26 8.80	N 6.53 6.67	m/e = 428/30 m/e = 428/30	
45	4-[(1-(6-Meth Yield: 3.5% o M.p.: <20°0	of theory,	o-phenyl)-ethy	I)-aminocarbo	nylmethyl]ben	zoic acid ethyl ester	45
	Calc.: Found:	C 73.49 73.80	H 7.89 7.61	N 6.85 7.01	m/e = 408 m/e = 408		50
50	ester		.2]nonane-3-y	l)-phenyl)-eth	yl)-aminocarbo	nyImethyl]benzoic acid ethy	/l
55			m/e = 434				5
	Calc.: Found:		m/e = 434	17 A1 _b	domina		
60	1 Yield: 53.5%	ro-2-piperiding of theory, 36°C (ethand	o-phenyl)-ethyl ol)	ij-N-phenacet	yiamin <del>o</del>		6
	Calc.: Found:	C 70.67 70.40	H 7.06 7.32	CI 9.94 9.77	N 7.85 7.68		

	•	
5	Example 6 4-[(1-(2-Piperidino-phenyl)-ethenyl)-aminocarbonylmethyl]benzoic acid ethyl ester A solution of 2.49 g (0.011 mol) of 4-ethoxycarbonylphenacetyl chloride in 10 ml of methylene chloride was added with ice cooling over 15 minutes to a stirred solution of 2.02 g (0.010 mol) of freshly prepared methyl-(2-piperidinophenyl)-ketimine and 1.53 ml of (0.011 mol) of triethylamine in 10 ml of methylene chloride at an internal temperature of 1 to 6°C. The reaction mixture was stirred for 20 minutes at 20°C and poured into cold sodium hydrogen	5
10	carbonate solution. After extracting several times the organic extract was washed once with water, dried over sodium sulfate, filtered, and evaporated in vacuo. The evaporation residue was purified by column chromatography on silica gel (toluene/acetone = 50:1). Yield: 1.86 g (47.7% of theory), M.p.: 113-116°C (ethanol)	10
15	Calc.: C 73.44 H 7.19 N 7.14 m/e = 392 Found: 72.95 6.98 7.22 m/e = 392	15
	Analogously to Example 6 the following compounds were prepared:	
20	4-[(1-(6-Chloro-2-piperidino-phenyl)-ethenyl)-aminocarbonylmethyl]benzoic acid ethyl ester Yield: 37% of theory, M.p.: 102–105°C	20
25	Calc.: C 67.51 H 6.37 Cl 8.30 N 6.56 $m/e = 426/28$ Found: 67.86 6.39 8.58 6.23 $m/e = 426/28$	25
25	4-[(1-(6-Methyl-2-piperidino-phenyl)-ethenyl)-aminocarbonylmethyl]benzoic acid ethyl ester Yield: 41% of theory, M.p.: 116-118°C	23
30	Celc.: C 73.86 H 7.43 N 6.89 Found: 73.75 7.43 6.77	30
35	Example 7 4-[(1-(2-Piperidino-phenyl)-ethenyl)-aminocarbonylmethyl]benzoic acid ethyl ester A solution of 1.55 g (6.86 m mol) of 4-ethoxycarbonylphenacetyl chloride in 5 ml of methylene chloride was added with stirring to a suspension of 2.20 g (6.24 m mol) of magnesium iodide-[methyl-(2-piperidino-phenyl)-ketimino]-complex in 15 ml of methylene chlo- ride, whereby the internal temperature rose from 20 to 30°C. After stirring for 2 hours at room	35
40	temperature, the reaction mixture was mixed with water whilst stirring and extracted several times with methylene chloride. The methylene chloride solution was washed thrice with water, dried over sodium sulfate, filtered and evaporated in vacuo. The evaporation residue was purified by column chromatography on silica gel (toluene/acetone = 50:2). Yield: 1.1 g (45.8% of theory), M.p.: 115–118°C (ethanol)	40
45	Calc.: C 73.44 H 7.19 N 7.14 Found: 73.30 7.06 7.16	45
	Analogously to Example 7 the following compound was prepared:	
50	4-[(1-(5-Chloro-2-piperidino-phenyl)-ethenyl)-aminocarbonylmethyl]benzoic acid ethyl ester Yield: 39.5% of theory, M.p.: 142–145°C (ethanol)	50
55	Calc.: C 67.51 H 6.37 Cl 8.30 N 6.56 Found: 67.51 6.37 8.36 6.49	55
60	bonylmethyl]benzoic acid-methyl ester and 0.32 g (0.00801 mol) of sodium hydroxide in 23 ml of ethanol and 7 ml of water was stirred for 2 hours at 50°C. After evaporating in vacuo, water was added and the reaction mixture was adjusted to pH 6 by means of 2 N-hydrochloric acid	60
65	and extracted with ethyl acetate. The organic phase was extracted with water, dried over sodium sulfate, filtered and evaporated in vacuo. The evaporation residue was recrystallized from ether.	65

	·						
5	Example 6 4-[(1-(2-Piperidino-phenyl)-ethenyl)-aminocarbonylmethyl]benzoic acid ethyl ester A solution of 2.49 g (0.011 mol) of 4-ethoxycarbonylphenacetyl chloride in 10 ml of methylene chloride was added with ice cooling over 15 minutes to a stirred solution of 2.02 g (0.010 mol) of freshly prepared methyl-(2-piperidinophenyl)-ketimine and 1.53 ml of (0.011 mol) of triethylamine in 10 ml of methylene chloride at an internal temperature of 1 to 6°C. The reaction mixture was stirred for 20 minutes at 20°C and poured into cold sodium hydrogen carbonate solution. After extracting several times the organic extract was washed once with	5					
10	water, dried over sodium sulfate, filtered, and evaporated in vacuo. The evaporation residue was purified by column chromatography on silica gel (toluene/acetone = 50:1). Yield: 1.86 g (47.7% of theory), M.p.: 113-116°C (ethanol)	10					
15	Calc.: C 73.44 H 7.19 N 7.14 $m/e = 392$ Found: 72.95 6.98 7.22 $m/e = 392$	15					
	Analogously to Example 6 the following compounds were prepared:						
20	4-[(1-(6-Chloro-2-piperidino-phenyl)-ethenyl)-aminocarbonylmethyl]benzoic acid ethyl ester Yield: 37% of theory, M.p.: 102–105°C	20					
25	Calc.: C 67.51 H 6.37 Cl 8.30 N 6.56 $m/e = 426/28$ Found: 67.86 6.39 8.58 6.23 $m/e = 426/28$	25					
20	4-[(1-(6-Methyl-2-piperidino-phenyl)-ethenyl)-aminocarbonylmethyl]benzoic acid ethyl ester Yield: 41% of theory, M.p.: 116-118°C						
30	Calc.: C 73.86 H 7.43 N 6.89 Found: 73.75 7.43 6.77	30					
35	Example 7 4-[(1-(2-Piperidino-phenyl)-ethenyl)-aminocarbonylmethyl]benzoic acid ethyl ester A solution of 1.55 g (6.86 m mol) of 4-ethoxycarbonylphenacetyl chloride in 5 ml of methylene chloride was added with stirring to a suspension of 2.20 g (6.24 m mol) of magnesium iodide-[methyl-(2-piperidino-phenyl)-ketimino]-complex in 15 ml of methylene chlo-						
40	ride, whereby the internal temperature rose from 20 to 30°C. After stirring for 2 hours at room temperature, the reaction mixture was mixed with water whilst stirring and extracted several times with methylene chloride. The methylene chloride solution was washed thrice with water, dried over sodium sulfate, filtered and evaporated <i>in vacuo</i> . The evaporation residue was purified by column chromatography on silica gel (toluene/acetone = 50:2). Yield: 1.1 g (45.8% of theory),	40					
45	M.p.: 115–118°C (ethanol)  Calc.: C 73.44 H 7.19 N 7.14  Found: 73.30 7.06 7.16	45					
50	Analogously to Example 7 the following compound was prepared:	50					
30	4-[(1-(5-Chloro-2-piperidino-phenyl)-ethenyl)-aminocarbonylmethyl]benzoic acid ethyl ester Yield: 39.5% of theory, M.p.: 142–145°C (ethanol)	50					
55	Calc.: C 67.51 H 6.37 Cl 8.30 N 6.56 Found: 67.51 6.37 8.36 6.49	55					
60	bonylmethyl]benzoic acid-methyl ester and 0.32 g (0.00801 mol) of sodium hydroxide in 23 ml of ethanol and 7 ml of water was stirred for 2 hours at 50°C. After evaporating in vacuo, water was added and the reaction mixture was adjusted to pH 6 by means of 2 N-hydrochloric acid	60					
65	and extracted with ethyl acetate. The organic phase was extracted with water, dried over sodium sulfate, filtered and evaporated in vacuo. The evaporation residue was recrystallized from ether.	65					

27						GB 2 090 834A	<u> 27</u>
	Yield: 1.7 g ( M.p.: 190-1	88% of theory 92°C			· .		
5	Calc.: Found:	C 63.24 62.90	H 5.87 5.81	CI 9.83 10.02	N 7.76 7.90		5
	Analogous	ly to Example 8	the followin	g compounds	were prepared:		
10	4-[(1-(5-Chlo Yield: 87.6% M.p.: 203-2	of theory,	nino-phenyl)-«	ethyl)-aminoca	rbonylmethyl]benzoic	acid .	10
15	Calc.: Found:	C 66.25 65.97	H 7.01 6.96	CI 8.50 8.52	N 6.72 6.55		15
15	4-[(1-(5-Chlo Yield: 77.3% M.p.: 200-2	of theory,	ino-phenyl)-et	hyl)-aminocar	bonylmethyl]benzoic a	cid	
20	Calc.: Found:	C 67.47 67.45	H 7.48 7.60	CI 7.97 8.28	N 6.30 6.44		20
25	Yield: 88.2%	oro-2-N-cyclohe: 5 of theory, 200°C (ether).	xyl-N-methyla	mino-phenyl)-	ethyl)-aminocarbonyln	nethyl]benzoic acid	25
	Calc.: Found:	C 67.20 67.10	H 6.81 6.73	CI 8.27 8.16	N 6.53 .6.47		
30	Yield: 84.2%	<i>2-pyrrolidino-b</i> 6 of theory, 210°C (ethyl ac		arbonylmethy	l]benzoic acid		30
35	Calc.: Found:	C 64.42 64.70	H 5.68 5.68	CI 9.51 9.58	N 7.51 7.60		35
40	Yield: 81.19 M.p.: 202-2	oro-2-pyrroliding 6 of theory, 204°C (ethyl ac		/l)-aminocarbo	nylmethy∏benzoic aci	1	40
70	Calc.: Found:	C 65.20 65.02	H 5.99 6.12	CI 9.17 9.32	N 7.24 7.10		
45	4-[(5-Chloro- Yield: 78% o M.p.: 164-		nzyl)-aminoc	arbonylmethyl	]benzoic acid		45
50	Calc.: Found:	C 65.19 65.50	H 5.99 5.76	CI 9.17 9.24	N 7.24 7.36		50
	Yield: 81.19	oro-2-piperidino 6 of theory, 216°C (acetone		ocarbonyl)-eti	nyl]benzoic acid		
55	Calc.: Found:	C 65.90 66.30	H 6.29 6.40	CI 8.84 9.00	N 6.99 7.04		55
60	Yield: 84.99	oro-2-piperidino % of theory, 215°C (ether)	-phenyl)-ethy	l)-aminocarbo	nylmethyl]benzoic acid	i	60
	Calc.: Found:	C 65.91 66.18	H 6.29 6.19	CI 8.85 8.88	N 6.99 7.12		

	4-[(1-(5-Chlore Yield: 69.2% M.p.: 208-21	of theory,		nenyl)-ethyl)-aı	minocarbonylmeti	hy∏benzoic acid			
5	Calc.: Found:	C 66.57 66.36	H 6.56 6.77	CI 8.55 8.58	N 6.75 6.80	•	5		
10	4-[(1-(5-Chlore Yield: 82.2% M.p.: 212-21	of theory,	methyl-piperio	dino)-phenyl)-(	athyl)-aminocarbo	nylmethyl]benzoic acid	10		
	Calc.: Found:	C 67.20 66.95	H 6.81 6.69	CI 8.26 8.43	N 6.53 6.68		•		
15	4-[(1-(5-Chlore Yield: 81.5% M.p.: 200-20	of theory,	-phenyl)-propy	yl)-aminocarbo	onyimethyi]benzo	ic acid	15		
20	Calc.: Found:	C 66.57 66.74	H 6.56 6.35	CI 8.55 8.59	N 6.75 6.45		20		
0.5	4-[(1-(5-Chloro-2-piperidino-phenyl)-2-methyl-propyl)-aminocarbonylmethyl]benzoic acid Yield: 82.7% of theory, M.p.: 236-240°C (ethyl acetate)								
25	Calc.: Found:	C 67.20 67.19	H 6.81 6.56	CI 8.27 8.14	N 6.53 6.39	•	25		
30	4-[(1-(5-Chlord Yield: 85.6% M.p.: 201-20	of theory,	o-phenyl)-eth	yl)-aminocarbo	onylmethyl]benzol	ic acid	30		
35	Calc.: Found:	C 62.60 62.30	H 5.75 5.82	CI 8.80 8.83	N 6.95 6.85		25		
39	4-[(1-(5-Chlord Yield: 87.6% M.p.: 216-21	of theory,	olino-phenyl)-	ethyl)-aminoc	arbonylmethyl]be	nzoic acid	35		
40	Calc.: Found:	C 60.20 59.90	H 5.53 5.51	CI 8.46 8.61	N 6.69 6.53		40		
45	4-[(1-(5-Chlord Yield: 81.2% M.p.: 202-20	of theory,		-phenyl)-ethyl	)-aminocarbonyIm	nethyl]benzoic acid	45		
	Calc.: Found:	C 66.58 66.60	H 6.56 6.37	CI 8.55 8.50	N 6.75 6.59				
50	4-[(1-(5-Chlord Yield: 44.4% M.p.: 195-19	of theory,			ocarbonylmethyl]. •	benzoic acid	50		
55	Calc.: Found:	C 67.19 67.10	H 6.81 6.97	N 6.53 - 6.37			55		
	4-[(1-(5-Chlord Yield: 74.7% M.p.: 204-20	of theory,		•	-aminocarbonylm	ethyl]benzoic acid			
60	Calc.: Found:	C 67.78 67.50	H 7.05 7.03	N 6.32 6.04			60		

29								
	Yield: 82.9	nloro-2-piperidino- 9% of theory, 229°C (acetone)	ohenyl)-2-pro	pyl)-aminocar	bonylmethyl]bei	nzoic acid		
	Calc.: Found:	C 66.57 66.03	H 6.56 6.66	CI 8.55 8.67	N 6.75 6.59		5	
10	Yield: 95.6	itro-2-piperidino-pl 3% of theory, –254°C (ether)	nenyl)-ethyl)-	aminocarbony	lmethyl]benzoic	acid	· 10	
	Calc.: Found:	C 64.22 64.20	H 6.12 6.17	N 10.21 10.12				
15	<i>4-[(1-(2-Pi</i> Yield: 85% M.p.: 170	peridino-phenyl)-e 6 of theory, 1–172°C	thyl)-aminoca	arbonylmethyl	]benzoic acid		15	
20	Calc.: Found:	C 72.11 71.94	H 7.15 7.03	N 7.64 7.72	-		20	
25	Yield: 72.	<i>4-[(2-(2-Piperidino-phenyl)-2-propyl)-aminocarbonylmethyl]benzoic acid</i> Yield: 72.7% of theory, M.p.: 213–215°C						
25	Calc.: Found:	C 72.61 72.52	H 7.42 7.31	N 7.36 7.45				
30	4-[(5-Meti Yield: 64. M.p.: 120	h <i>yl-2-piperidino-be</i> 6% of theory <i>,</i> )–122°C	enzyl)-aminod	arbonylmethy	l]benzoic acid		30	
35	Calc.: Found:	C 72.11 72.42	H 7.15 7.38	N 7.64 7.45	m/e = 366 m/e = 366		35	
-	M.p. of	the hydrochloride	: 266°C (ded	comp.)				
40	Calc.: Found:	C 65.58 65.00	6.76 6.62	8.80 9.40	N 6.95 7.00		40	
40	4-[(2-Pipe	oridino-anilino)-cari 5% of theory, 3–217°C	bonylmethy[]	benzoic acid >	< 0.25 HCl			
45	Calc.: Found:	(X 0.25 HCI)C 6 69.40	9.11	H 6.45 6.32	CI 2.55 3.08	N 8.06 8.37	48	
50	Yield: 51.	oro-2-piperidino-an .3% of theory, 2°C (decomp.)	nilino)-carbon	ylmethyl]benz	oic acid hydroc	hloride	50	
	Calc.: Found:	C 58.68 58.26	H 5.42 5.44	CI 17.32 17.97	N 6.84 6.74			
55	Yield: 69	i <i>peridino-anilino-ca</i> .9% of theory, 1–153°C (petrolet			semihydrate		5	
60	Calc.: Found:	(× 0.5 H <sub>2</sub> O)		C 69.78 69.30	H 6.97 6.82	N 7.75 7.46	6	
	Yield: 71	?-Piperidino-pheny .4% of theory, 1–172°C (acetone			hyl]benzoic acid	× 0.2 H₂O		

						•	
	Calc.: Found:	(× 0.2 H₂O)		C 71.91 71.90	H 7.45 7.30	N 7.29 7.03	
5	244 mg	<i>nzyloxy-2-piperid</i> (0.487 m mol) o	f 4-[(1-(5-ben	zyloxy-2-pipe	ridino-phenyl)-e	enzoic acid ethyl)-aminocarbonylmeth ng with 0.73 ml of 1N	5 y-
	sodium hy the thinlay mixture wa extract was was recrys Yield: 191	droxide solution in er chromatogram as evaporated in s dried over sodiu tallized from meth mg (83% of the	n a bath of 50 . After addition Pacuo and dis m sulfate, filt nanol.	0°C, until (af on of 0.73 m tributed betw	ter 3 hours) no I of 1N hydroch reen ethyl aceta	ester could be detected in loric acid, the reaction te and water. The organic to. The evaporation residu	10
15	M.p.: 220	-222°C					15
	Calc.: Found:	C 73.71 73.21	H 6.83 6.67	N 5.93 5.80			
20	Analogo	usly to Example 9	the following	g compounds	were prepared	:	20
0.5	Yield: 68.	xahydroazepino-p 5% of theory, –176°C (ethyl ac		aminocarbon	ylmethyl]benzoi	c acid	0.5
25	Calc.: Found:	C 72.61 72.36	H 7.42 7.34	N 7.36 7.38			25
30	Yield: 68.2	<i>,2,3,6-Tetrahydro</i> 2% of theory, –160°C (ethyl ac		enyl)-ethyl)-aı	minocarbonylme	ethyl]benzoic acid	30
35	Calc.: Found:	C 72.51 72.20	H 6.64 6.66	N 7.69 7.74			35
40	Yield: 75%	nloro-2-piperidino- 6 of theory, –195°C (ethyl ac		)-aminocarbo	nylmethyl]benze	oic acid	40
	Calc.: Found:	C 65.91 66.39	H 6.29 6.17	CI 8.84 8.45	N 6.99 6.78		
45	Yield: 52.	<i>uoro-2-piperidino-</i> 9% of theory, –176°C (ethyl ac		}-aminocarbo	nylmethyl]benzo	oic acid	45
50	Calc.: Found:	C 68.73 68.30	H 6.55 6.48	N 7.29 7.45	•		50
55	Yield: 53.9	n <i>yl-2-piperidino-be</i> 9% of theory, –122°C (ethanol)		arbonylmethy	l]benzoic acid		55
60	Calc.: Found:	C 72.11 72.45	H 7.15 7.04	N 7.64 7.65	m/e = 366 m/e = 366		60
	4-[(1-(5-C)	/ano-2-piperidino-	phenyl)-ethyl)	-aminocarboi	nylmethyl]benzo	oic acid	

4-[(1-(5-Cyano-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid Yield: 71.6% of theory, M.p.: 198–200°C (ether)

31						GB 2 090 834A	31
	Calc.: Found:	C 70.57 70.17	H 6.44 6.38	N 10.73 11.00			
5	Prepare sodium hy Yield: 73.		<i>o-phenyl)-eth</i> conding dieth	<i>yl)-aminocarbo</i> yl ester by sap	onylmethyl]ber conification wit	zoic acid th 2.5 equivalents of	5
10	Calc.: Found:	C 67.30 67.76	H 6.38 6.62	N 6.82 6.85			10
15	semihydra Yield: 85.				ethyl)-aminoca	rbonylmethyl]benzoic acid	15
20	Calc.: Found:	(× 0.5 H₂O)		C 66.49 66.56	H 6.74 6.65	N 6.46 6.46	20
25	Viold: 659	coxy-1-(2-piperiding % of theory, 5-157°C (decomp m/e = 382 m/e = 382				zoic acid	25
30	Yield: 64.	thloro-2-(2-methyl- 1% of theory. 5–198°C (ethyl ac		henyi)-ethyi)-aı	minocarbonyln	nethyl]benzoic acid	30
35	Calc.: Found:	C 66.57 66.01	H 6.56 6.25	CI 8.54 8.32	N 6.75 6.90		35
40	Yield: 86	minocarbonyl-2-p % of theory, 1–235°C (ethyl ac			nocarbonylme	thyl]benzoic acid	40
	Calc.: Found:	C 67.46 67.96	H 6.65 6.68	N 10.26 10.11			
45	Yield: 67	4-Methyl-piperidin .7% of theory, 3-175°C (chlorofo		hyl)-aminocarb	onylmethyl]be	nzoic acid	45
50	Calc.: Found:	C 72.61 72.20	H 7.42 7.36	N 7.36 7.45			50
55	Conversed acid in ise Yield: 32	Piperidino-phenyl)- sion of the viscous opropanolic solution of theory, 2-230°C (decom	s betain (729 on.	nylaminocarbor 6 crude) into tl	nylmethyl]benz ne hydrochlori	zoic acid hydrochloride de by means of hydrochlori	55
60	Calc.: Found:	C 66.25 66.07	H 7.01 6.37	CI 8.50 8.37	N 6.71 6.58		60

	2-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid							
5	Yield: 7% of th M.p.: 135°C (c				٠		5	
	Calc.: Found:	C 72.10 72.29	Н 7.15 7.03	N 7.64 7.37				
10	3-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid Yield: 86% of theory, M.p.: 205–207°C							
15	Calc.: Found:	C 72.11 72.30	H 7.15 7.29	N 7.64 7.71			15	
20	3-Chloro-4-[(1-(2-piperidino-phenyl)-ethyl)-amino-carbonylmethyl]benzoic acid Yield: 38% of theory, M.p.: from 175°C sintering, from 190°C clear melt							
	Calc.:	C 65.91	H 6.29 6.32	CI 8.84 9.05	N 6.99 6.77			
25	Found:	65.42				ethyl)benzoic acid	25	
•	4-[(1-(2-(1,2,3,4-Tetrahydro-isoquinoline-2-yl)-phenyl)-ethyl)-aminocarbonylmethyl)benzoic acid Yield: 59% of theory, M.p.: 207–209°C							
30	Calc.: Found:	C 75.34 75.30	H 6.32 6.29	N 6.76 6.67	٠		30	
35	4-[(1-(3-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid Yield: 33% of theory, M.p.: 206–208°C						35	
40	Calc.: Found:	C 72.09 72.04	H 7.15 7.14	N 7.64 7.57			40	
	4-[(1-(6-Chloro-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid Yield: 35% of theory, M.p.: 148–150°C							
45	Calc.: Found:	C 65.91 65.45	H 6.28 6.36	CI 8.84 9.63	N 6.98 6.84		45	
50	4-[(1-(6-Methyl-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid Yield: 33% of theory, M.p.: 170°C						50	
55	Calc.: Found:	C 72.60 72.45	H 7.41 7.34	N 7.36 7.32			55	
60	4-[(1-(2-(Octahydro-isoindole-2-yl)-phenyl)-ethyl)-aminocarbonyl]benzoic acid Yield: 64% of theory, ) M.p.: 130°C						60	
	Calc.: Found:	C 73.85 73.60	H 7.43 7.47	N 6.89 6.72				

•	• • • • • • •						
æ	4-[(1-(2-Deca Yield: 71% o M.p.: 220-2	f theory.	oline-2-yl)-phe	nyl)-ethyl)-am	aminocarbonylmethyl]benzoic acid		
5	Cala .	. C 74.25	H 7.66	N 6.66	m/e = 420		
	Calc.:	74.45	7.50	6.58	m/e = 420		
	Found:	74.40	7.00	0.00			
10	4-[(1-(2-(1,2,3,4,5,6,7,8-Octahydro-isoquinoline-2-yl)-phenyl)-ethyl)-aminocarbonylmethyl]ben-						
10	zoic acid						
	Yield: 99% of theory,						
	M.p.: 70°C (						
	•				N 0 54 /a 440	15	
15	Calc.: (	× 0,5 H <sub>2</sub> O)	C 73.05	H 7.30	N 6.54  m/e = 418	13	
	Found:		73.00	7.16	5.98  m/e = 418		
					•		
		0 '!d:	-6(l) -4b.(l)	aminacarbar	wlmethyΠhenzoic acid		
	4-[(1-(4-Chio	ro-2-pipeiaino-	рпепуіј-витут	-ammocarbor	nylmethyl]benzoic acid	20	
20	Yield: 82.1%	or theory,			•		
	M.p.: 200-2	.U2 C					
	Calc.:	C 65.91	H 6.29	CI 8.84	N 6.99		
	Found:	66.06	6.40	9.01	6.93		
25	round.	55.55				25	
25							
	4-[(1-(4-Meti	hyl-2-piperidin	o-phenyl)-ethy	l)-aminocarbo	onylmethyl]benzoic acid		
	Yield: 66.5%	of theory,			·		
	M.p.: 110-1	15°C	= 40	N 7 00		30	
30	Calc.:	C 72.60	H 7.42	N 7.36		30	
	Found:	72.50	7.52	7.46	•		
	4 sta ni veridina kambudad) aminagarhanylmethylihenzaic scid						
25	4-[(2-Piperidino-benzhydryl)-aminocarbonylmethyl]benzoic acid 35 Yield: 88% of theory,					35	
35	M.p.: 232-2	234°C					
	W.p., 252-2						
	Calc.:	C 75.68	H 6.59	N 6.54			
	Found:	75.16	6.52	6.74			
40						40	
	4-[(5-Chloro-	-2-piperidino-b	enzhydryl)-am	inocarbonyln	nethyl]benzoic acid		
	Yield: 78.59	6 of theory,					
	M.p.: 255-2		= 00	01.7.00	N 6.05	45	
45	Calc.:	C 70.05	H 5.88	CI 7.66 7.36	6.06		
	Found:	70.50	5.76	7.30	0.00		
	4-[(1-(4-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid						
EΛ	4-[(1-(4-F1pe	of theory	oury ij arminoo	2.201.,	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	50	
90	O Yield: 81% of theory, M.p.: 208–210°C						
	.vi.p 200-						
	Calc.:	C 72.11	H 7.15	N 7.64			
	Found:	72.24	7.26	7.54			

	4-[(1-(2-(4-Methyl-piperazino)-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid Yield: 65% of theory, M.p.: 150153°C						
5	Calc.: Found:	C 69.27 69.62	H 7.13 7.65	N 11.02 10.64			5
10	4-[(1-(2-(4-Benzyl-piperazino)-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid hydrochloride Yield: 32% of theory, M.p.: 180°C						10
15	Calc.: Found:	C 68.07 67.85	H 6.53 6.56	CI 7.18 7.18	N 8.51 8.51		15
20	4-[(1-(2-(4-p-Chlorophenyl-piperazino)-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid Yield: 75% of theory, M.p.: 212°C (decomp.)						20
	Calc.: Found:	C 67.84 67.74	H 5.90 6.22	CI 7.42 7.59	N 8.79 8.82		
25	4-[(α-Cyclohexyl-2-piperidino-benzyl)-aminocarbonylmethyl]benzoic acid Yield: 33% of theory, M.p.: 199–202°C						25
30	Calc.: Found:	C 74.62 74.60	H 7.89 7.54	N 6.45 6.66			30
35	(+)-4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid $\times$ 0.3 H <sub>2</sub> O Yield: 40% of theory, M.p.: 107°C (decomp. (isopropanol/ether) [ $\alpha$ ] <sup>20</sup> <sub>D</sub> = +7.3° (c = 1; methanol)						35
40	Calc.: Found:	(× 0.3 H <sub>2</sub> O)	C 71.02 70.90		N 7.52 7.42	m/e = 366 m/e = 366	40
45	( — )-4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid sodium salt Crude yield of betain: 77% of theory,						45
	Calc.: Found:	m/e = 366 m/e = 366					
50	Conversion into the sodium salt by means of 1 equivalent of sodium hydroxide solution in ethanol.  M.p. of the sodium salt: 190°C (decomp.)						50
55	4-[(1-(2-Piperidino-phenyl)-ethenyl)-aminocarbonylmethyl]benzoic acid 5 Yield: 53.6% of theory, M.p.: 158–160°C (ethanol)						<b>55</b>
,	Calc.: Found:	C 72.51 72.40	H 6.64 6.34	N 7.69 7.51			

	_					
					,	
	Yield: 78.79	6 of theory,		nyl)aminocarl	oonylmethyl]benzoic acid	-
5	M.p.: 198-4 Calc.: Found:	200°C (acetone) C 66.24 65.74	H 5.81 5.72	Cl 8.88 9.37	N 7.02 7.10	5
10	4-[(α-Cycloh Yield: 21% Μ.ρ.: 213–2	of theory,	eridino-benzy	l)-aminocarbo	nylmethyl]benzoic acid	· 10
15	Calc.: Found:	C 74.97 74.73	H 7.46 7.52	N 6.48 6.48		15
	4-[(1-(6-Chic Yield: 39% M.p.: 162°C	of theory,	-phenyl)-ethe	nyl)-aminoca	bonylmethyl]benzoic acid	20
20	Calc.: Found:	C 66.24 66.48	H 5.81 5.84	CI 8.88 8.88	N 7.02 m/e = 398/400 6.85 m/e = 398/400	
25	4-[(1-(6-Met Yield: 49% M.p.: 128-	of theory,	p-phenyl)-eth	enyl)-aminoca	rbonylmethyl]benzoic acid	25
30	Calc.: Found:	m/e = 378 m/e = 378				30
35	Yield: 65%	eridino-phenyl)- <sub>l</sub> of theory, n): 185–187°C	(ethyl acetat	e)	nethyl]benzoic acid	35
	Calc.: Found: M.p. (E-forn	(Z-form) n): 108–110°C	C 72.99 73.10	H 6.92 6.99	N 7.40 7.56	
40	4-[(1-(5-Hyd Saponific	droxy-2-piperidication with 2.5 e % of theory,	n <i>o-phenyl)-e</i> t equivalents o	<i>hyl)-aminocai</i> f sodium hyd	bonylmethyl]benzoic acid semihydroxide.	40 drate
45	Foam (from		C 67.50 67.11	H 6.95 7.15	N 7.16 . 6.87	45
50	4-[(1-(2-(2- Yield: 62% M.p.: 169-	of theory,	no)-phenyl)-e	thyl)-aminoce	rbonylmethyl]benzoic acid	50
55	Calc.: Found:	C 72.11 71.96	H 7.15 6.82	N 7.64 7.51		55
60	<i>4-[(1-(5-Am</i> Yield: 19.2 M.p.: 210°	% of theory,	iperidino-phe	nyl)-ethyl)-an	ninocarbonylmethyl]benzoic acid	60
	Calc.: Found:	C 59.30 58.80	H 6.11 5.87	N 9.43 9.06	m/e = 445 m/e = 445	

	<i>4-[(1-(2-Piperidino-phenyl)-propyl)-aminocarbonylmethyl]benzoic acid</i> Yield: 71.4% of theory, M.p.: 208–210°C (ethanol)	
5	Calc.: C 72.61 H 7.42 N 7.36 Found: 72.30 7.44 7.45	5
10	Example 10 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid A solution of 13.5 g (0.338 mol) of sodium hydroxide in 50 ml of water was added to 88.8 g (0.225 mol) of 4-[(1-(2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester in 890 ml of ethanol and the mixture was stirred at an internal temperature of 60°C until no	10
15	starting product could be detected in the thinlayer chromatogram (approx. 45 minutes). After	15
20	chloride and washed with a little water. After drying the organic phase over sodium sulfate, the solution was filtered and the solvent was removed in vacuo, whereby a solid evaporation residue of 57.5 g was obtained.	20
25	The ethanolic hydrochloric filtrate (pH = 5.8) was adjusted to pH = 5.0 by means of semi-concentrated hydrochloric acid, then the ethanol was distilled of <i>in vacuo</i> and the evaporated solution was cooled in ice. The resultant precipitate was filtered off, dissolved in methylene chloride, separated from the aqueous phase, the methylene chloride solution was dried, filtered and evaporated <i>in vacuo</i> . The solid evaporation residue obtained was 13.0 g. Both evaporation residues (together 70.5 g) were recrystallized from the 5-to 6-fold amount of ethanol/water (80/20) under addition of activated charcoal.	25
30	Yield: 62% of theory, M.p.: 163–164°C	30
35		35
	If on completion of the saponification, after the addition of water and cooling to 25°C immediately the pH is adjusted to 5.0, and then continued as described above, 75.9% of the dried evaporation residue may be obtained without further processing the ethanolic hydrochloric filtrate, which even before the final recrystallization gave a correct elementary analysis.	
40	M.p.: 172-176°C	40
45	Calc.: C 72.11 H 7.15 N 7.64 Found: 71.90 7.08 7.52	45
	Analogously to Example 10 the following compounds were prepared:	
50	4-[(1-(5-Methyl-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid Yield: 56.6% of theory, M.p.: 215–217°C (ethanol)	50
	Calc.: C 72.61 H 7.42 N 7.36 Found: 72.71 7.49 7.25	
55	4-[(α-Carboxy-2-piperidino-benzyl)-aminocarbonylmethyl]benzoic acid × 0.66 H₂0 Prepared by saponification of the 4-[(α-methoxycarbonyl-2-piperidino-benzyl)-aminocarbonylmethyl]benzoic acid ethyl ester with 2.5 equivalents of sodium hydroxide. Yield: 72.2% of theory,	55
60	Calc.: (× 0.66 H₂O) C 64.69 H 6.33 N 6.85	60
65	Found: 64.64 6.23 6.61	65
	•	

37	GB 2 U9 U 8 3 4 A	3/:
5	4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid sodium salt monohydrate 500 mg (1.26 m mol) of 4-[(2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester in 5 ml of ethanol were stirred together with 1.26 ml of 1N sodium hydroxide solution for 1 hour at 50°C. After cooling to 0°C, the precipitated crystals were filtered off and washed with cold ethanol and with ether. Yield: 238 mg (48.6% of theory), M.p.: 245-250°C	5
10	Calc.: (× 1 H₂O) C 65.01 H 6.69 N 6.89 Found: 65.40 6.83 6.72	10
	Analogously to Example 11 the following compound was prepared:	
15	4-[(1-(5-Methoxy-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid sodium salt monohydrate Yield: 17.5% of theory, M.p.: 212-215°C	15
20	Calc.: $(\times 1 \text{ H}_2\text{O})$ C 63.28 H 6.70 N 6.42 Found: 63.20 6.82 6.51	20
25	Calc.: (× 1 H₂O) C 66.40 H 7.29 N 6.76	25
30	Example 12 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid sodium salt × 0.6 H <sub>2</sub> O 8.4 g (0.0229 mol) of 4-[(1-(2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid	30
35	hydroxide solution were added with stirring and stirring was continued for 30 minutes. After cooling to 20°C. a precipitate was obtained. After cooling to 0°C, the precipitate was filtered and washed with cold ethanol and ether. The precipitate thus obtained, of m.p. 250–251°C, was recrystallized from ethanol/water (7/3). Yield: 7.2 g (78.6% of theory), M.p.: 253–255°C	35
40	Calc.: (× 0.6 H <sub>2</sub> 0): C 66.18 H 6.61 N 7.02 Found: 66.10 6.64 7.13	40
45	Example 13 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid 100 mg (0.237 m mol) of 4-[(1-(2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid-tert.butyl ester in 5 ml of benzene were heated together with some crystals of p-toluene sulfonic acid hydrate to reflux temperature for half a day. According to the thinlayer chromato- gram then no starting product could be detected, and according to the R <sub>r</sub> -value and mass spectrum the desired product was formed.	45
50	M.p.: 163–165°C	50
	Calc.: m/e = 366 Found: m/e = 366	<del>-</del> -
55	Example 14 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid 0.46 g (1 m mol) of 4-[(1-(2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid benzyl ester in 20 ml of ethanol were hydrogenated at 0.25 g of palladium/charcoal at 50°C	55
60	and a hydrogen pressure of 5 bar. After 5 hours the catalyst was filtered off over celite and the filtrate was evaporated in vacuo. The evaporation residue was recrystallized from ethanol/water (8/2).  Yield: 0.26 g (71% of theory), M.p.: 163–165°C	60

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	Calc.: C 72.11 H 7.15 N 7.6 Found: 72.30 7.25 7.8	
	Example 15 5 4-[(1-(5-Chloro-2-piperidino-phenyl)-ethyl)-aminocc 2.54 g (0.02 mol) of oxalyl chloride were drop (0.01 mol) of N-[(1-(5-chloro-2-piperidino-phenyl) disulfide and subsequently 2.67 g (0.02 mol) of again the same amounts of oxalyl chloride and all was heated subsequently for 3 hours up to 50°C, were added and the reacting mixture was extracted dried and filtered and evaporated in vacuo. The echromatography on silica gel (chloroform/methan)	ped at 0 to 5°C to a stirred solution of 3.57 g ethyl]-N-[phenacetyl]amine in 16 ml of carbon aluminium chloride were added. After one hour uminium chloride were added and the mixture After cooling, ice water and hydrochloric acid d with chloroform. The organic extract was vaporation residue was purified by column
15	Yield: 0.60 g (15% of theory), 15 M.p.: 213-214°C (ether)	15
	Calc.: C 65.91 H 6.29 CI 8.9 Found: 66.13 6.05 8.9	
20	20 Example 16 N-[4-Acetyl-phenacetyl]-N-[1-(5-chloro-2-piperidin	anhenyll-ethyllemine
25	A solution of 0.6 ml (8.43 m mol) of acetyl chl at an internal temperature of 0 to 5°C to 1.12 g of methylene chloride. Subsequently, at 0 to 5°C 25 chloro-2-piperidinophenyl)-ethyl]-N-[phenacetyl]ar with stirring. The reaction mixture was stirred for decomposing under cooling with ice water and hy was separated and the aqueous phase was extract	oride in 5 ml of methylene chloride was added (8.43 m mol) of aluminium chloride in 10 ml, a solution of 1 g (2.81 m mol) of N-[1-(5-nine in 5 ml of methylene chloride was added 1 hour at 3°C and for 2 days at 20°C. After ordrochloric acid, the methylene chloride phase ted with chloroform. The combined oganic
30	phases were dried over sodium sulfate, filtered ar 30 residue was purified by column chromatography (Yield: 0.28 g (25% of theory), M.p.: 160–161°C	od evaporated <i>in vacuo</i> . The evaporation on silica gel (toluene/acetone = 4:1).
35	Calc.: C 69.24 H 6.82 CI 8.89 35 Found: 69.55 6.99 9.45	N 7.02 m/e = 398/400 6.85 m/e = 398/400 35
	Example 17 4-[(1-(5-Chloro-2-piperidino-phenyl)-ethyl)-aminoc A solution of 1.23 g (0.0031 mol) of N-[4-aced phenyl)-ethyl]amine in 12 ml of dioxan was adde sodium hypobromite solution [prepared from 1.8 dissolved in 9 ml of water, and 0.72 ml (0.014 minutes at 35-40°C aqueous sodium hydrogen s mixture was evaporated in vacuo. The residue was with 2N hydrochloric acid and extracted with ethe and filtered, and evaporated in vacuo. The evapo Yield: 0.14 g (11% of theory),	cyl-phenacetyl]-N-[1-(5-chloro-2-piperidino-d over 15 minutes at 35–40°C to a stirred 40 4 g (0.046 mol) of sodium hydroxide, mol) of bromine under ice cooling]. After 40 culfite solution and water was added and the s dissolved with water, acidified under cooling er/ethyl acetate. The organic phase was dried 45
	M.p.: 213–215°C	25 N.O.O.
50	50 Calc.: C 65.91 H 6.29 Cl 8.5 Found: 65.78 5.98 8.	
	Analogously to Example 17 the following comp	oound was prepared:
55	55 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylm Yield: 15% of theory, M.p.: 170-171°C	ethyl]benzoic acid 55
60	Calc.: C 72.11 H 7.15 N 7. 60 Found: 72.45 7.01 7.	64 48 60
65	Example 18 4-[1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylm Prepared from 4-[(1-(2-piperidino-phenyl)-ethyl 65 tion with active manganese dioxide in absolute active	-aminocarbonylmethyl]benzyl alcohol by oxida-

```
chromatography on silica gel (chloroform/acetone = 20:1).
   Yield: 4% of theory,
    Mp.:159°C
                                                                                                    5
5
                                H 7.48
                                             N 7.99
                   C 75.40
    Calc.:
                                   7.18
                                               7.67
                     75.05
    Found:
    Example 19
                                                                                                   10
10 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid
      Prepared from 4-[(1-(2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzaldehyde by heat-
   ing with silver oxide in the presence of 1N sodium hydroxide solution for 20 minutes on a
   steam bath, subsequent acidification with 2N sulfuric acid at pH = 5, extraction with ethyl
   acetate and purification by column chromatography on silica gel (toluene/acetone = 1:1).
                                                                                                   15
15 Yield: 3% of theory
    Mp.: 168-170°C
                m/e = 366
    Calc.:
    Found:
                m/e = 366
                                                                                                   20
20
    Example 20
    4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester
      5.5 g (0.014 mol) of 4-[(1-(2-piperidino-phenyl)-ethenyl)-aminocarbonylmethyl]benzoic acid
    ethyl ester in 110 ml of ethanol were hydrogenated at 1.5 g of palladium/charcoal (10%) at
25 20°C and a hydrogen pressure of 5 bar. After 30 minutes the catalyst was filtered off over celite
   and the filtrate was evaporated in vacuo to a volume of 20 ml. 100 ml of petroleum ether were
    adde and the mixture was cooled to 0°C.
    Yield: 4.7 g (85.5% of theory),
    M.p.: 152-154°C
                                                                                                   30
30
                                H 7.66
                                             N 7.10
    Calc.:
                   C 73.07
                                                7.08
                                   7.63
                     72.80
    Found:
      Analogously to Example 20 the following compound was prepared:
                                                                                                   35
35
    4-[1-(2-Piperidino-phenyl)-propyl)-aminocarbonylmethyl]benzoic acid ethyl ester
    Yield: 70.8% of theory,
    M.p.: 132-134°C
                                                                                                    40
                                 H 7.90
                                             N 6.86
                   C 73.00
40 Calc.:
                                                6.77
                     73.71
                                   7.88
    Found:
    Example 21
    4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid
      100 mg (0.2744 m mol) of 4-[(1-(2-piperidino-phenyl)-ethenyl)-aminocarbonylmethyl]benzoic
                                                                                                    45
    acid in 5 ml of absolute ethanol were hydrogenated at 50 mg of palladium/charcoal (10%) at
    20°C and at a hydrogen pressure of 1 bar under shaking. After 1.5 hours the catalyst was
    filtered off and the filtrate was evaporated in vacuo.
    Yield: 91% of theory,
                                                                                                    50
50 M.p.: 170-171°C
                 m/e = 366
    Calc.:
                 m/e = 366
    Found:
                                                                                                    55
55 Example 22
    4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid semihydrate
    200 mg (0.5014 m mol) of 4-[(1-(5-chloro-2-piperidinophenyl)-ethenyl)-aminocarbonylmethyl]-
    benzoic acid in 10 ml of absolute ethanol were hydrogenated at 100 mg of palladium/charcoal
    (10 %) at 50°C and at hydrogen pressure of 1 bar under shaking. After 1.5 hours the catalyst
60 was filtered off, 5 ml of water were added, adjusted to pH = 6 by means of 1N-sodium
                                                                                                    60
    hydroxide solution and the ethanol was evaporated in vacuo. A colourless precipitate was
    obtained, which was filtered after cooling.
    Yield: 100 mg (53.1% of theory),
    M.p.: 135°C
```

	Calc.: Found:	(× 0.5 H <sub>2</sub> O)	C 70.36	H 7.24 70.31 <sup>7</sup> .44	N 7.46 7.78	m/e = 366 m/e = 366		
5	5 Example 23 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester 1.6 ml of conc. sulfuric acid were added in little drops to a mixture of 2 g (9.74 m mol) of 1- (2-piperidino-phenyl)-ethanol and 4 g (21.1 m mol) of 4-cyanomethyl-benzoic acid ethyl ester whilst stirring and cooling with ice by keeping the internal temperature at 35 to 40°C.							
10	Subseque mol) of 4-heating walcohol co	ntly, the mixture was cyanomethyl benzo as continued for 1 ald be detected in	as heated for ic acid ethyl hour at 80°C the thinlayer	r 2.5 hours in a ester and 0.8 cand for 3 hours chromatogran	a bath of 80° ml of conc. si urs at 100°C. n. After coolin	C, further 2 g (10.5 m ulfuric acid were added and After that time no starting g to 20°C the mixture was added. After extracting		
	several tinevaporate gel (toluer Yield: 0.6	nes with ethyl acets d <i>in vacuo.</i> The eve	ate, the organ aporation resi ). From the p	nic extract was idue was purifi	dried over so led by column	edium sulfate, filtered and chromatography on silica idino-styrol were isolated.	15	
20	Calc.: Found:	C 73.07 73.26	H 7.66 7.55	N 7.10 6.90			20	
25	0.4 ml of 4-carbo ethylamin	hloro-2-piperidino-p (5.55 m mol) of the exy-phenylacetic act e in 10 ml of absol	ionyl chloride id and of 1.3 lute pyridine,	e were added t 32 g (5.55 m r , whereby the i	o a stirred sol nol) of (5-chlo internal tempe	ution of 1 g (5.55 m mol) ro-2-piperidino-phenyl)- rature rised from 20°C to	25	
30	35°C. The deep-brown reaction mixture was stirred for 3 hours at 20°C and evaporated in vacuo. The evaporation residue was distributed between water (at pH = 3 after addition of 2N hydrochloric acid) and chloroform. The organic extract was dried and filtered and evaporated in vacuo. The evaporation residue was purified by column chromatography on silica gel (chloroform/methanol = 10:1).  Yield: 1.06 g (48% of theory),							
35		2-214°C (ether)					35	
	Calc.: Found:	C 65.91 65.79	H 6.29 6.01	CI 8.85 8.69	N 6.99 6.87			
40	Analoge	ously to Example 2	4 the following	ng compounds	were prepare	d:	40	
	4-[(1-(2-P Yield: 529 M.p.: 169	iperidino-phenyl)-et % of theory, 3–171°C	thyl)-aminoca	rbonylmethyl]l	benzoic acid			
45	Calc.: Found:	C 72.11 71.84	H 7.15 6.87	N 7.64 7.72			45	
	4-[(1-(2-(4	1-Oxo-piperidino)-pl	henyi)-ethyi)-	aminocarbonyl	methyl]benzoi	ic acid		
50	50 Yield: 32% of theory, M.p.: 177–180°C (decomp.) (acetone/petroleum ether)						50	
55	Calc.: Found:	C 69.46 69.62	Н 6.36 6.41	N 7.36 7.50			55	
60	Yield: 23	4-Hydroxy-piperidin .5% of theory, 6–179°C (decomp.				enzoic acid × 0.66 H₂O	60	
	Calc.: Found:	(× 0	).66 H₂O) 67.12	C 66.97 6.78	H 6.81 7.26	N 7.10		

					·····	
	4-[(1-(2-Piper Prepared fr Yield: 51% o	ridino-phenyl)-et rom 4-cyano-pho	hyl)-aminoca enyl acetic ac	<i>rbonylmethyl</i> id.	[benzonitrile	
5	M.p.: 155-1	57°C (ethyl ace	tate)			5
	Calc.: Found:	C 76.05 76.41	H 7.25 7.10	N 12.09 12.20	•	
10	Example 25					10
	4-[(1-(2-Piper Prepared fr by lithium alu Yield: 39% o	ıminium hydride f theory.	eridino-pheny	/i)-ethyi)-amır	locarbonylmethyljbenzoic acid ethyl ester	15
15	M.p.: 104-1	06°C C 74.96	н 8.00	N 7.94		
	Calc.: Found:	74.80	7.80	7.80		
20	A solution zyl chloride [	of 3.7 g (10 m m.p.: 123–125	mol) of 4-[(1 i°C; prepared orm1 in 35 m	-(2-piperiding from the alc of absolute	[benzyl malonic acid diethyl ester opening] ben-phenyl)-ethyl)-aminocarbonylmethyl]ben-ohol described in Example 25 by means ethanol was added to a solution of	20
25	sodium malo absolute etha potassium ion vacuo, the ex-	nic acid diethyl nol and 4.8 (3) dide was added raporation resid	ester [prepar	nalonic acid of ture was refluited to neutral reanic extract	diethyl ester]. A catalytic amound of ixed for 16 hours. After evaporating in by means of hydrochloric acid and was dried over sodium sulfate, filtered	25
30	and and evap	oorated <i>in vacu</i> e (toluene/aceton (60% of theory)	o. The evapo le = 6:1).	ration residue	was purified by column chromatograph	30
35	Calc.: Found:	m/e = 494 m/e = 494	•			35
40	5 ml of 11 [(1-(2-piperio of ethanol. A	N-sodium hydro dino-phenyl)-eth after stirring for	xide solution yl)-aminocart 2 hours at 50	were added to conylmethyl]t 0°C, the mixt dded. The for	hyl]-phenyl]propionic acid to a solution of 0.85 g (1.7 m mol) of 4- penzyl malonic acid diethyl ester in 18 ml ture was evaporated in vacuo, and water timed precipitate was filtered off, dried in the carbon dioxide was liberated. The	40
45	product was	purified by colu g (22.2% of the	imn chromat	ography on s	by carbon dioxide was liberated. The ilica gel (chloroform/methanol = 20:1).	45
50	Calc.: Found:	C 73.06 72.64	H 7.67 7.42	N 7.10 6.81	m/e = 394 m/e = 394	50
	Example 28 4-[(1-(2-Pipe Prepared I	oridino-phenyl)-c by heating crud	e N'-[4-[(1-(2	z-piperiainopi	rl]benzaldehyde nenyl)-ethyl)-aminocarbonylmethyl]benzoyl]- 160–170°C in ethylene glycol [prepared ethyl]benzoic acid and tosyl-hydrazine with	
99	carbonyl diir Yield: 10% M.p.: 159°C	nidazole in tetra of theory,	anydrofuranj.			60
60	Calc.: Found:	C 75.40 74.99	H 7.48 7.24	N 7.99 7.60		00
65	Example 29 4-[(1-(2-Pipe 0.50 g (1	- ::-!:	ethyl)-aminoc 4-[(1-(5-chlo	earbonylmeth) pro-2-piperidii	vl]benzoic acid no-phenyl)-ethyl)-aminocarbonylmethyl]ben	- 65

_	(10%) at 50°C celite and after ethyl acetate.	and a hydroge evaporating in	n pressure of vacuo the rea	nydrogenated at 0.25 g of palladium/charcoal is bar. After 2 hours the catalyst was filtered off due was distributed at pH = 6 between water a d with water, dried and filtered and evaporated	nd <i>in</i>
ð	<i>vacuo</i> . Yield: 0.31 g ( M.p.: 170–17		),		5
10	Calc.: Found:	C 72.11 71.76	H 7.15 6.98	N 7.64 7.51	10
	Analogously	to Example 29	the following	compounds were prepared:	
15	4-[(2-(2-Piperio Yield: 68.5% o M.p.: 213-21	of theory,	oropyl)-aminod	rbonylmethyl]benzoic acid	15
	Calc.: Found:	C 72.61 72.43	H 7.42 7.25	N 7.36 7.40	
20					20
25	4-[(1-(2-Dimet) Yield: 53.3% o M.p.: 165-16	of theory,		carbonylmethyl]benzoic acid )	25
20	Calc.: Found:	C 69.92 69.88	H 6.79 6.83	N 8.59 8.49	25
30	4-[(2-Pyrrolidin Yield: 55% of M.p.: 212-21	theory,		/l]benzoic acid	30
35	Calc.: Found:	C 70.99 70.97	H 6.55 6.91	N 8.28 8.15	35
	4-[(1-(2-Pyrroli	done-phenyl)-e:	thyl)-aminocar	onylmethyl]benzoic acid	
40	Yield: 25% of M.p.: 155–15		ether)		40
	Calc.: Found:	C 71.57 71.22	H 6.86 6.75	N 7.95 8.42	
45	4-[(2-Piperiding Yield: 60.4% o M.p.: 175-17	of theory,	ocarbonylmeth	l]benzoic acid	45
50	Calc.: Found:	C 71.57 71.48	H 6.86 7.00	N 7.95 8.09	50
55	4-[(2-(2-Piperio Yield: 60.4% o M.p.: 164-16	of theory,		nylmethyl]benzoic acid	55
	Calc.: Found:	C 72.11 72.35	H 7.15 7.18	N 7.64 7.76	,

•						•
	4-[(1-(2-(2-Metal Yield: 90.9% o M.p.: 171-173	f theory,		l)-aminocarbonylmethy	l]benzoic acid	
· 5	M.p.: 171-173	C (berroleui	ii ether/acet	nej		5
	Calc.: Found:	C 72.61 72.30	H 7.42 7.39	N 7.36 7.43		
10	4-[(1-(2-(3-Meti Yield: 86.3% o M.p.: 170–173	f theory,		l)-aminocarbonylmethy ne)	I/benzoic acid	10
15	Calc.: Found:	C 72.61 72.20	H 7.42 7.28	N 7.36 7.12		15
20	4-[(1-(2-Diprop) Yield: 51.1% o M.p.: 175-178	f theory,		ocarbonylmethyl]benzo	oic acid	20
	Calc.: Found:	C 72.22 72.10	H 7.91 8.05	N 7.32 7.69		
25						25
20	4-[(1-(2-Piperide Yield: 86% of t M.p.: 215-217	heory,	-methyl-propy	l)-aminocarbonylmethy	l]benzoic acid	
30	Calc.: Found:	C 73.06 73.10	H 7.67 7.55	N 7.10 6.99		30
35	4-[(1-(2-Piperide Prepared from methyl ester. Yield: 37.2% of M.p.: 145-147	n 4-[(1-(5-chl f theory,	<i>hyl)-aminoca</i> oro-2-piperid	<i>bonylmethyl]benzoic a</i> no-phenyl)-ethyl)-amino	cid methyl ester ocarbonylmethyl]benzoic acid	35
40	Calc.: Found:	C 72.61 72.47	H 7.42 7.30	N 7.36 7.56		40
45	4-[(2-Piperidino Prepared from Yield: 60% of t M.p.: 85-86°C	n 4-[(5-chloro heory,	-2-piperidino		l]benzoic acid methyl ester.	45
50	Calc.: Found:	C 71.57 71.48	H 6.86 6.92	N 7.96 8.39		50
55	N-Phenacetyl-N- Prepared from Yield: 54.6% of M.p.: 120-121	า N-[1-(5-chlo f theory,	ro-2-piperidi	o-phenyl)-ethyl]-N-phe	nacetyl-amine.	55
60	Calc.: Found:	C 78.22 77.90	H 8.13 8.24	N 8.69 8.75		60
	2.0 g (0.004 acid methyl este	7 mol) of 4-[ or in 20 ml o	1-(5-nitro-2- f dimethyl fo	iperidino-phenyl)-ethyl mamide were hydroge	benzoic acid methyl ester )-aminocarbonylmethyl]benzoic nated at 0.2 g of palladium/- essure of 1 bar. When the	65

5	hydrogen absorption was finished (2 hours), the catalyst was filtered off over celite and evaporated to dryness <i>in vacuo</i> .  Yield: 1.8 g (95% of theory),  M.p.: 140–142°C (toluene).  Analogously to Example 30 the following compounds were prepared:	5
	4-[(1-(5-Amino-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester Yield: 97.8% of theory, M.p.: 148–149.5°C (cyclohexane)	
10	Calc.: C 70.39 H 7.63 N 10.26 Found: 70.20 7.67 9.60	10
15	4-[(1-(5-Amino-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid Prepared from 4-[(1-(5-nitro-2-piperdino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid. Yield: 85.7% of theory, M.p.: 223-225°C (ether)	15
20	Calc.: C 69.27 H 7.13 11.02 Found: 69.18 7.04 11.35	20
25	N-[4-Amino-phenacetyl]-N-[1-(2-piperidino-phenyl)-ethyl]-amine dihydrochloride semihydrate Prepared from N-[4-nitro-phenacetyl]-N-[(1-(2-piperidinophenyl)-ethyl]amine. Conversion of the crude amino compound into the dihydrochloride in ethanol was by means of ethereal hydrochloric acid.  Yield: 17.5% of theory,	25
30	M.p.: 238°C (decomp.)	30
	Calc.: $(\times 2 \text{ HCl} \times 0.5 \text{ H}_2\text{O})$ C 60.12 H 7.21 Cl 16.91 Found: 60.52 7.52 17.05	
35	Example 31 4-[(1-(5-Bromo-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid A solution of 0.072 g (1.05 m mol) of sodium nitrite in 0.5 ml of water was added at an internal temperature of 0 to 5°C to 0.40 g (1.05 m mol) of 4-[(1-(5-amino-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid in 2 ml of semi-conc. aqueous hydrobromic acid. The	35
40	resultant diazonium salt solution was then added to 0.196 g of copper (I) bromide in 2 ml of 48% hydrobromic acid, whereby considerable formation of gas occurred. The reaction mixture was stirred for 1.5 hours at an internal temperature of 45–50°C, cooled and adjusted to pH 4 by means of 4N sodium hydroxide solution. After extraction with warm ethyl acetate, the extract was washed with water, dried and filtered. After evaporating in vacuo, the obtained residue was	40
45	purified by column chromatography on silica gel (chloroform/methanol = 7:1).  Yield: 0.08 g (17% of theory),  M.p.: 212-213°C (ethyl acetate/petroleum ether)	45
50	Calc.: C 59.32 H 5.66 Br 17.94 N 6.29 Found: 59.30 5.71 17.85 6.48	50
	Analogously to Example 31 the following compound was prepared:	
55	4-[(1-(5-Chloro-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid Prepared by diazotization of 4-[(1-(5-amino-2-piperidinophenyl)-ethyl)-aminocarbonylmethyl]- benzoic acid in conc. HCl and Sandmeyer reaction with copper (I) chloride. Yield: 25.2% of theory, M.p.: 213-215°C	55
60	Calc.: C 65.91 H 6.29 CI 8.85 N 6.99 Found: 66.20 6.31 8.87 6.82	60

If the reaction is carried out in hydrochloric acid without copper (I) chloride, a yield of 19% of theory is obtained. Furthermore, 9% of the corresponding 5-hydroxy compound is obtained.

Exa	m	n	ما	32
CIZ	III	u	•	JZ

	Example 32	
5	4-[(1-(5-lodo-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester A solution of 0.17 g (2.44 m mol) of sodium nitrite in 0.52 ml of water was slowly added at 0 to 5°C whilst stirring to 1.0 g (2.44 m mol) of 4-[(1-(5-amino-2-piperidino-phenyl)-ethyl)- aminocarbonylmethyl]benzoic acid ethyl ester in 1.9 ml of semi-conc. hydriodic acid and the solution was warmed to 20°C over 1 hour. After heating for 2 hours at 100°C, the reaction mixture was cooled and extracted with ethyl acetate. The organic phase was washed with dilute	5
10	sodium bicarbonate solution and with Water, dried over solution surface, interest, and over solution surface, and over solution surface	10
15	Calc.: C 55.39 H 5.62 N 5.38 m/e = 520 Found: 55.95 5.53 5.05 m/e = 520	15
20	stirring at -5 to 0°C, to 2.0 g (4.88 m mol) of 4-[(1-(5-amino-2-piperiolito-piteriy)/-ct/y)/- aminocarbonylmethyl]benzoic acid ethyl ester in 4.0 ml of water and 3.5 ml of conc.	20
25	water into a 0°C solution, which was prepared from 0.568 g (6.34 m mol) of copper (I) water into a 0°C solution, which was prepared from 0.568 g (6.34 m mol) of copper (I) cyanide, 1.24 g (19 m mol) of potassium cyanide and 5.8 ml of water, whereby immediately a red-coloured precipitate was obtained. The reaction mixture was heated whilst stirring for 30 red-coloured precipitate was obtained. The reaction mixture was heated whilst stirring for 30 red-coloured precipitate was obtained.	25
30	95°C. The red-coloured spot was now no longer visible in the trimayer chromatogram.  The reaction mixture was cooled to 20°C and extracted with ethyl acetate. The organic extract was dried over sodium sulfate, filtered, and evaporated in vacuo. The evaporation residue was purified by two column chromatographies on silica gel ((a) toluene/acetone = 10:1, (b)	30
35	5-CI- and 5-H-compounds, the 5-cyano compound was obtained.  5 Yield: 0.186 g (9% of theory),  M.p.: 165-167°C (ether)	35
40		40
45	Example 34 4-[(1-(5-Aminosulfonyl-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester (a) A solution of 0.37 g (5.36 m mol) of sodium nitrite in 0.7 ml of water was added with stirring at 4 to 6°C to a suspension of 2.0 g (4.88 m mol) of 4-[(1-(5-amino-2-piperidino- phenyl)-aminocarbonylmethyl]benzoic acid ethyl ester in 2.02 ml of semi-conc. hydrochloric acid. Subsequently, 0.37 g (3.89 m mol) of magnesium chloride were added. The mixture thus obtained was dropped subsequently at 30°C to a solution, which was prepared from 4.9 ml of glacial acetic acid (saturated with sulfur dioxid) and 0.27 g of copper(II)chloride dihydrate.	45
50	glacial acetic acid (saturated with stitul dioxide) and discogen was formed. After stirring for 15 Thereby the internal temperature rose to 40°C and nitrogen was formed. After stirring for 15 minutes in a bath at 50°C, 7.5 ml of water were added and the mixture was extracted with chloroform. The organic extract was dried over sodium sulfate, filtered, and evaporated in vacuo. The viscous, red-brown evaporation residue (2.7 g; still chloroform-containing) contained besides the corresponding 5-chloro-compound the desired 4-[(1-(5-chlorosulfonyl-2-piperidino-	50
55	phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester.  (b) A solution of the evaporation residue obtained according to Example a) in 10 ml of chloroform was added at 2°C whilst stirring to 50 ml of conc. ammonia. After 30 minutes saturated sodim chloride solution was added to obtain separation of the phases. After extracting saturated sodim chloride solutions was added to obtain separation of the phases. The	55
60	with chloroform, the organic extract was dried and intered and ordeposited with the desired state of the evaporation residue was purified by column chromatography on silica gel (chloroform/methaevaporation residue was purified by column chromatography on silica gel (chloroform/methaevaporation residue was purified by column chromatography on silica gel (chloroform/methaevaporation residue was purified by column chromatography on silica gel (chloroform/methaevaporation residue was purified by column chromatography on silica gel (chloroform/methaevaporation residue was purified by column chromatography on silica gel (chloroform/methaevaporation residue was purified by column chromatography on silica gel (chloroform/methaevaporation residue was purified by column chromatography on silica gel (chloroform/methaevaporation residue was purified by column chromatography on silica gel (chloroform/methaevaporation residue was purified by column chromatography on silica gel (chloroform/methaevaporation residue) and silica gel (chlorof	60

	Calc.: Found:	m/e = 473 m/e = 473						
5	10 g (1.5	ethylamino-2-pip 89 m mol) of sod	lium-cyanoboi	o-hydride and afte	bonylmethyl]benzoic acid er 2 minutes 0.056 ml of glacial g (0.5242 m mol) of 4-[(1-(5-	Б		
10	amino-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid and 0.45 ml of 40% for- 0 malin in 2 ml of aetonitrile and 1 ml of absolute dimethyl formamide. After 1.5 hours the reaction mixture was evaporated <i>in vacuo</i> . The evaporation residue was dissolved in water by addition of hydrochloric acid at pH 2–3. After several extractions with chloroform the aqueous							
15	phase was adjusted to pH 6 to 7 by means of saturated sodium hydrogen carbonate solution and further extracted several times with chloroform. This organic extract was dried and filtered.  5 After evaporating in vacuo the evaporation residue was recrystallized from isopropanol. The colourless crystals were washed with absolute ether.  Yield: 0.09 g (42.8% of theory), M.p.: 185°C (decomp. from 175°C)							
20	Calc.: Found:	C 70.39 70.10	H 7.63 7.63	N 10.26 10.47		20		
25	0.10 g (0. zoic acid in	.262 m mol) of 4 1 ml of acetic and	-[(1-(5-amino hydride were	2-piperidino-phen stirred for 6 hours	by/methy/]benzoic acid by/)-ethy/)-aminocarbony/methy/]ben- bentalized at 20°C, then evaporated in vacuo, besidue was recrystallized from	25		
30	Yield: 0.08 ( M.p.: 241–2	g (72.7% of theo 243°C	ory),			30		
	Calc.: Found:	C 68.07 67.53	H 6.90 6.83	N 9.92 9.72	•			
35	Example 37 4-[(1-(5-Benzoylamino-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid 0.30 ml (2.62 m mol) of benzoyl chloride were added to a solution of 1 g (2.62 m mol) of 4- [(1-(5-amino-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid and 0.37 ml (2.62 m							
40	mol) of triethylamine in 10 ml of anhydrous dimethyl formamide. After stirring for 2 hours at 20–30°C, the reaction mixture was evaporated <i>in vacuo</i> and distributed between water and ethyl acetate. The organic phase was dried and filtered and evaporated <i>in vacuo</i> . The evaporation residue (1.12 g) was recrystallized from ethanol by addition of activated charcoal. Yield: 0.5 g (39.4% of theory), M.p.: 225–224°C							
45	Calc.: Found:	C 71.73 71.70	H 6.43 6.50	N 8.65 8.66		45		
50	Analogous	ly to Example 37	the following	compound was p		50		
	4-[(1-(5-Ethoxycarbonylamino-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid Yield: 34.2% of theory, M.p.: 220°C (decomp.)							
55	Calc.: Found:	C 66.21 65.97	H 6.89 6.83	N 9.26 9.57		55		
<b>60</b>	0.20 ml (0 of 4-[(1-(5-ardrous pyridin 4 hours at 2 evaporation s	0.262 m mol) of mino-2-piperidino ne. After the exot 0°C. Subsequent residue was distri	mesyl chloride phenyl)-ethyl hermic reaction ly the reaction buted at pH 2	e were added to a )-aminocarbonylm in was finished thi in mixture was eva 2–3 between wate	carbonylmethyl]benzoic acid solution of 0.10 g (0.262 m mol) sethyl]benzoic acid in 1 ml of anhy- e mixture was allowed to stand for porated in vacuo and the er and chloroform. The acidic	60		
65	aqueous pha	se was adjusted	to pH 6 to 7	oy means of sodiu	im hydrogen carbonate solution and	65		

5	extracted with chloroform. This chloroform extract was dried and filtered. The residue obtained after evaporating in vacuo was purified by column chromatography on silica gel (chloroform/methanol = 4:1).  Yield: 0.03 g (25% of theory),  M.p.: 210-220°C (decomp.) (ether)	5
	Calc.: mol peak . m/e = 459	
10	Example 39 4-[(1-(5-Acetoxy-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid 0.35 g (0.915 m mol) of 4-[(1-(5-hydroxy-2-piperidinophenyl)-ethyl)-aminocarbonylmethyl]- benzoic acid were heated together with 0.103 ml (1.098 m mol) of acetic anhydride on the steam bath and after standing for 4 days at 20°C, the reaction mixture was recrystallized from	10
15	methanol. Yield: 0.16 g (41.2% of theory), M.p.: 218–221°C	15
20	Calc.: C 67.91 H 6.65 N 6.60 Found: 67.70 6.95 6.55	20
25	Example 40 4-[(1-(5-Methoxy-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid methyl ester A solution of 60 mg (0,157 m mol) of 4-[(1-(5-hydroxy-2-piperidino-phenyl)-ethyl)-aminocar- bonylmethyl]benzoic acid in 1 ml of methanol ( + 1 drop of water) was added dropwise to an ethereal diazomethane solution, until no formation of gas took place. To destroy excess diazomethane 2N acetic acid was added. After evaporating in vacuo, the evaporation residue was distributed between toluene/ether and dilute sodium hydroxide solution. After drying,	25
30	filtering and evaporating the organic phase in vacuo, the evaporation residue was purified by column chromatography on silica gel (chloroform/methanol = 5:1).  Yield: 27% of theory, M.p.: Foam	30
35	Cal.: mol/peak $m/e = 410$ Found: $m/e = 410$	35
40	Example 41 4-[(1-(5-Benzyloxy-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester A solution of 0.50 g (1.218 m mol) of 4-[(1-(5-hydroxy-2-piperidino-pheyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester in 10 ml of anhydrous dimethyl formamide was quickly added to a suspension of 1.353 m mol of sodium hydride (32.5 mg of a 50% suspension in oil) in 2 ml of anhydrous dimethyl formamide. After stirring for 1.5 hours at 20°C, 0.16 ml (1.353 m mol) of benzyl bromide, dissolved in 2.3 ml of anhydrous dimethyl formamide, verifications.	40
45	and stirring was continued for 16 hours at 20°C. After evaporating in vacuo the residue was distributed between water and ether. The organic extract was dried, filtered and evaporated in vacuo. The evaporation residue was purified by column chromatography on silica gel (toluene/acetone = 10:1).  Yield: 0.34 g (55.5% of theory),	45
50	M.p.: 155–157°C (ether)  Calc.: C 74.37 H 7.25 N 5.60  Found: 74.11 7.41 5.39	50
55	Example 42 4-[(1-(5-Aminocarbonyl-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester 3.8 g (9.06 m mol) of 4-[(1-(5-cyano-2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid ethyl ester and 38 g of polyphosphoric acid were stirred for 2.5 hours at 80-90°C. Under	55
60	ice-cooling, water was added carefully and the reaction mixture was extracted with ethyl acetate and adjusted to alkaline by means of conc. ammonia. The organic phase was washed with water, dried and evaporated in vacuo. The evaporation residue was purified by column chromatography on silica gel (chloroform/methanol = 20/1).	60

48						GB 2 090 834A	48
		25.2% of theory 189°C (ethanol)					
5	Calc.: Found:	C 68.63 68.42	H 7.14 6.95	N 9.60 9.46		•	5
	Under refl 4-[(1-(5-cyar of absolute e evaporated i hydrogen ca filtered, and	oxycarbonyl-2-p lux, dried hydro no-2-piperidino- ethanol until aft in vacuo, mixed rbonate solution evaporated in v	gen chloride the phenyl)-ethyl)-ethyl)-er 4 hours no with water as a. The separate.	was introduced in -aminocarbonylm initrile could be not ether, and acted ether phase aporation residu	into a solution of methyl]benzoic detected. The djusted to alkali was extracted was extracted was purified	[]benzoic acid ethyl ester of 1.1 g (2.62 m mol) of acid ethyl ester in 22 ml reaction mixture was ne by means of sodium with water, dried and by column chromato-	10
15	Yield: 0.6 g	lica gel (methyl (49.2% of theo 138°C (ether)		acetonitrile/glad	cial acetic acid:	10:1:0.05).	15
20	Calc.: Found:	C 69.51 69.28	H 7.35 7.34	N 6.00 5.83			20
25	A solution ethyl)-amino pH = 2 by the conc. hydrod The reaction	Oxo-piperidino)-pof 2.9 g (6.86) carbonylmethyline addition of 2 chloric acid were mixture was evereidino).	m mol) of 4- ]benzoic acid !N hydrochlori e added and t /aporated <i>in v</i>	semihydrate in ic acid. After sti the mixture was vacuo, mixed wit	a-8-aza-spiro[4.] 40 ml of acetor rring for 6 hour allowed to star th water and et	5]decane-8-yl]phenyl)- ne was adjusted to s at 50°C 5 drops of nd for 16 hours at 20°C. hyl acetate and adjusted	25
30	combined or The evapora Yield: 1.9 g	ganic extracts v	vere washed v s recrystallized ory),		d, filtered, and	ethyl acetate, the evaporated <i>in vacuo</i> . er.	30
35	Calc.: Found:	C 69.46 69.75	H 6.36 6.33	N 7.36 7.29			35
40	0.244 g (solution of 1 benzoic acid	<i>lydroxy-piperidi</i> 5.92 m mol) of g (2.63 m mo in 20 ml of ab	sodium boro- l) of 4-[(1-(2-( solute ethano	hydride were ac 4-oxo-piperiding I. After stirring f	dded in portions o)-phenyl)-ethyl) for 1.5 hours at	oic acid × 0.66 H₂O s with stirring to a -aminocarbonylmethyl]- room temperature, the id, evaporated in vacuo,	40
45	mixed with visolution. After and the extra petroleum et Yield: 0.78	vater and ethyl er extracting se act was evapora ther. g (75% of theo	acetate, and a veral times wi ated <i>in vacuo</i> . ry),	adjusted to pH = th ethyl acetate	<ul><li>6 by means o</li><li>, the organic pl</li></ul>	f 2N sodium hydroxide nase was dried, filtered, ecrystallized from	45
50	•	180°C (decomp ( ×	o.) 0.66 H₂O) 66.72	C 66.97 6.62	H 6.81 N	1 7.10	50
55	0.94 g (5. 4-[(1-(2-pipe	eridino-phenyl)-e .80 m mol) of o eridino-phenyl)-e	arbonyl diimi thyl)-aminoca	rbonylmethyl]be	led to a solution enzoic acid in 2	by lester of 2 g (5.46 m mol) of mol of absolute tetrahytes excluding moisture.	55
60	Subsequenti for 18 hours	y, 1.64 ml (2.2 at 20°C and h	? m mol) of 1- eated for 8 ho	propanol were a ours to reflux ter	added, the reac mperature. Afte	tion mixture was stirred or evaporating in vacuo	60

the evaporation residue was purified by column chromatography on silica gel (toluene/acetone = 10:1).

	Yield: 1.3 M.p.: 150	g (58.3% of th 0-151°C (ethyl	ieory), acetate)	•		
5	Calc.: Found:	C 73.51 73.70	Н 7.90 7.78	N 6.86 6.92		5
	Analoge	ously to Example	e 46 the following	ng compour	ds were prepared:	
10	Yield: 45	iperidino-pheny % of theory, 1–143°C (ether)		rbonylmeth	yl]benzoic acid isopropyl ester	10
15	Calc.: Found:	C 73.51 73.20	H 7.90 7.79	N 6.86 6.70		15
	Yield: 49' M.p.: 14	Piperidino-pheny % of theory, B°C (ether/tolue		rbonylmeth	yl]benzoic acid butyl ester	20
20	Calc.: Found:	C 73.90 74.10	H 8.11 7.99	N 6.63 6.70		
25	Yield: 41	Chloro-2-piperidi % of theory, 0–133°C (ether)		-aminocarb	onylmethyl]benzoic acid ethyl ester	25
30	Calc.: Found:	C 67.21 66.90	H 6.81 6.65	CI 8.26 8.32	N 6.53 6.67	30
35	4-[(1-(5-0 Yield: 30 M.p.: 11	.7% of theory,	no-phenyl)-ethyl)	-aminocarb	onylmethyl]benzoic acid butyl ester	35
	Calc.: Found:	C 68.33 68.20	H 7.27 7.23	CI 7.75 7.68	N 6.12 5.95	
40	4-[(1-(5-0	Chloro-2-piperidi 5 of theory,	no-phenyl)-ethyl	)-aminocarb	onylmethyl]benzoic acid tert.butyl ester	40
45	Calc.: Found:	mol peak	m/e = 456/8 m/e = 456/8			45
50	Yield: 56 M.p.: 15	Piperidino-pheny % of theory, 5-157°C (ethyl		arbonylmeth	yl]benzoic acid-(2-methoxyethyl ester)	50
30	Calc.: Found:	C 70.74 70.55		N 6.60 6.47		
55	<i>yl)-methy</i> Yield: 30	Piperidino-pheny v]]ester v.5% of theory, 0–112°C (ether		arbonylmeth	yl]benzoic acid]-(2,2-dimethyl-dioxolane-4-	55
60	Calc.: Found:	C 69.98 69.80			m/e = 480 m/e = 480	60

5	4-[(1-(2-Piperid Yield: 73.7% o M.p.: 126-128 Calc.: Found:	of theory,	•	nbonylmethyl N 6.14 6.03	]benzoic acid benzyl ester	5		
10	4-[(1-(2-Piperid After addition temperature for Yield: 71.4% o	lino-phenyl)-en n of 10 equiver 17 hours. of theory,	thyl)-aminoca alents of ethy	arbonylmethy	[]benzoic acid-(2-hydroxy-ethyl)-ester he reaction mixture was heated to reflux	10		
15	M.p.: 128-129 Calc.: Found:	9°C (ethyl ace C 70.21 70.14	H 7.36 7.42	N 6.82 6.70	m/e = 410 m/e = 410	15		
20	1,2-Bis[4-[(1-(2 After addition temperature for Yield: 43.5% o	n of 0.5 equiv 17 hours.	<i>henyl)-ethyl)-</i> valents of eth	aminocarbon ylene glycol	ylmethyl]benzoyloxy]ethane the reaction mixture was heated to reflux	20		
25	M.p.: 188-19 <sup>o</sup> Calc.: Found:	1°C (toluene) C 72.80 72.85	H 7.17 7.07	N 7.38 7.37	m/e = 758 m/e = 758	25		
30	<i>4[(1-(2-Piperidi</i> Yield: 56.7% o M.p.: 99–101°	f theory,	• •	rbonylmethylj	benzoic acid-(2-diethylamino-ethyl)-ester	30		
35	Calc.: Found:	C 72.23 72.40	H 8.44 8.37	N 9.03 8.95		35		
40	4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid-2-(1,3-dimethyl-xanthine-7-yl)-ethyl ester  O As solvent absolute pyridine was used. After addition of 1 equivalent of 7-(2-hydroxy-ethyl)-theophylline and after addition of a little piece of metallic sodium the reaction mixture was stirred for 4 hours in the bath of 130°C.  Yield: 40.9% of theory,							
45	M.p.: 121–123 Calc.: Found:	C 65.01 64.78	H 6.34 6.38	N 14.68 14.90	m/e = 572 m/e = 572	45		
50	A mixture of	2 g (5.46 m	mol) of 4-[(1	-(2-piperiding	[]benzoic.acid methyl ester -phenyl)-ethyl)-aminocarbonylmethyl]ben-	50		
	dichloroethane and extracted w	was refluxed vith diluted so do, filtered, an stography on st44.8% of the	for 24 hours odium hydrog nd evaporate silica gel (toli	, then evapor gen carbonate d <i>in vacuo</i> . T	ric acid, and 1.65 ml of 1,2-rated in vacuo, dissolved in chloroform, a solution. The organic phase was washed he evaporation residue was purified by e = 5:1).	55		
60	Calc.: Found:	C 72.60 72.19	H 7.42 7.33	N 7.36 7.01		60		

5	Example 48 4-[(2-(2-Piperidino-phenyl)-2-propyl)-aminocarbonylmethyl]benzoic acid ethyl ester 0.20 g (0.526 m mol) of 4-[(2-(2-piperidino-phenyl)-2-propyl)-aminocarbonylmethyl]benzoic acid and 2 ml of 4N ethanolic hydrochloric acid were stirred at 20°C. After 36 hours, the reaction mixture was evaporated in vacuo, and the evaporation residue was distributed between water (at pH = 8 by addition of ammonia (10%)) and ethyl acetate. The organic phase was washed with water, dried, filtered, and evaporated in vacuo. The evaporation residue was	5
10	purified by column chromatography on silica gel (toluene/acetone = 10:1). Yield: 0.079 g (36.7% of theory), M.p.: 151-153°C (ether)	10
15	Calc.: C 73.50 H 7.90 N 6.86 Found: 73.40 7.95 6.96	15
15	Example 49 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid tert.butyl ester A mixture of 3.60 g (17.4 m mol) of N,N'-dicyclohexylcarbodiimide, 1.9 ml (20.4 m mol) of tert.butanol and 0.036 g (0.36 m mol) of copper(I)chloride was stirred for 3 days at room temperature, then 12 ml of methylene chloride were added, and the solution thus obtained was	20
20	added to a solution of 2 g (5.46 m mol) of 4-[(1-(2-piperidino-phenyi)-ethyi)-aminocarbonyime-thyi]benzoic acid in 80 ml of methylene chloride. After stirring for 16 hours at 20°C, the	
25	chloride solution was evaporated <i>in vacuo</i> . The evaporation residue was purified by column chromatography on silica gel (toluene/acetone = 15:1).  Yield: 0.45 g (19.7% of theory),  M.p.: 125-127°C (ether)	25
30	Calc.: C 73.90 H 8.11 N 6.63 Found: 74.20 8.09 6.77	30
35	Example 50 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid 2-(nicotinoyloxy)-ethyl ester A solution of 0.16 g (1.13 m mol) of nicotinic acid chloride in 5 ml of methylene chloride was quickly added to a solution of 0.45 g (1.10 m mol) of 4-[(1-(2-piperidino-phenyl)-ethyl)- aminocarbonylmethyl]benzoic acid (2-hydroxy-ethyl)-ester and 0.16 m mol) of triethylamine in 10 ml of methylene chloride. After stirring for 4 horide celution was filtered and evaporated in	35
40	extracted with water, dried, and the methylene chloride solution was filtered and evaporated in vacuo. The evaporation residue was purified by column chromatography on silica gel (chlorofor-m/acetone = 3:1).  Yield: 0.34 g (60% of theory), M.p.: 103-105°C (ether)	40
45	Calc.: C 69.88 H 6.45 N 8.15 Found: 70.13 6.55 8.13	45
50	Example 51 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzamide 2.3 g (0.0142 mol) of carbonyl diimidazole were given to 4.76 g (0.013 mol) of 4-[(1-(2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid in 60 ml of absolute pyridine and the mixture was subsequently heated for 45 minutes to 50°C. After cooling in a carbon dioxide/methanol bath 7 ml of liquid ammonia were added and heated for 20 hours to 80°C in	50
55	an autoclave. Subsequently the reaction mixture was cooled and evaporated in vacuo. The residue was dissolved in 50 ml of hot methanol, 200 ml of water were added and the mixture was allowed to rest over-night. The crystalline precipitate was suction filtered and recrystallized from methanol by addition of activated charcoal. Yield: 3.5 g (73.6% of theory), M.p.: 197–199°C	55
60	Calc.: C 72.30 H 7.45 N 11.50 Found: 72.30 7.45 11.32	60
65	Example 52 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]-N-methylbenzamide 2 g (5.46 m mol) of 4-[(1-(2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid and	65

5	0.94 g (5.80 m mol) of carbonyl diimidazole in 20 ml of absolute pyridine were heated to reflux temperature for 1 hour. Subsequently, 0.41 g (6.07 m mol) of methylamine hydrochloride were added and the mixture was stirred for 1 hour at 20°C and refluxed for 2 hours. After evaporating in vacuo, the residue was distributed between water and methylene chloride; the organic extract was dried, filtered, and evaporated in vacuo. The evaporation residue was purified by column chromatography on silica gel (chloroform/methanol/conc. ammonia = 10:1:0.05).  Yield: 1.7 g (82% of theory),								
10	M.p.: 218-220 Calc.: Found:	*C (isopropand C 72.77 72.88		N 11.07 10.91		10			
15	• ,	no-phenyl)-eth theory,	yl)-aminocarb		was prepared: I-N,N-dimethyl-benzamide	15			
20	Calc.: Found:	C 73.26 73.60	H 7.94 7.85	N 10.68 10.73		20			
25	Example 53 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]-N-butyl-benzamide 0.94 g (5.80 m mol) of carbonyl diimidazole were added to the solution of 2 g (5.46 m mol) of 4-[(1-(2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]benzoic acid in 20 ml of absolute tetrahydrofuran. The mixture was heated to reflux temperature for 30 minutes, 0.44 g (6.1 m mol) of 1-butylamine were added, and the reaction mixture was again refluxed for 2 hours. After								
30	evaporating in v gel (chloroform/ Yield: 1.65 g (7 M.p.: 178–181	<i>acuo,</i> the evap acetone:6:1). 1.7% of theor	ooration resid y),	ue was puri	fied by column chromatography on silica	30			
35	Calc.: Found:	C 74.09 74.34	H 8.37 8.26	N 9.97 9.95		35			
	Analogously to	o Example 53	the following	compound	s were obtained:				
40	4-[(1-(2-Piperidi Yield: 73.8% of M.p.: 131-133	theory,	yl)-aminocart	onylmethyl <u></u>	]benzoic acid piperidide	40			
45	Calc.: Found:	C 74.79 75.13	H 8.14 7.99	N 9.69 9.48	m/e = 433 m/e = 433	45 .			
F^	4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]-benzoic acid morpholide Yield: 60.5% of theory, M.p.: 148-150°C (ethyl acetate/ether)								
50	Calc.: Found:	C 71.69 71.60	H 7.64 7.80	N 9.65 9.57	•	50			
	at room tempera aminocarbonylm mixture was stirr	mol) of p-tolue ature to a mixt nethyl]benzam red for 15 mir	ne-sulfonic ac ure of 2.19 g ide and 1.07 nutes at 20°C	cid chloride g (6 m mol) g (13.5 m and then fo	were added in two portions whilst stirring of 4-[(1-(2-piperidinophenyl)-ethyl)-mol) of absolute pyridine. The reaction or 2 hours at 50°C. After cooling, water	55 ·			
60	eans of conc. ammonia, and extracted cts were washed with water, dried over evaporation residue was purified by I acetete = 4:1).	60							

=	2

53						GB 2 C	090834A —————	53 ——
	Yield: 1.15 g (55.3% of M.p.: 155–157°C (ethy	theory) acetate	, )					
5	Calc.: C 76.05 Found: 76.30		7.25 7.07	1	N 12.09 11.90			5
	Example A Tablets containing 5 mg	of 4-[(1	1-(2-pip	eridi	no-phen	/l]ethyl)-aminocarbonylmethyl]be	nzoic acid	
10	Composition:						•	10
	1 tablet contains: Active ingredient Corn starch	(1) (2)	5.0 62.0	mg				
15	Lactose Polyvinyl pyrrolidone Magnesium stearate	(3) (4) (5)		mg mg mg				15
			120.0	) mg	_	•		
	_			Ŭ				20
20	through a screen of mes	h size 1	.5 mm 1 0 mn	and n me	dried at sh size a	ater. The moist mixture was grar approx. 45°C. The dry granulate nd mixed with 5. The finished m f 7 mm diameter and an unilater	nixture was	
25	Weight of tablet: 120 m	ig Bulets p	1033 W	ui pi		•		25
	Example B Coated tablets containing zoic acid	g 2.5 n	ng of 4	-[(1-(	2-piperio	lino-phenyl)-ethyl)-aminocarbony	lmethyl]ben-	30
30	1 coated tablet core con	tains:						
	Active ingredient Potato starch	(1) (2) (3)	2.5 44.0 30.0	mġ				
35	Lactose Polyvinyl pyrrolidone Magnesium stearate	(4) (5)	3.0 0.5	mg				35
	•		80.0	mg				
40	Method of preparation:							40
	1, 2, 3, and 4 were rethrough a screen of metagranulated through the	sh size 1 same so nresseo	reen. A I on a t	dried After sables	at appr adding o	ith water. The moist mass was gox. 45°C and the granulate was of 5, curvatured coated tablet cong machine. The coated tablet congrated tablet	res of a pres thus	45
45	wore covered	in conv	entions	al ma	nner wil	h a coating, which essentially co polished with wax. Weight of co	1121212 01	40
	Example C							50
50	Tablets containing 10 r. Composition: 1 tablet contains:	ng of 4-	[(1-(2-p	oiper	idino-ph	nyl)-ethyl)-aminocarbonylmethyl	]benzoic acid	•
	Active ingredient				.0 mg			55
55					.0 mg .0 mg			00
	Corn starch Polyvinyl pyrrolidone				.0 mg			
	Magnesium stearate			1	.0 mg			
60		•		120	0.0 mg			60
	at nobality i pyrrolidone	ingredi in wate	er. The	mois	st mass v	n starch was moistened with a 2 was granulated through a screen	AAIfii a iiiooii	05
65	size of 1.5 mm and dri	ed at 45	5°C. Th	e dri	ed grani	late was granulated through a s	creen of 1	65

mm mesh size and homogeneously mixed with magnesium stearate.

Weight of tablets:

120 mg

Punch:

7 mm & with a notch.

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5

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Example D

Coated tablets containing 5 mg of 4-[(1-(2-piperidino-phenyl)-ethyl)-aminocarbonylmethyl]ben-

zoic acid

10 1 coated tablet core contains:

5.0 mg 70.0 mg

Active ingredient Calcium phosphate secondary

50.0 mg

Corn starch Polyvinyl pyrrolidone

4.0 mg 1.0 mg

Magnesium stearate

15

20

130.0 mg

Method of preparation:

The mixture, consisting of the active ingredient, the calcium phosphate and the corn starch, 20 was moistened with a 15% solution of polyvinyl pyrrolidone in water. The moist mass was granulated through a screen of 1 mm mesh size, dried at 45°C and again passed through the same screen. The granulate was mixed with the above mentioned amount of magnesium stearate and the mixture thus obtained was pressed into coated tablet cores.

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Weight of core: Punch:

130 mg 7 mm φ

The thus prepared coated tablet cores were covered according to conventional manner with a 30 layer consisting of sugar and talcum. The finished coated tablets were polished with wax. Weight of coated tablet: 180 mg.

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**CLAIMS** 

1. Compounds of general formula I

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$$\begin{array}{c}
R_{40} \\
R_{3} \\
R_{1} \\
R_{2}
\end{array}$$

$$\begin{array}{c}
R_{1} \\
R_{2}
\end{array}$$

$$\begin{array}{c}
R_{1} \\
R_{2}
\end{array}$$

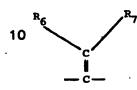
$$\begin{array}{c}
R_{1} \\
R_{2}
\end{array}$$

45 [wherein R<sub>1</sub> and R<sub>2</sub>, which may be the same or different, each represents an alkyl group containing 1 to 6 carbon atoms or a cycloalkyl group containing 5 to 7 carbon atoms, or R1 and R2 together with the nitrogen atom to which they are attached represent an unbranched alkyleneimino group containing 3 to 6 carbon atoms optionally substituted by 1 or 2 alkyl 50 groups, each containing 1 to 3 carbon atoms, or by a hydroxy group and in which a methylene group may optionally be replaced by a carbonyl group, by an oxygen or sulfur atom or by an imino group (which may optionally be substituted by an alkyl group containing 1 to 3 carbon atom, an aralkyl group containing 7 to 10 carbon atoms or by a phenyl or halophenyl group) or an ethylene group may optionally be replaced by an O-phenylene group; and unbranched 55 alkenyleneimino group containing 4 to 6 carbon atoms; a saturated or partly unsaturated 55 azabicycloalkyl group containing 6 to 10 carbon atoms; an aza-1,4-dioxaspiro-alkyl group containing 6 to 8 carbon atoms; or a heptamethyleneimino, octamethyleneimino, nonamethyleneimino or decamethyleneimino group; R<sub>3</sub> represents a hydrogen or halogen atom, a trifluoromethyl, alkyl, hydroxy, alkoxy, alkanoyloxy, mercapto, alkylmercapto, nitro, amino, cyano, alkanoyl, 60 carboxy, alkoxycarbonyl, aminocarbonyl, alkylaminocarbonyl, dialkylaminocarbonyl, aminosulfo-60 nyl, alkylamino, dialkylamino, alkanoylamino, alkoxycarbonylamino or alkylsulfonylamino group (wherein each alkyl part in the above mentioned groups may contain from 1 to 3 carbon atoms), an aralkoxy group containing 7 to 10 carbon atoms or an arylcarbonylamino group; R4 represents a hydrogen atom or an alkyl group containing 1 to 3 carbon atoms; Rs represents a

65 hydrogen atom, a halogen atom or an alkyl group containing 1 to 3 carbon atoms; A represents

a bond, a methylene or ethylene group optionally substituted by an alkyl group containing 1 to 5 carbon atoms, a methylene or ethylene group substituted by two alkyl groups each containing 1 to 3 carbon atoms, a methylene group substituted by a cycloalkyl group containing 3 to 7 carbon atoms or by a hydroxyalkyl, alkoxyalkyl, cyano, carboxyl, alkoxycarbonyl, aminocarbonyl, 5 alkylaminocarbonyl, dialkylaminocarbonyl, aryl or aralkyl group, wherein each of the alkyl parts may contain from 1 to 3 carbon atoms, a cycloalkylidene group containing 3 to 7 carbon atoms or a vinylidene group of formula

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15 wherein R<sub>8</sub> and R<sub>7</sub>, which may be the same or different, each represents a hydrogen atom or an alkyl group containing 1 to 3 carbon atoms or one of the radicals R<sub>8</sub> and R<sub>7</sub> represents a cycloalkyl group containing 3 to 7 carbon atoms or an aryl or aralkyl group and the other is as defined above, or R<sub>8</sub> and R<sub>7</sub> together with the carbon atom to which they are attached represent a cycloalkylidene radical containing 5 to 7 carbon atoms; B represents a methylene or ethylene 20 group optionally substituted by an alkyl group 1 to 3 carbon atoms; and W represents a 20 hydrogen or halogen atom, a nitro group, an amino group (optionally substituted by an alkanoyl group containing 1 to 3 carbon atoms) an alkyl group containing 1 to 3 carbon atoms (optionally substituted by a hydroxy or carboxy group or bt one or two alkoxycarbonyl groups containing 2 to 4 carbon atoms each), an alkenyl group containing 2 to 5 carbon atoms 25 substituted by a carboxy or alkoxycarbonyl group containing 2 to 4 carbon atoms, an alkanoyl 25 group containing 1 to 3 carbon atoms, a dialkoxymethyl or trialkoxymethyl group containing 1 to 3 carbon atoms in each alkyl part, an alkylenedioxymethyl group containing 2 or 3 carbon atoms in the alkylene part, a 1,3-oxazoline-2-yl or cyano group, an aminocarbonyl group (optionally substituted by one or two alkyl groups containing 1 to 4 carbon atoms in each alkyl 30 part), an unbranched alkyleneiminocarbonyl group containing 5 to 8 carbon atoms, a morpholi-30 nocarbonyl group, a (dialkyldioxolane-yl)-alkoxycarbonyl group containing 7 to 10 carbon atoms or a carboxy group or esterfied carboxy group wherein if the said ester group consists of an alkyl group containing 1 to 6 carbon atoms this may be substituted, in any but the  $\alpha$ -position, by a hydroxy, alkoxy, amino, alkylamino, dialkylamino, 1,3-dimethylxanthine-7-yl, alkanolyoxy, aroy-35 loxy, aralkanoyloxy or pyridine-carbonyloxy group or by two hydroxy groups—except in the case 35 of any methyl or methylene group in the above cases, which can only be substituted by one

hydroxy group or by a group of formula

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45 wherein A, B,  $R_1$ ,  $R_2$ ,  $R_3$ ,  $R_4$  and  $R_5$  are as hereinbetore defined whereby each alkyl part of the 45 above alkyl ester substituted may contain from 1 to 3 carbon atoms], and salts thereof. 2. Physiologically compatible salts, formed with inorganic or organic acids or bases, of compounds of general formula I as claimed in claim 1.

3. Compounds as claimed in claim 1 or claim 2, wherein  $R_1$  and  $R_2$  together with the 50 nitrogen atom to which they are attached, represent a dialkylamino or N-alkylcyclohexylamino 50 group (wherein each alkyl part may contain from 1 to 4 carbon atoms), an unbranched alkyleneimino group containing 3 to 6 carbon atoms (optionally substituted by one or two methyl groups), a hydroxypiperidino, piperidone-1-yl, tetrahydro-pyridino, morpholino, thiomorpholino, N-methyl-piperazino, N-benzyl-piperzino, N-chlorophenyl-piperazino, heptamethyleneim-55 ino or octamethyleneimino group, a saturated or partly unsaturated azabicycloalkyl group

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containing 7 to 9 carbon atoms, an unbranched alkyleneimino group containing 4 to 6 carbon atoms (wherein an ethylene group is replaced by an o-phenylene group), or a 1,4-dioxa-azaspiroalkyl group containing 7 to 8 carbon atoms; R3 represents a hydrogen, fluorine, chlorine, bromine, or iodine atom or a methyl, trifluoromethyl, hydroxy, methoxy, benzyoxy, acetoxy, 60 mercapto, methylmercapto, nitro, amino, dimethylamino, acetylamino, methylsulfonylamino, benzoylamino, ethoxycarbonylamino, cyano, carboxy, methoxycarbonyl, ethoxycarbonyl, aminocarbonyl, acetyl or aminosulfonyl group; R4 represents a hydrogen atom or a methyl group; R5 represents a hydrogen atom, a chlorine atom or a methyl group; A represents a bond, a

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methylene group optionally substituted by an alkyl group containing 1 to 3 carbon atoms, a 65 phenyl, cyclohexyl, carboxy, methoxycarbonyl or hydroxymethyl group, a dimethylmethylene,

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cyclopropylidene or ethylene group or a vinylidene group of formula



wherein R<sub>8</sub> and R<sub>7</sub>, which may be the same or different, each represents a hydrogen atom or a 10 methyl group or R<sub>s</sub> and R<sub>7</sub> together with the carbon atom to which they are attached represent a 10 cycloalkylidene radical containing 5 or 6 carbon atoms; B represents a methylene, ethylidene or ethylene group; and W represents a hydrogen atom, a methyl, ethyl, hydroxymethyl, cyano or carboxyvinylene group, an alkyl group containing 1 to 3 carbon atoms substituted by a carboxy group or by one or two alkoxycarbonyl groups containing 2 to 4 carbon atoms each, a carbonyl 15 group (substituted by a hydrogen atom, a methyl, ethyl, hydroxy, alkoxy, (2,2-dimethyldioxolane-4-yl)-methoxy, benzyloxy, pyridylmethyoxy, amino, alkylamino, dialkylamino, piperidino or morpholino group, each alkyl part in the above groups containing from 1 to 3 carbon atoms) or

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a group of formula

wherein n is 2, 3, or 4, and R<sub>8</sub> represents a hydroxy, methoxy, ethoxy, acetoxy, benzoyloxy, or 25 pyridinecarbonyloxy group, a dialkylamino group containing 1 to 3 carbon atoms in each alkyl part, a 1,3-dimethylxanthine-7-yl group, or a group of formula

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35 wherein A, B and R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub> and R<sub>5</sub> are as defined above. 4. Compounds as claimed in claim 3, wherein the radical

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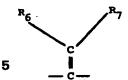
is present in the 2-position and the radical W is present in the 4'-position. 5. Compounds of general formula I a

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$$\begin{array}{c|c}
R_{3} & & \\
\hline
 & \\
R_{2}
\end{array}$$

55 wherein R<sub>1</sub> and R<sub>2</sub> together with the nitrogen atom to which they are attached, represent a 55 dimethylamino, pyrrolidino, methylpyrrolidino, piperidino, methylpiperidino, dimethylpiperidino, tetrahydro-pyridino, 2-octahydro-isoindolo, or hexamethyleneimino group, R<sub>3</sub> represents a hydrogen, fluorine or chlorine atom or a methyl group, A represents a methylene group (optionally substituted by a cyclohexyl, phenyl, methoxycarbonyl or ethoxycarbonyl group or an alkyl group 60 containing 1 to 3 carbon atoms), a dimethylmethylene group or a vinylidene group of formula 60



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wherein R<sub>8</sub> and R<sub>7</sub> each represents a hydrogen atom or together with the carbon atom to which they are attached represent a cyclohexylidene group, and W represents a methyl, hydroxymethyl 10 or carboxymethyl group, a carbonyl group (substituted by a hydrogen atom or by a methyl, hydroxy, methoxy, ethoxy, propoxy, isopropoxy, 2-hydroxyethoxy, 2-methoxyethoxy, (2,2-dimethyl-dioxolane-4-yl)-methoxy or 2-diethylaminoethoxy group) and salts thereof.

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6. Compounds as claimed in claim 5 wherein R<sub>1</sub> and R<sub>2</sub> together with the nitrogen atom to which they are attached, represent a pyrrolidino, piperidino, methylpiperidino, hexamethyleneim-15 ino, tetrahydro-pyridino or 2-octahydro-isoindolo group, R<sub>3</sub> represents a hydrogen, fluorine or chlorine atom or a methyl group, A represents a methylene group (optionally substituted by a methyl, isopropyl, phenyl or methoxycarbonyl group) or a dimethyl-methylene or vinylidene group and W represents a methyl, hydroxymethyl, carboxymethyl, formyl or carboxy group or an alkoxycarbonyl group optionally substituted by a (2,2-dimethyl-dioxolane-4-yl) group, wherein 20 the alkoxy group may contain from 1 to 3 carbon atoms.

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7. 4-[(1-(2-Piperidino-phenyl)-ethyl)-aminocarbonylmethyl]-benzoic acid.

8. 4-[(2-Piperidino-benzhydrile)-aminocarbonylmethyl]benzoic acid.

C<sub>1-3</sub> alkyl esters of compounds as claimed in claim 7 or claim 8. 10. Physiologically compatible salts of compounds as claimed in any one of claims 7 to 9

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25 formed with organic or inorganic acids or bases. 11. Compounds as claimed in claim 1 wherein R<sub>1</sub> and R<sub>2</sub>, which may be the same or different, each represents an alkyl group containing 1 to 6 carbon atoms or a cycloalkyl group containing 5 to 7 carbon atoms, or R1 and R2 together with the nitrogen atom to which they are attached, represent an alkyleneimino group containing 4 to 10 carbon atoms in the alkylene ring 30 (optionally substituted by one or two alkyl groups each containing 1 to 3 carbon atoms), a morpholino or a thiomorpholino group, R<sub>3</sub> represents a hydrogen or a halogen atom, a trifluoro-

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methyl, alkyl, hydroxy, alkoxy, mercapto, alkylmercapto, cyano, nitro, amino, aminocarbonyl, alkylamino, dialkylamino, or alkylsulfonylamino group, whereby each alkyl part in the above mentioned groups may contain from 1 to 3 carbon atoms, A represents a methylene or ethylene 35 group optionally substituted by one or two alkyl groups each containing 1 to 3 carbon atoms, R4 and  $R_s$  each represent a hydrogen atom, B is as defined in claim 1, and W, which is in the para position, represents a carboxy group and its esters.

12. Compounds as claimed in claim 1 as herein described in ay one of the examples. 13. Compounds as claimed in claim 11, as herein described in any one of Examples 1, 8,

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40 24, 29-31, 35, 36, 38, 40 or 48. 14. A process for the preparation of compounds as claimed in claim 1, which comprises reacting an amine of general formula II

45 45 ,(II) 50

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wherein A, R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub> and R<sub>4</sub> are as defined in claim 1 (or if A represents one of the above mentioned vinylidene groups one of its tautomers, or a lithium or magnesium-halide complex thereof) with a carboxylic acid of general formula III

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wherein  $R_{\delta}$  and B are as defined in claim 1 and W' represents W as defined in claim 1 or represents a carboxyl group protected by a protective radical, or with reactive derivatives thereof, optionally prepared in the reaction mixture, and if necessary cleaving off a protective 65 radical.

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- 15. A process as claimed in claim 14, wherein the reaction is carried in a solvent at temperatures between -25 and 250°C.
- 16. A process as claimed in claim 14 or claim 15 wherein the reaction is carried out in the presence of an acid-activating or dehydrating agent.
- 17. A process as claimed in claim 14 or claim 15 wherein the reaction is carried out in the presence of an amine-activating agent.
- 18. A process as claimed in any one of claims 14 to 17 wherein the reaction is carried out in the presence of an inorganic or tertiary organic base.
- 19. A process as claimed in any of claims 14 to 18 wherein the water formed during the 10 reaction is removed by azeotropic distillation or by addition of a drying agent.
  - 20. A process for the preparation of compounds of general formula I as claimed in claim 1, wherein W represents a carboxy group, which comprises hydrolytically, thermolytically or hydrogenolytically reacting a compound of general formula IV

15
$$R_{3}$$
 $R_{1}$ 
 $R_{1}$ 
 $R_{2}$ 
 $R_{3}$ 
 $R_{1}$ 
 $R_{2}$ 
 $R_{3}$ 
 $R_{1}$ 
 $R_{2}$ 
 $R_{3}$ 
 $R_{3}$ 
 $R_{4}$ 
 $R_{5}$ 
 $R_{5}$ 

- wherein R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub>, R<sub>5</sub>, A and B are as defined in claim 1 and D represents a group being 25 transformable into a carboxy group by means of hydrolysis, thermolysis or hydrogenolysis.
- 21. A process as claimed in claim 20 wherein the reaction is carried out in a solvent at temperatures between room temperature and the boiling temperature of the reaction mixture.
  - 22. A process as claimed in claim 20 or claim 21 wherein the hydrolysis or thermolysis is carried out in the presence of an acid or base.
- 30 23. A process as claimed in claim 20 or claim 21 wherein, if in the compound of formula IV 30 D represents a nitrile or aminocarbonyl group, the reaction is carried out in the presence of a nitrite and an acid.
  - 24. A process as claimed in claim 23 wherein the nitrite is sodium nitrite and the acid is sulfuric acid.
- 35 25. A process for the preparation of compounds as claimed in claim 1, which comprises alkylating a compound (optionally formed in the reaction mixture) of general formula V

- wherein R<sub>3</sub>, R<sub>4</sub>, R<sub>5</sub>, A, B and W are as defined in claim 1 and R<sub>2</sub>' represents a hydrogen atom or as defined in claim 1, with a compound of general formula VI
- 50 R<sub>1</sub>'-E, (VI) 50
- wherein R<sub>1</sub>' represents R<sub>1</sub> as defined in claim 1 or together with the radical R<sub>2</sub>' in the above compound of formula V represents a straight-chained alkylene group containing 4 to 6 carbon atoms (optionally substituted by one or two alkyl groups containing 1 to 3 carbon atoms) or an n-pentylene group wherein the third methylene group is replaced by an oxygen or sulfur atoms, and E represents a nucleophillically exchangeable group or (if in the radical R<sub>1</sub>' a methylene
- group is replaced by an aldehyde or ketone carbonyl group) a hydrogen atom, if necessary in the presence of a reducing agent and optionally subsequently hydrolyzing.

  26. A process as claimed in claim 25 wherein the reaction is carried out in a solvent at
- 60 temperatures between 0 and 150°C.

  27. A process as claimed in claim 25 or claim 26 wherein the reaction is carried out in the
  - presence of an inorganic or tertiary organic base.

    28. A process as claimed in claim 25 or claim 26 wherein the alkylation is carried out with a carbonyl compound in the presence of a hydride at pH 7.
- 65 29. A process as claimed in claim 28 wherein the hydride is sodium cyanoborohydride.

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11.3

- 15. A process as claimed in claim 14, wherein the reaction is carried in a solvent at temperatures between -25 and 250°C.
- 16. A process as claimed in claim 14 or claim 15 wherein the reaction is carried out in the presence of an acid-activating or dehydrating agent.
- 17. A process as claimed in claim 14 or claim 15 wherein the reaction is carried out in the presence of an amine-activating agent.
- 18. A process as claimed in any one of claims 14 to 17 wherein the reaction is carried out in the presence of an inorganic or tertiary organic base.
- 19. A process as claimed in any of claims 14 to 18 wherein the water formed during the 10 reaction is removed by azeotropic distillation or by addition of a drying agent.
  - 20. A process for the preparation of compounds of general formula I as claimed in claim 1, wherein W represents a carboxy group, which comprises hydrolytically, thermolytically or hydrogenolytically reacting a compound of general formula IV

15
$$R_{3}$$
 $R_{1}$ 
 $R_{1}$ 
 $R_{2}$ 
 $R_{3}$ 
 $R_{1}$ 
 $R_{2}$ 
 $R_{3}$ 
 $R_{1}$ 
 $R_{2}$ 
 $R_{3}$ 
 $R_{3}$ 
 $R_{4}$ 
 $R_{5}$ 
 $R_{5}$ 
 $R_{5}$ 
 $R_{5}$ 
 $R_{5}$ 
 $R_{5}$ 
 $R_{5}$ 
 $R_{5}$ 
 $R_{5}$ 

- wherein R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub>, R<sub>5</sub>, A and B are as defined in claim 1 and D represents a group being 25 transformable into a carboxy group by means of hydrolysis, thermolysis or hydrogenolysis.
  - 21. A process as claimed in claim 20 wherein the reaction is carried out in a solvent at temperatures between room temperature and the boiling temperature of the reaction mixture.
  - 22. A process as claimed in claim 20 or claim 21 wherein the hydrolysis or thermolysis is carried out in the presence of an acid or base.
- 30 23. A process as claimed in claim 20 or claim 21 wherein, if in the compound of formula IV 30 D represents a nitrile or aminocarbonyl group, the reaction is carried out in the presence of a nitrite and an acid.
  - 24. A process as claimed in claim 23 wherein the nitrite is sodium nitrite and the acid is sulfuric acid.
- 35 25. A process for the preparation of compounds as claimed in claim 1, which comprises alkylating a compound (optionally formed in the reaction mixture) of general formula V

wherein R<sub>3</sub>, R<sub>4</sub>, R<sub>5</sub>, A, B and W are as defined in claim 1 and R<sub>2</sub>' represents a hydrogen atom or as defined in claim 1, with a compound of general formula VI

wherein R<sub>1</sub>' represents R<sub>1</sub> as defined in claim 1 or together with the radical R<sub>2</sub>' in the above compound of formula V represents a straight-chained alkylene group containing 4 to 6 carbon atoms (optionally substituted by one or two alkyl groups containing 1 to 3 carbon atoms) or an 55 n-pentylene group wherein the third methylene group is replaced by an oxygen or sulfur atoms, and E represents a nucleophillically exchangeable group or (if in the radical R<sub>1</sub>' a methylene group is replaced by an aldehyde or ketone carbonyl group) a hydrogen atom, if necessary in the presence of a reducing agent and optionally subsequently hydrolyzing.

- 26. A process as claimed in claim 25 wherein the reaction is carried out in a solvent at 60 temperatures between 0 and 150°C.
  - 27. A process as claimed in claim 25 or claim 26 wherein the reaction is carried out in the presence of an inorganic or tertiary organic base.
  - 28. A process as claimed in claim 25 or claim 26 wherein the alkylation is carried out with a carbonyl compound in the presence of a hydride at pH 7.
- 65 29. A process as claimed in claim 28 wherein the hydride is sodium cyanoborohydride.

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30. A process as claimed in claim 25 or claim 26 wherein a methylation reaction is carried out using formaldehyde in the presence of formic acid, or hydrogen in the presence of a hydrogenation catalyst.

31. A process for the preparation of compounds of general formula I, wherein W represents a carboxy group, an alkanoyl group, an alkanoyl group containing 1 to 3 carbon atoms or an alkyl group containing 1 to 3 carbon atoms, which comprises reacting a compound of general formula VII

wherein R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub>, R<sub>5</sub>, A and B are as defined in claim 1, with phosgene, an oxalyl halide, an alkyl or alkanoyl halide containing 1 to 3 carbon atoms each in the alkyl part or with 20 hydrogen cyanide and a hydrogen halide in the presence of a Lewis acid.

32. A process as claimed in claim 31, wherein the reaction is carried out in a solvent at temperatures between 0 and 120°C.

33. A process as claimed in claim 31 or claim 32, wherein the Lewis acid is aluminium chloride.

25 34. A process for the preparation of compounds of general formula I wherein W represents 25 a carboxy group, which comprises reacting a compound of general formula VIII

30 
$$R_3$$
 $R_4$ 
 $R_4$ 
 $R_5$ 
 $R_1$ 
 $R_2$ 
 $R_5$ 
 $R_5$ 
 $R_5$ 
 $R_5$ 
 $R_5$ 
 $R_5$ 

wherein  $R_1$ ,  $R_2$ ,  $R_3$ ,  $R_4$ ,  $R_5$ , A and B are as defined in claim 1 with a hypohalite (optionally formed in the reaction mixture) in the presence of an alkali base.

35. A process as claimed in claim 34 wherein the reaction is carried out in a solvent at
40 temperatures between 0 and 80°C.
36. A process for the preparation of compounds of general formula I, wherein W represents the carboxy group, which comprises oxidizing a compound of general formula IX

45
$$R_{3}$$

$$R_{1}$$

$$R_{1}$$

$$R_{2}$$

$$R_{3}$$

$$R_{1}$$

$$R_{2}$$

$$R_{3}$$

$$R_{3}$$

$$R_{1}$$

$$R_{2}$$

$$R_{3}$$

$$R_{3}$$

$$R_{4}$$

$$R_{5}$$

$$R_{5}$$

$$R_{5}$$

wherein  $R_1$ ,  $R_2$ ,  $R_3$ ,  $R_4$ ,  $R_5$ , A and B are as defined in claim 1 and G represents a group which may be converted into a carboxy group by means of oxidation.

37. A process as claimed in claim 36 wherein the reaction is carried out in a solvent at temperatures between 0 and 100°C.

38. A process for the preparation of compounds of general formula I, wherein  $R_3$  represents a nitro group, which comprises reacting a compound of general formula X

10 wherein R<sub>4</sub>, R<sub>5</sub>, A, B and W are as defined in claim 1. R<sub>3</sub> represents a nitro group and Y 10 represents a nucleophilically exchangeable radical, with an amine of general formula XI

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wherein  $R_1$  and  $R_2$  are as defined in claim 1, and optionally subsequently hydrolyzing. 39. A process as claimed in claim 38, wherein the reaction is carried out in a solvent at 20 temperatures between 20 and 150°C.

40. A process as claimed in claim 38 or claim 39 wherein the reaction is carried out at the boiling temperature of the reaction mixture.

41. A process as claimed in any one of claims 38 to 40 wherein the reaction is carried out 25 in the presence of an excess of the amine of formula XI and/or the N-formyl derivative thereof.

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42. A process as claimed in any one of claims 38 to 41 wherein the reaction is carried out in the presence of an inorganic or tertiary organic base and/or a reaction accelerator and/or in a pressure vessel. 43. A process as claimed in claim 42 wherein the reaction accelerator comprises copper or a

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30 copper salt.

44. A process for the preparation of compounds of general formula I, wherein A represents a group of formula

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wherein R<sub>s</sub> and R<sub>7</sub> are as defined in claim 1, which comprises reducing a compound of general formula XII

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(III)

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55 wherein R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub>, R<sub>5</sub> R<sub>8</sub>, R<sub>7</sub>, B and W are as defined in claim 1, with hydrogen in the presence of a hydrogenation catalyst.

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45. A process as claimed in claim 44 wherein the reaction is carried out in a solvent.

46. A process as claimed in claim 44 or claim 45 wherein the reaction is carried out at a hydrogen pressure of 1 to 5 bar.

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47. A process as claimed in any of claims 44 to 46 wherein the reaction is carried out at temperatures between 0 and 100°C.

48. A process for the preparation of compounds of general formula I, [wherein R₄ represents a hydrogen atom and A represents a methylene or ethylene group (optionally substituted by an alkyl group containing 1 to 5 carbon atoms), a methylene or ethylene group substituted by two 65 alkyl groups containing 1 to 3 carbon atoms each, a methylene group (substituted by a

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cycloalkyl group containing 3 to 7 carbon atoms, an alkoxyalkyl, carboxyl, alkoxycarbonyl, aryl or aralkyl group, whereby each of the above mentioned alkyl parts may contain from 1 to 3 carbon atoms), or a cycloalkylidene group containing 4 to 7 carbon atoms] which comprises, reacting a compound of general formula XIII

5 5 (XIII) 10 10

wherein  $R_1$ ,  $R_2$  and  $R_3$  are as defined in claim 1, and A' represents a methylene or ethylene 15 group (optionally substituted by an alkyl group containing 1 to 5 carbon atoms), a methylene or ethylene group substituted by two alkyl groups containing 1 to 3 carbon atoms each, a methylene group (substituted by a cycloalkyl group containing 3 to 7 carbon atoms, or an alkoxyalkyl, carboxyl, alkoxycarbonyl, aryl or aralkyl group, whereby each of the aforementioned alkyl parts may contain from 1 to 3 carbon atoms), or a cycloalkylidene group containing 4 to 7 20 carbon atoms, with a compound of general formula XIV,

(VIV) 25 25

wherein  $R_{\text{B}}$ , B and W are as defined in claim 1, in the presence of a strong acid.

49. A process as claimed in claim 48, wherein the strong acid is sulfuric acid. 50. A process as claimed in claim 48 or claim 49, wherein the reaction is carried out in a

solvent at temperatures between 20 and 150°C.

51. A process as claimed in any one of claims 14 to 50 wherein a compound of formula I wherein W represents a carboxy group, initially obtained, is converted by means of esterification or amidation into an ester of amide derivative thereof.

52. A process as claimed in any one of claims 14 to 50 wherein a compound of formula I, 35 wherein R<sub>3</sub> and/or W represent nitro groups, initially obtained, is reduced to a compound of formula I wherein R<sub>3</sub> and/or W represent amino groups.

53. A process as claimed in any one of claims 14 to 50 wherein a compound of formula I initially obtained, wherein R<sub>3</sub> and/or W represent an amino group, is converted via a diazonium 40 salt into a compound of formula I wherein R<sub>3</sub> represents a hydrogen or a halogen atom, a 40 hydroxy, alkoxy, mercapto, alkylmercapto, chlorosulfonyl or cyano group and/or W represents a hydrogen or a halogen atom or a cyano group.

54. A process as claimed in claim 53 wherein, a compound of formula A wherein R<sub>3</sub> represents a hydroxy group thereby obtained is alkylated to yield a compound of formula I 45

45 wherein R<sub>3</sub> represents an alkoxy group. 55. A process as claimed in claim 53 wherein a compound of formula I wherein R<sub>3</sub> represents a chlorosulfonyl group thereby obtained is converted by means of ammonia to a compound of formula I wherein R<sub>3</sub> represent an aminosulfonyl group.

56. A process as claimed in any one of claims 14 to 50 wherein a compound of formula I 50 initially obtained wherein R<sub>3</sub> represents an amino group is acylated to yield a compound of 50 formula I wherein R<sub>3</sub> represents an alkanoylamino, aroylamino, alkoxycarbonylamino or alkylsul-

fonylamino group. 57. A process as claimed in any one of claims 14 to 50 wherein a compound of formula I initially obtained wherein R<sub>3</sub> represents an amino group, group is converted by alkylation to a 55 compound of formula I wherein R<sub>3</sub> represents an alkyl- or dialkylamino group.

58. A process as claimed in any one of claims 14 to 50 wherein a compound of formula I initially obtained wherein R<sub>3</sub> represents a chlorine or a bromine atom is converted by dehalogentation to a compound of formula I wherein R<sub>3</sub> represents a hydrogen atom.

59. A process as claimed in any one of claims 14 to 50 wherein a compound of formula I 60 initially obtained wherein R<sub>3</sub> represents a nitrile group is converted by hydrolysis or alcoholysis 60 to a compound of formula I wherein R<sub>3</sub> represents an aminocarbonyl, carboxycarbonyl or alkoxycarbonyl group.

60. A process as claimed in any one of claims 14 to 50 wherein a compound of formula I initially obtained wherein R<sub>3</sub> represents a carboxycarbonyl or alkoxycarbonyl group and/or W 65 represents a carboxy or esterified carboxy group, is reduced to a compound of formula I wherein 65

5	R <sub>3</sub> and/or W represents a formyl or hydroxymethyl group. 61. A process as claimed in any one of claims 14 to 50 wherein a compound of formula I initially obtained, wherein W represents an alkoxycarbonyl group (wherein the alkoxy group may contain from 2 to 6 carbon atoms) substituted in any but the α-position by a hydroxy group, is acylated to a compound of formula I wherein W represents an acyloxy group. 62. A process as claimed in any one of claims 14 to 50 wherein a compound of formula I initially obtained, wherein W represents a hydroxymethyl group, is halogenated and then reacted with a malonic acid diester to form a compound of formula I wherein W represents an ethyl group substituted by two alkoxycarbonyl groups.	5
10	63. A process as claimed in any one of claims 14 to 50 wherein a compound of formula I initially obtained, wherein W represents a formyl group, is converted by means of condensation and optional subsequent hydrolysis and/or decarboxylation to a compound of formula I wherein W represents a vinyl group substituted by a hydroxycarbonyl or alkoxycarbonyl group.	10
15	64. A process as claimed in any one of claims 14 to 50 wherein a compound of formula I initially obtained, wherein W represents an ethyl group substituted by two alkoxycarbonyl groups, is converted by hydrolysis and decarboxylation to a compound of formula I wherein W represents an ethyl group substituted by one carboxy group.  65. A process as claimed in any one of claims 14 to 50 wherein a compound of formula I	15
20	initially obtained, wherein W represents a carboxy group, is converted via a sulfonic acid hydrazide and subsequent disproportionation into a compound of formula I wherein W represents a formyl group.	20
25	66. A process as claimed in any one of claims 14 to 50 wherein a compound of formula I initially obtained, wherein R <sub>1</sub> and R <sub>2</sub> together with the nitrogen atom to which they are attached represent an aza-1,4-dioxa-spiro-alkyl group containing 6 to 8 carbon atoms, is hydrolysed to a compound of formula I wherein R <sub>1</sub> and R <sub>2</sub> together with the nitrogen atom to which they are attached represent an unbranched alkyleneimino group containing 4 to 6 carbon atoms wherein a methylene group is replaced by a carbonyl group.	25
30	67. A process as claimed in any one of claims 14 to 50 wherein a compound of formula I initially obtained, wherein R <sub>1</sub> and R <sub>2</sub> together with the nitrogen atom to which they are attached represent an unbranched alkyleneimino group containing 4 to 6 carbon atoms wherein a methylene group is replaced by a carbonyl group, is reduced to a corresponding hydroxyalkyleneimino compound of formula I.	30
35	68. A process as claimed in any one of claims 14 to 50 wherein a compound of formula I initially obtained, wherein W represents an aminocarbonyl group, is dehydrated to a compound of formula I wherein W represents a cyano group.  69. A process as claimed in any one of claims 14 to 68 wherein a compound of formula I initially obtained is subsequently converted into a salt thereof with an organic or inorganic acid or base, or a salt of a compound of formula I initially obtained is subsequently converted into a	35
40	compound of formula !.  70. A process as claimed in any one of claims 14 to 69 for the preparation of compounds as claimed in claim 11.	40
45	71. A process for the preparation of compounds as claimed in claim 1 substantially as herein described in any one of the Examples.  72. A process for the preparation of compounds as claimed in claim 11 substantially as herein described in any one of Examples 1, 8, 24, 29–31, 35, 36, 38, 40 or 48.  73. Compounds as claimed in claim 1 when prepared by a process as claimed in any one of claims 14 to 72.	45
50	<ul> <li>74. Compounds as claimed in claim 11 when prepared by a process as claimed in claim 70 or claim 72.</li> <li>75. Pharmaceutical compositions comprising as active ingredient at least one compound of formula I as defined in claim 1 or a physiologically compatible salt thereof, in association with</li> </ul>	50
55	one or more pharmaceutical carriers or excipients.  76. Compositions as claimed in claim 75 in a form suitable for oral or parenteral administration.  77. Compositions as claimed in claim 75 or claim 76 in the form of tablets, coated tablets,	55
99	capsules, powders or suspensions.  78. Compositions as claimed in any one of claims 75 to 77 in the form of dosage units.  79. Compositions as claimed in claim 78 wherein each dosage unit contains from 1 to 50	ວວ
60	mg of active ingredient. 81. Compositions as claimed in any one of claims 75 to 80 wherein the compound of	60
65	formula I is as defined in claim 11.  82. Pharmaceutical compositions as claimed in claim 75 substantially as herein described.  83. Pharmaceutical compositions substantially as herein described in any one of Examples A	65

34A 63

to D.

84. Compounds of general formula I as claimed in claim 1 and physiologically compatible salts thereof for use in a method of treatment of patients suffering from disorders of intermediary metabolism and/or blood sugar disorders.

85. A method of treating patients suffering from, or susceptible to disorders of intermediary metabolism and/or blood sugar disorders which comprises administering to the said patient an effective amount of a compound of formula I as defined in claim 1 or a physiologically compatible salt thereof.

86. Each and every novel method, process, compound or composition herein disclosed.

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